

Polydentate P, N-based ligands for palladium-catalyzed cross-coupling reactions

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ABSTRACT

In recent years, Pd-catalyzed cross-coupling reactions have been used widely as expeditious reactions in various fields of chemistry. Traditionally, these cross-coupling reactions have been carried out by using different phosphine ligands. Though the phosphine ligands have been extensively used, they suffer from many limitations like poor air, moisture, and thermal stability. Hence, in recent years phosphine-free ligands such as N-heterocyclic carbenes and amines have attracted countless attention in the field of catalysis. Unfortunately, the ligands having only N as a donor atom can activate aryl iodides and bromides, while aryl chlorides are less reactive. Hence, the P–N ligands containing functional moiety, with donor N or P atom have been prepared and used for different organic transformations. An even more effective mixed donor P, N-ligands have many advantages in asymmetric catalysis because of the distinctly different characteristics like a 'soft' P-ligand as a π -acceptor and a 'hard' N-ligand as a σ -donor. This compressive review highlights the results of the highly active ligands and complexes containing N in combination with P as a donor atom.

1. Introduction

Palladium-catalyzed C–C and C–X coupling reactions [1] have tremendous applications in the field of natural products, pharmaceuticals, agrochemicals, and functional materials [2]. A large number of homogeneous and heterogeneous Pd catalysts have been applied in these coupling reactions [3]. Traditionally, these reactions have been performed using different phosphine ligands. Though the phosphine ligands are extensively used, they suffer from many limitations like poor air, moisture, and thermal stability. Hence, in recent years phosphine-free ligands such as N-heterocyclic carbenes [4] and amines [5] have attracted countless attention in the field of catalysis. Similar to Pd complexes containing N with the C [6], and O [7] as donor atoms, it is possible to develop tunable P–N ligands such as PHOX, Pincers, Atropos, Schiff bases, and Aminophosphines for various cross-coupling reactions.

2. Ligands containing N and P atoms (P, N ligands)

Phosphines (PR₃) are more commonly known as a spectator rather than actor ligands, as electronic and steric properties can be altered systematically and predictably by varying the R groups [8]. Unlike NR₃,

they are also π -acids to some extent, which depends on the nature of the R groups. As the R group becomes more electronegative, the σ^* orbital of the P–R bond becomes more stable and the P contribution to the σ^* orbital increases, so the size of the σ^* lobe that points toward the metal increases [9]. The electron-richness and sterically bulky skeletons can facilitate oxidative addition and reductive elimination steps [10]. However, mono phosphine ligands have restricted coordinative flexibility for stabilizing the unstable low-coordinate and low-valent Pd complex. The highly active complexes are converted into inactive Pd black at high temperatures or under prolonged heating, limiting their TON. The use of mixed P–N-type ligands can avoid these limitations. Heterobidentate P and N ligands are interesting ligands as “soft” (P atom), and “hard” (N atom) characters enable fine-tuning in the stabilization of various intermediates in catalytic cycles. Due to the electron disparity between the two P and N distinct coordination sites, and P, N-ligands containing sp² hybridized N donors like pyridine, imine, and oxazoline offer unique properties [11]. A large number of ligands containing functional moiety, with donor N or P heteroatoms (P–P and N–N ligands) have been prepared and used for different organic transformations [12]. Because of the distinctly different characteristics of P and N atoms, more effective P–N ligands have many advantages in

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- 1 •The P-N based ligands can be easily separated from the reaction mixture *via* phase separation under acidic conditions.
- 2 •The chelate effect of P-N ligand confers stability to the catalyst precursor in the absence of substrates.
- 3 •The P-N ligands play a dual role, the π -acceptor character of P stabilize a metal center in a low oxidation state and the σ -donor ability of N makes the metal more susceptible to oxidative addition.
- 4 • In this hard-soft combination, the hard end weakly coordinates to soft metal centers and easily dissociate in solution to afford a vacant site.
- 5 •This combination also helps to stabilize the intermediate oxidation states during the catalytic cycle.
- 6 •In addition, it is easy to independent structural alternation for each of the P and N donor sites that makes it convenient to build a set of ligands for special applications in catalysis.
- 7 •In addition, these ligands can employ a high degree of regiocontrol by 'trans' effect.

Fig. 1. Advantages of the P-N ligands in transition metal catalysis.

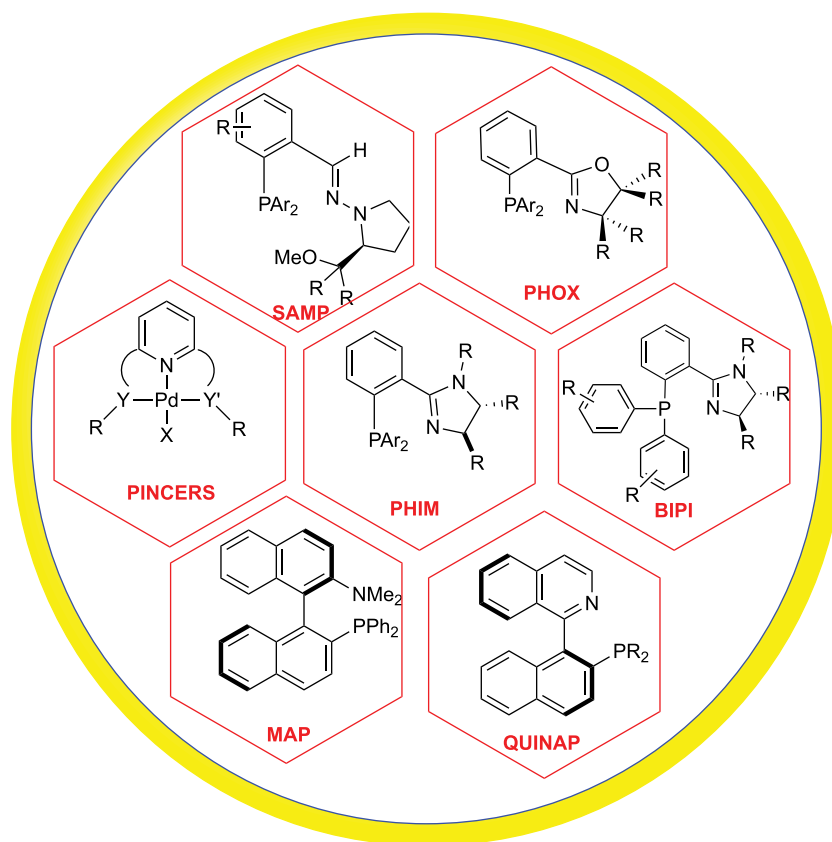


Fig. 2. Privileged P-N ligands used in Pd-catalyzed coupling reactions.

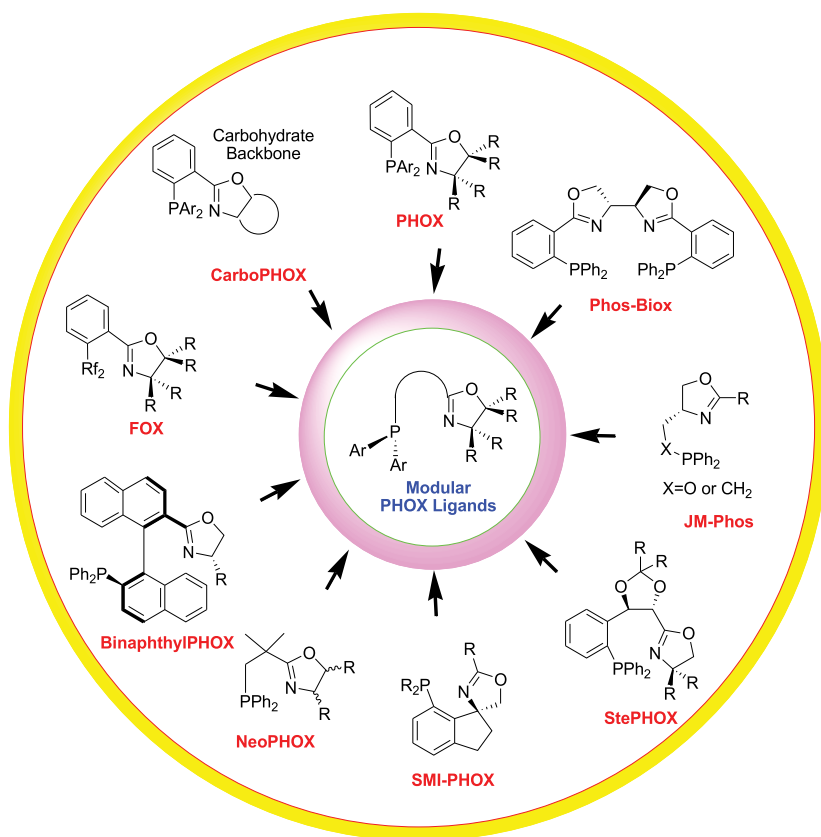


Fig. 3. The scaffolds of PHOX ligands used in Pd-catalyzed coupling reactions.

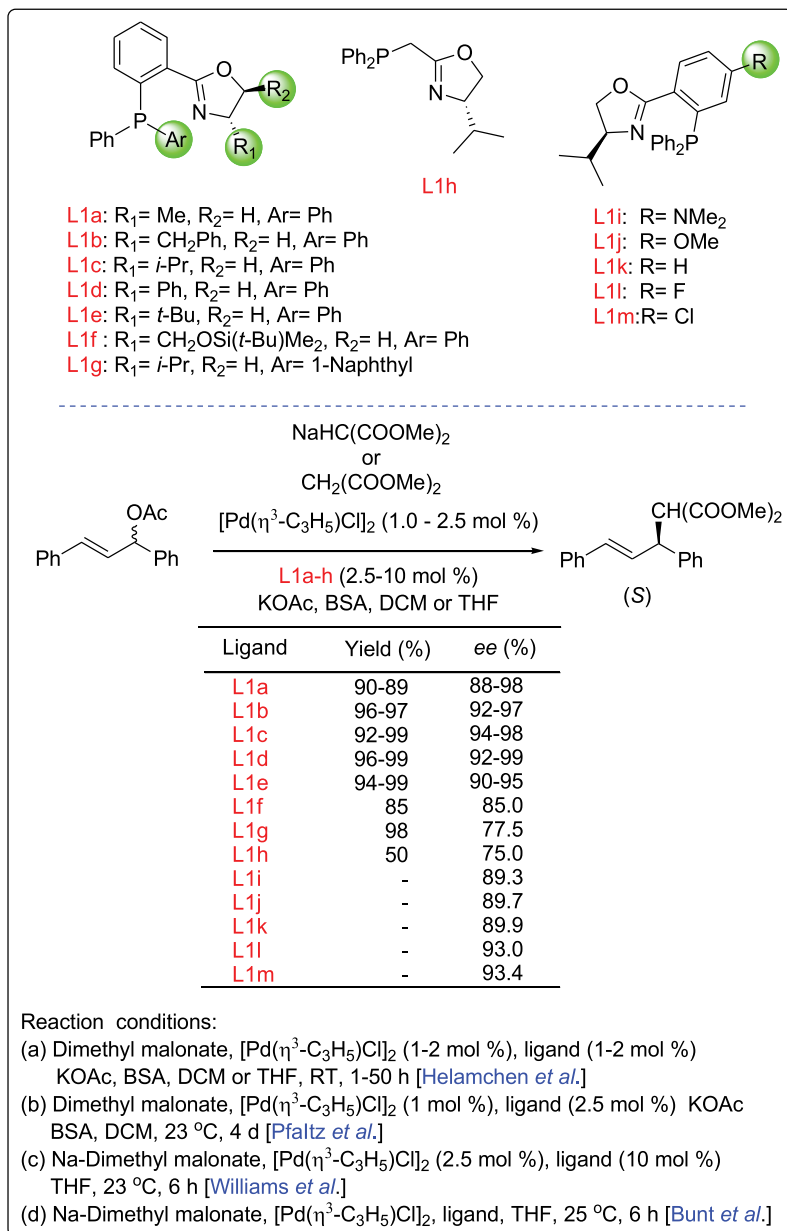
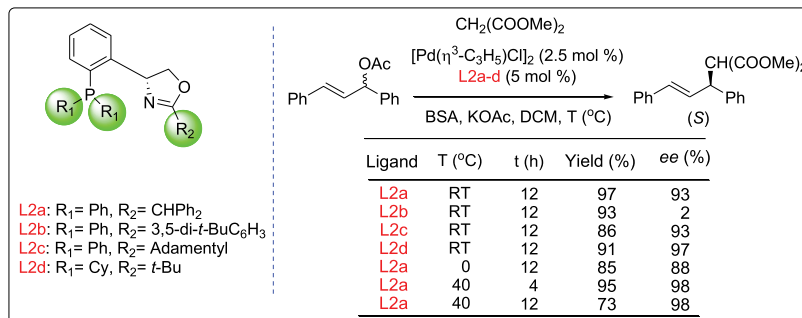
asymmetric catalysis. These mixed donor ligands are a class of ‘hemilabile ligands’ that have many applications in asymmetric catalysis [13]. The P–N ligand offers the following advantages (Fig. 1).

In summary, the good π -acceptor character of P and the σ -donor ability of the N atom of P–N ligands created a great interest in the field of asymmetric catalysis [14a]. It was reported that selectivity in asymmetric catalysis depends on the relative energy of the two diastereomeric transition states leading to ‘R’ and ‘S’ products concerning a common resting state. Many P–N-based ligands provide a unique chiral environment to different intermediates involved in catalytic cycles. Recently, P–N-based nonsymmetrical modular ligands have been applied successfully in various metal-catalyzed reactions [14b,c]. In the last decades, mechanistic studies of these reactions were supported by computational studies [15]. In this connection, the present review article covered the reports on different Pd-catalyzed cross-coupling reactions using several P–N ligands such as PHOX, QUNAP, SAMP, MAP, PHIM, BIPI, Aminophosphines, Pincers, and Schiff-bases (Fig. 2).

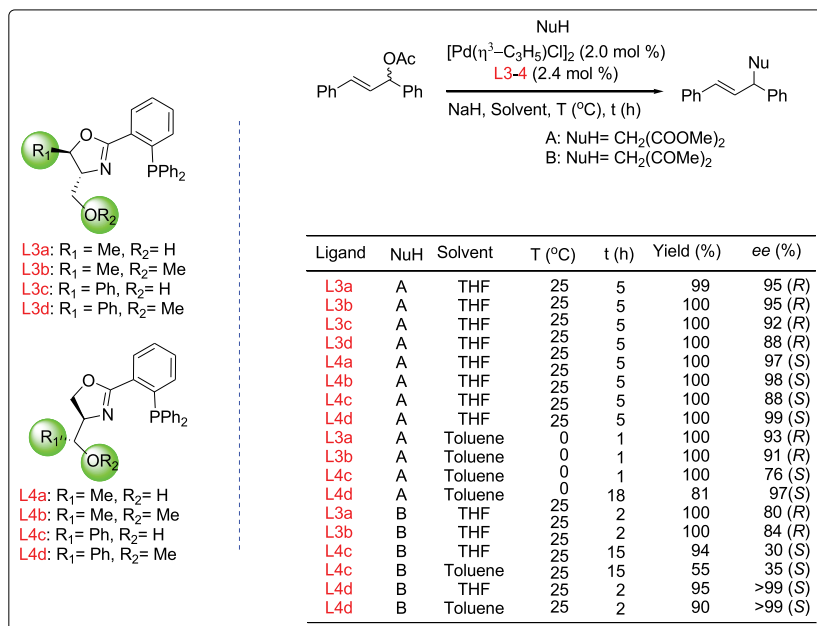
Below are the peculiar distinctive properties of the ligands underlined.

- Due to their ready accessibility and modular construction, PHOX ligands are the most commonly investigated ligands.
- HetPHOX has emerged as a new ligand for asymmetric catalysis, as it easily introduces the required diphenylphosphino groups on the ligand backbone.

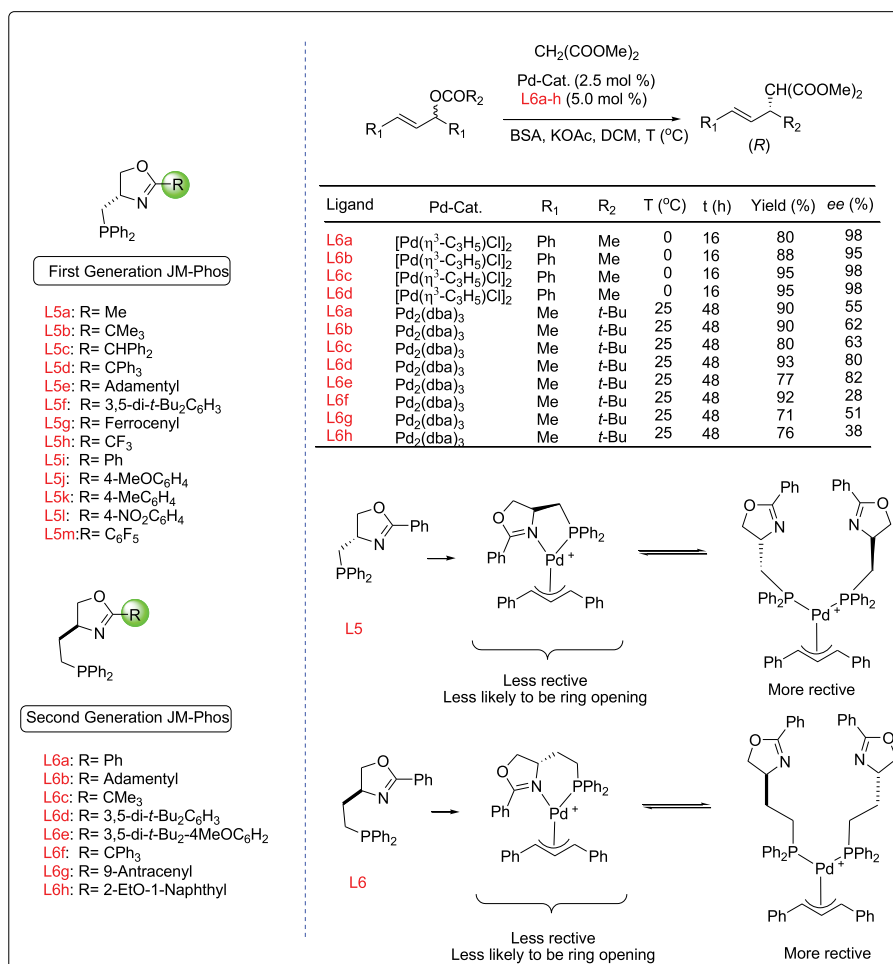
- Similar to PHOX ligands, more flexible phosphinoimidazoline-based ‘BIPI’ ligands containing additional nitrogen were used for the creation of a chiral quaternary center. It allows fine-tuning of the gross electronic properties by varying the substituents. In addition, BIPI ligands could be constructed from chiral diamine in a modular fashion.
- As analogs to that of BIPI-ligands. PHIM ligands have been acting as interesting ligands in asymmetric catalysis, as it is readily available from C_2 -symmetric diamine fragments. In addition, these ligands contain a second nitrogen atom that can work as an additional source of molecular diversity, as it allows to adjustment of the electronic properties of the coordinating nitrogen atom. In addition, it is possible to immobilize the ligand on a suitable solid support.
- SAMP and their analogs derived from chiral phosphine hydrazones have also been developed as efficient ligands for many Pd-catalyzed reactions.
- The transition metal ECE and ECE’ type pincer complexes also attracted great interest as a new class of metal complexes for the range of catalytic applications.
- The ‘Atropos’ ligands like ‘QUINAP, PINAP, and QUINAZOLINAP’ are a specific type of biaryls, with one component carrying a pendant diaryl substituted phosphine, and the other bears an sp^2 N-atom adjacent to the biaryl link acts as privileged chiral ligands for many asymmetric transformations.

Scheme 1. C₂ symmetric PHOX ligands used in asymmetric allylic alkylations.

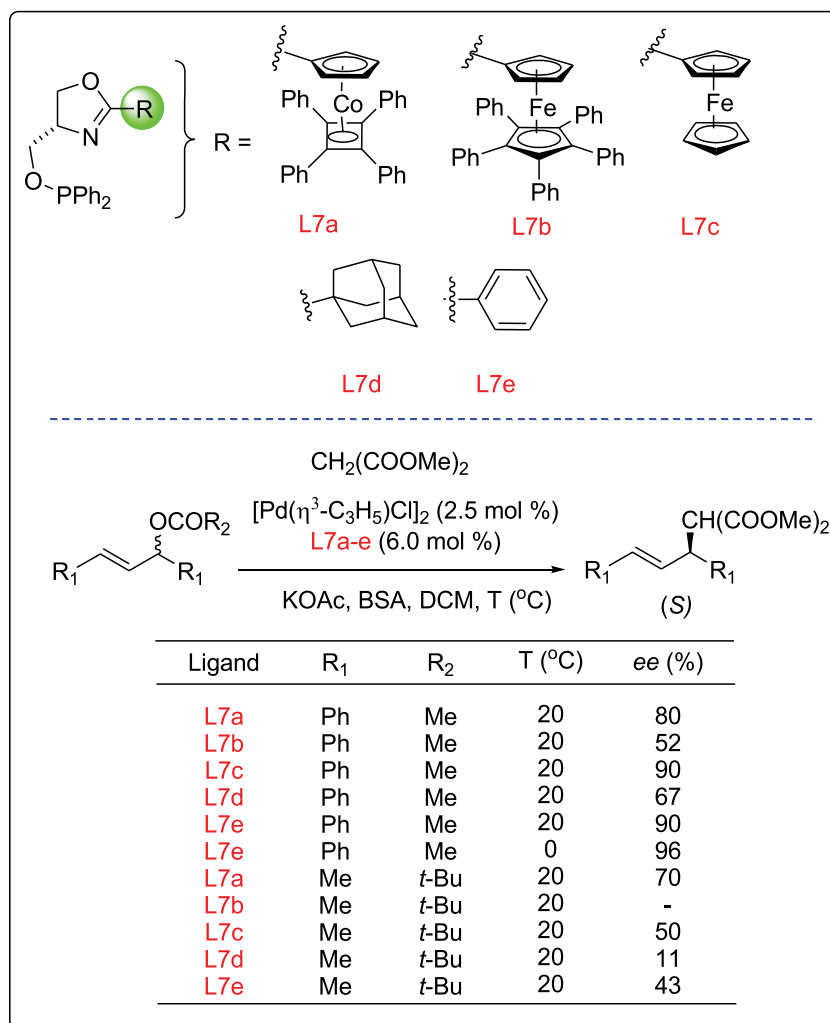
Scheme 2. Conformationally rigid PHOX ligands for asymmetric allylic alkylations.



Scheme 3. Effect of substituents on the 4-positions of the oxazoline ring on asymmetric allylic alkylations.



Scheme 4. Applications of JM-Phos in asymmetric allylic alkylations.



Scheme 5. Phosphinite-oxazoline P-N ligands used in asymmetric allylic alkylations.

3. Applications of P-N ligands in Pd-catalyzed reactions

All these modular ligands have been applied successfully in various Pd-catalyzed reactions such as asymmetric Tsuji, Heck, Suzuki, Sonogashira, Buchwald, Stille, and Hiyama coupling reactions under different reaction conditions.

3.1. Tsuji-Trost asymmetric allylic alkylation reactions

The 'Tsuji-Trost reaction' comprehended as 'Trost allylic alkylation' [16] was first introduced by Jirō Tsuji [17] in 1965 and, later in 1973, adapted by Barry Trost [18]. This reaction is expanded now to different nucleophiles, like active methylenes, enolates, amines, and phenols. The chiral phosphine ligands not only lead to improved reactivity but also provided high enantioselectivity and diastereoselectivity under milder reaction conditions. The mild reaction conditions with high selectivity greatly expand the effectiveness of this reaction in both pharmaceuticals as well as natural product synthesis [19]. Today various P-N-based ligands were applied for this reaction.

The chiral oxazolines called 'Phosphinooxazoline (PHOX)' ligands have an outstanding ability to differentiate the two allylic termini in Pd-catalyzed allylic alkylation reactions hence these ligands are widely used as 'spectator' ligands [20]. In addition, these tailor-made ligands can be readily accessible from various naturally occurring chiral amino alcohols and achieve a wide range of regio and diastereoselectivities

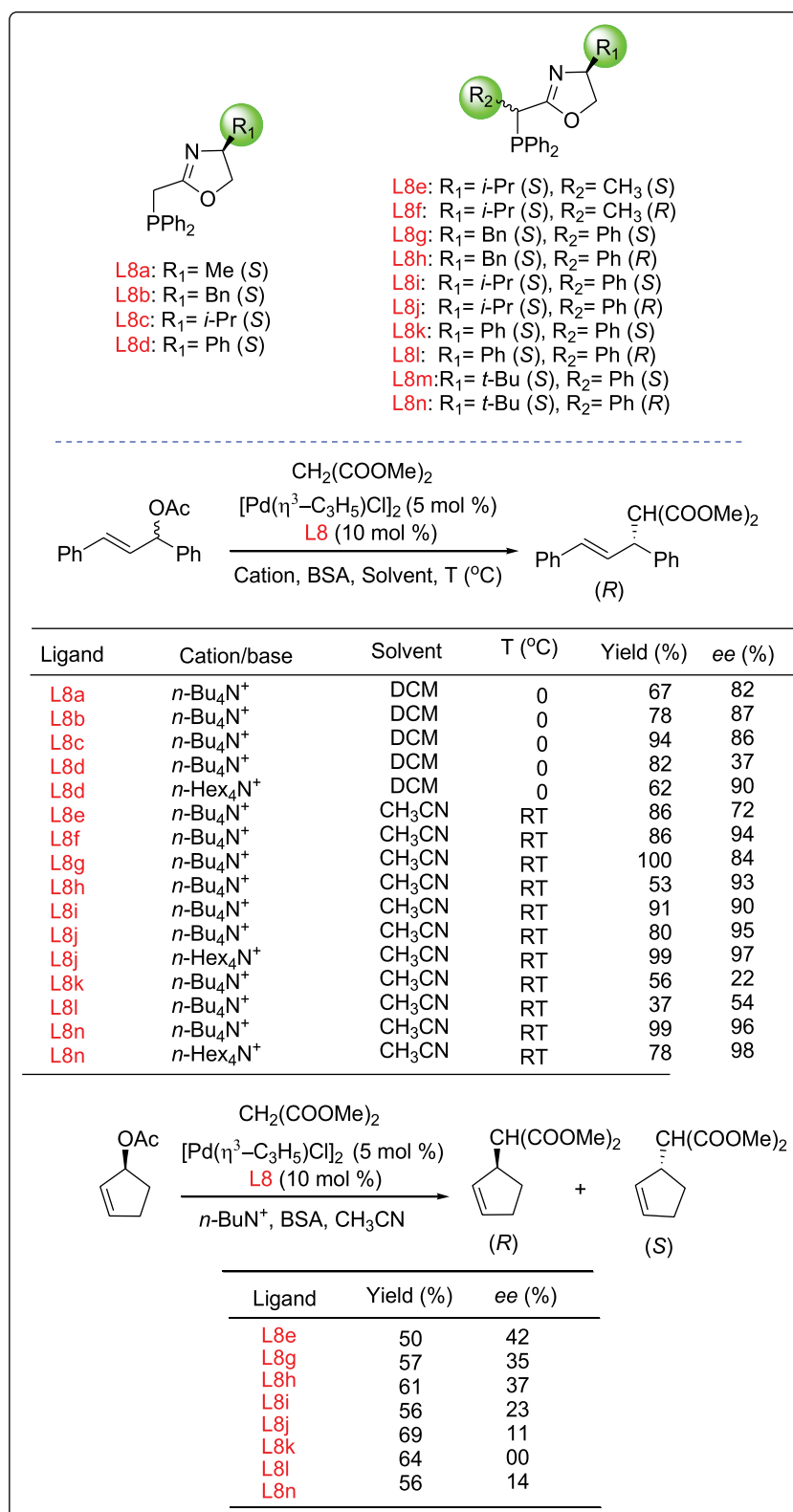
[21].

The following different factors determine regioselectivity and diastereoselectivity in allylic alkylation reactions.

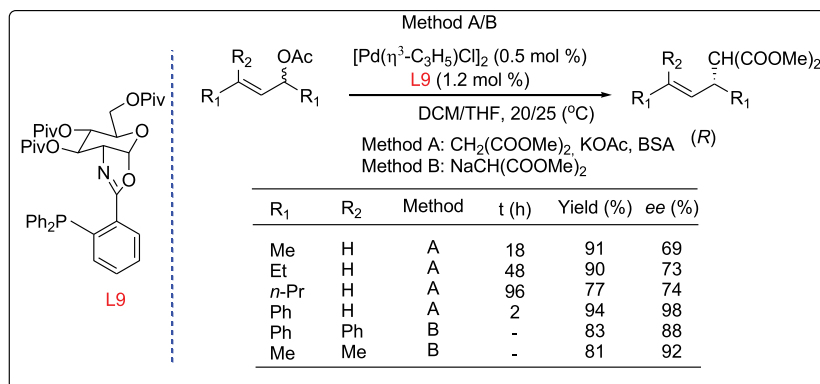
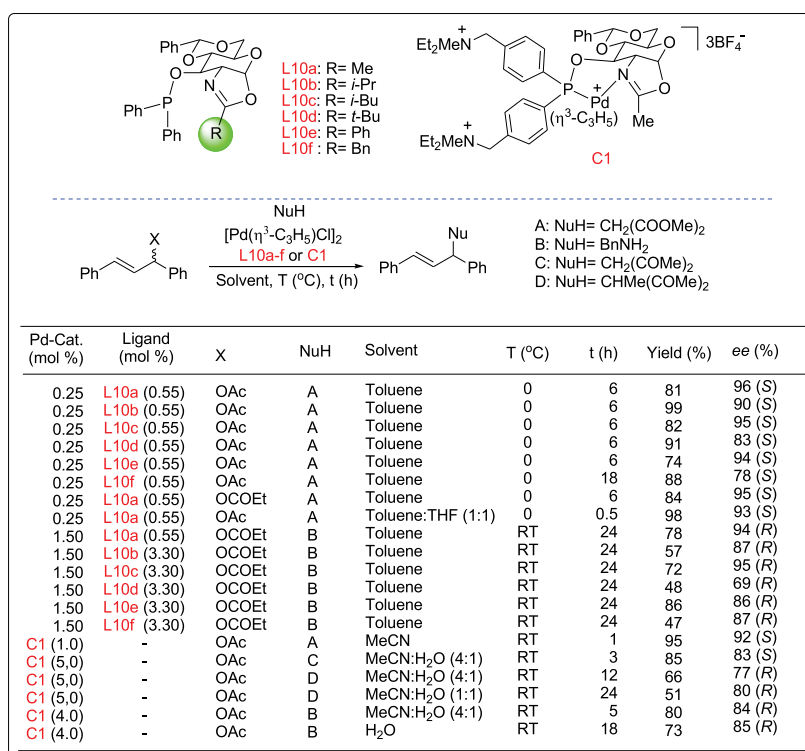
- Initial conformation of the allylic component.
- The nature of its leaving group.
- Conformations of the intermediate allylic Pd complexes.
- The hardness of the nucleophile.
- Electronic and steric properties of the ligands bound to the Pd center.

The traditional C_2 symmetric chelate diphosphines such as CHIR-APHOS and BINAPs were not well suited for allylic alkylation reactions. Hence the C_2 symmetric PHOX ligands having a wide bite angle showed good to excellent enantioselectivity. The PHOX ligands with P and N donor atoms have distinctly different π -acceptor strengths. Due to the contrasting 'trans' effects of P and N atoms, a nucleophilic attack on the allyl palladium complexes of PHOX ligands was well controlled. Various derivatives of PHOX can be easily obtained due to the modular nature of the PHOX ligands (Fig. 3). Hence, it is possible to fine-tune to optimize the ligand structures and substrates to give satisfactory results.

In this regard, the PHOX ligands were first introduced independently by Pfaltz (L1a-f), Helmchen (L1c, L1g, and L1h), and Williams (L1a-e) as highly effective C_2 symmetric ligands for asymmetric allylic alkylation reactions [22]. All the ligands showed very high enantioselectivity (up to 99%) for alkylation of racemic 1,3-diphenyl-2-propenyl acetate



Scheme 6. PHOX ligands containing one and two chiral centers used in asymmetric allylic alkylations.

Scheme 7. Phosphino- α -D-glucopyrano-oxazoline ligand for asymmetric allylic alkylations.

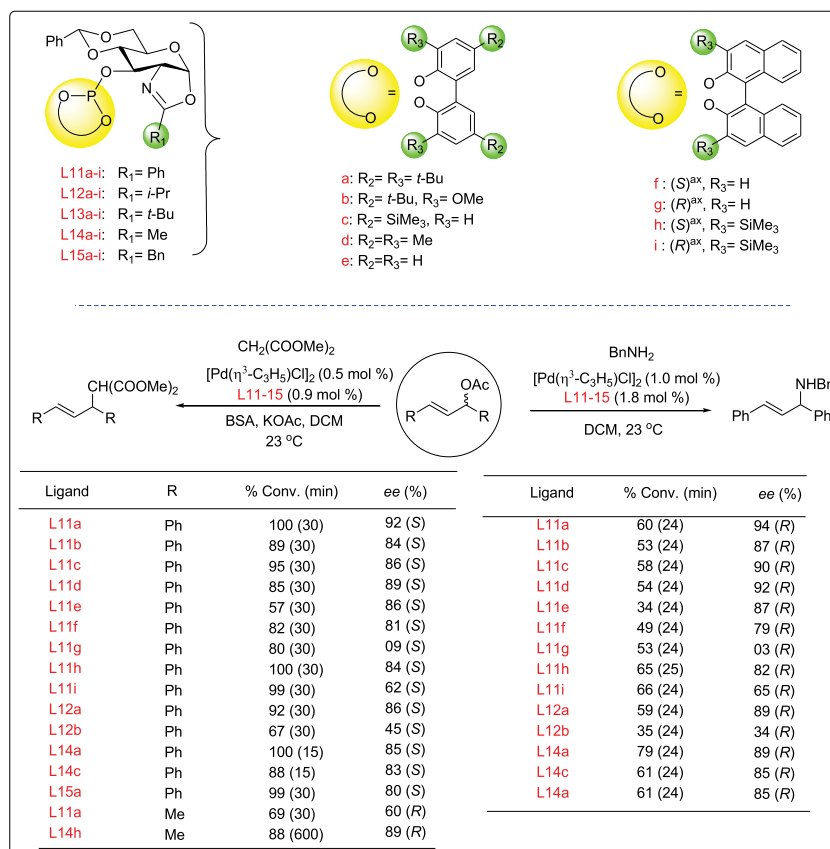
Scheme 8. Chiral PHOX ligands from D-glucosamine for asymmetric allylic alkylations.

with dimethyl malonate or its Na salt in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5\text{Cl})_2]$ (1.0-2.5 mol %) and KOAc-BSA in DCM or THF at room temperature (Scheme 1). The ligands **L1c** and **L1e** were also used for the Pd-catalyzed reaction of allyl acetates and Schiff base derivatives of α -amino phosphonates with good diastereoselectivities (up to 87:13) and enantioselectivities (> 96% *ee*) [23a]. Bunt et al. [23b] carried out the Hammett studies of PHOX ligands **L1i-m** in asymmetric allylic alkylation and amination reactions. It was observed that alkylation with Na salt of dimethylmalonate resulted in only a small difference in the *ee*, nevertheless, amination with benzylamine gave a much wider variation in the *ee*.

Zang et al. [24] synthesized a new class of conformationally rigid PHOX ligands (**L2a-d**) via an efficient *ortho*-substitution of phenyl glycinol. All the ligands (5 mol %) except **L2b** showed 86–97% yields with high enantioselectivity (93–97% *ee*) in the asymmetric allylic alkylation

reactions of 1,3-diphenyl-2-propenyl acetate with Na salt of dimethyl malonate in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5\text{Cl})_2]$ (2.5 mol %), BSA, and KOAc in DCM at room temperature (Scheme 2). It was observed that for ligand **L2a**, lowering the reaction temperature from RT to 0 °C, moderates the product yield from 97% to 85% as well as enantioselectivity from 93% to 88%. On the other hand, an increased temperature to 40 °C increased the yield to 97%, increasing enantioselectivity to 98%. Most interestingly, lowering the Pd loading to 0.1 mol % and 0.2 mol % decreased the yield to 73%, nevertheless, the enantioselectivity remains the same (98% *ee*).

In asymmetric metal catalysis, efficient transfer of chirality and control of the absolute configuration of new stereocenters depends on the ligands' properties. The bisoxazoline ligands containing 4-hydroxybenzyl and 4-methoxybenzyl substituents with the same stereochemistry provided the opposite absolute configuration of the alkylation

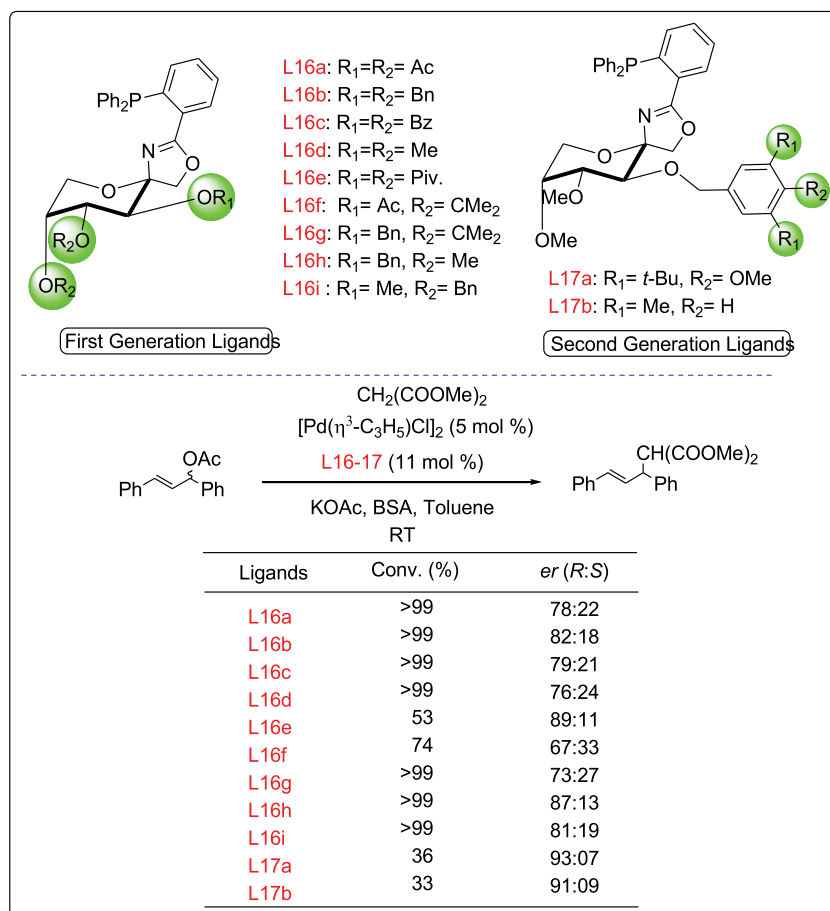


Scheme 9. A family of carbohydrate-based phosphite-oxazoline ligands for asymmetric allylic alkylations.

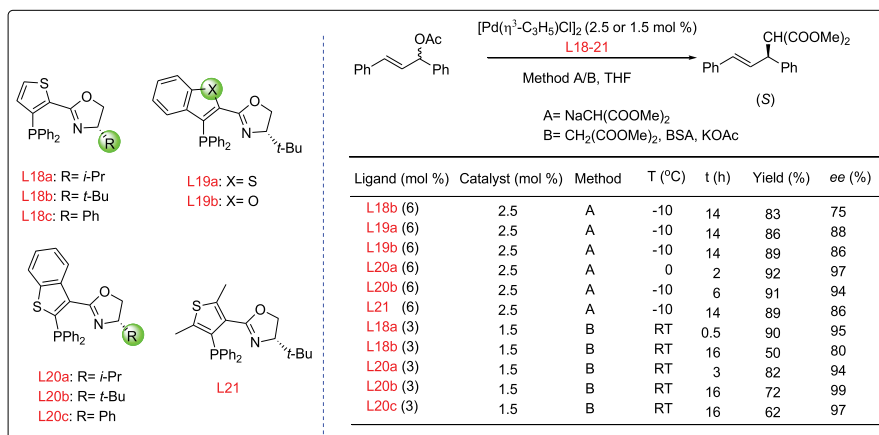
products [25]. It is due to the interaction of a hydroxy group of the ligand with the nucleophile *via* hydrogen bonding [26]. Similarly, due to the hydrogen bonding, different enantioselectivities were shown by 2-(1-hydroxyalkyl) and 2-(1-alkoxyalkyl)pyridinooxazolines in allylic alkylation reactions [27]. The effect of hydrogen bonding on selectivity was studied by applying several PHOX ligands containing (1-hydroxy-1-phenyl)methyl (**L3a-d**) and (1-methoxy-1-phenyl)methyl (**L4a-d**) substituents at the 4-positions of the oxazoline ring in allylic alkylations [28]. The conformational preferences and enantiodiscrimination of these ligands were also studied for the reactions of 1,3-diphenyl-2-propenyl acetate and 3-cyclohexenyl with dimethyl malonate and acetylacetone using $[(\eta^3\text{-C}_3\text{H}_5)\text{PdCl}]_2$ (2 mol %) in THF or Toluene at 0 °C (Scheme 3). Both types of ligands showed different enantioselectivities for alkylations of 1,3-diphenyl-2-propenyl acetate with malonate and acetylacetone. This enantiodiscrimination was arising due to the different conformations of the methoxy-containing ligands. The exo-allyl complex containing the methoxy-group resulted in the most stable conformation, leading to products with an *S* configuration, while, the two Pd(0)-olefin complexes originating from exo-allyl and endo-allyl complexes containing a hydroxy group of similar energy, lead to the enantioselectivity. Consequently, the catalysts with the methoxy-containing ligand provided products with high enantioselectivity compared to a hydroxyl-containing ligand. The presence of a hydrogen bond with Pd(0) as the proton acceptor in the hydroxy-containing olefin-Pd(0) complexes induces a conformational change in the ligand leading to different stereoselectivity as analyzed by DFT calculations.

Burgess and coworkers developed a series of PHOX ligands (JM-Phos) (**L5a-m** and **L6a-h**) and successfully used them in Pd-catalyzed allylic alkylation reactions. First-generation ligands **L5a-m** were prepared *via* a divergent synthesis involving the pivotal intermediate and phosphine-substituted amino alcohols (Scheme 4) [29]. The correlation between the nature of substituent (R) of ligand and asymmetric induction in allylation was achieved by using high-throughput screening. Ligands **L5a-m** were screened in the parallel reactor by reacting 1, 3-diphenyl-2-propenyl acetate and dimethyl malonate using $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$, BSA, and KOAc under different reaction conditions. It showed that DCM was the best solvent as it gave the highest selectivities. However, toluene and THF were better solvents for the pentafluorophenyl ligand (**L5m**). The aryl rings with electron-donating *para* substituents (4-Me and 4-OMe) showed higher enantioselectivities (up to 82% *ee*) than electron-withdrawing substituents (4-NO₂ and C₆F₅). Thus, the high enantiodiscriminations seem to be facilitated by electron-rich oxazoline substituents. It was also observed that the size and shape of the substituent has more significant effects than the electronic features as the pseudo-spherical adamantyl ligand **L5e** was superior (94% *ee*), to the smaller methyl-substituted ligand **L5a** (53% *ee*). The enantiomeric excess was reversed to -6% for the triphenylmethyl-substituted ligand **L5d**. It was revealed that in asymmetric reactions, the substituent topography can be more relevant than size.

They developed the second generation **L6a-d** [30] and **L6e-h** [31] ligands containing fairly conformationally flexible ethylene linkers for Pd-catalyzed allylation reactions using a multiwall reactor. The



Scheme 10. Spiro-fused ligands derived from carbohydrates for asymmetric allylic alkylations.

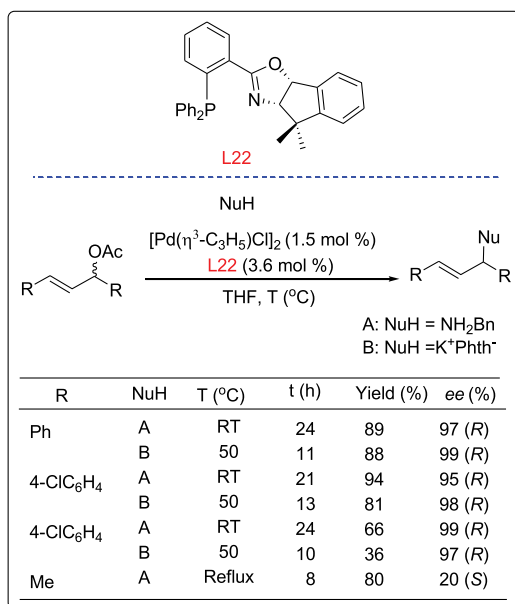


Scheme 11. HetPHOX based on thiophene and benzothiophene for asymmetric allylic alkylations.

second-generation ligands were superior to the first-generation ligands. The less likely ring-opening complexes of ligand L6 hurt the enantioselectivity. The ligands L6a-d (2.5 mol %) showed an excellent *ee* (>95%) for the reaction of 1,3-diphenyl-2-propenyl acetate and dimethyl malonate using [Pd(η³-C₃H₅)Cl]₂, BSA, and KOAc in DCM at 25 °C. While the ligand L6e gave a 77% yield with 82% *ee* for the allylation reaction of 1,3-dimethyl-2-propenyl pivalate and dimethyl

malonate in the presence of Pd₂(dba)₃ under similar reaction conditions. The ligand L6a was excellent in terms of the yield and enantioselectivity for the allylation of 1,3-diphenyl-2-propenyl acetate with 3-((Z)-1-trimethylsilyloxypro-1-enyl)-2-oxazolidone silyl enolate at 0 °C.

Similarly, Jones and Richards [32] utilized new phosphinite-oxazoline ligands (L7a-e) (6 mol %) in the allylic alkylation reaction of racemic propenyl substrates with dimethyl malonate in the presence of



Scheme 12. Indanol-based PHOX ligand for asymmetric allylic alkylations.

$[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (2.5 mol %), BSA, and KOAc in DCM at 20 °C (Scheme 5). The ligands were obtained from (*S*)-serine methyl ester and different acid chlorides (RCOCl). The ligands containing relatively small ferrocenyl (**L7c**) or phenyl (**L7e**) groups showed excellent enantioselectivity. The reaction at 0 °C using ligand **L7e** showed slightly increased selectivity (96%). All these results revealed that shifting alkyl diaryl phosphine ligands (**L6**) to structurally similar diarylphosphinite ligands (**L7**) does not change the selectivity or sense of asymmetric induction. However, the conformationally locked nature of phosphinite does not necessarily require for high selectivity.

Gilbertson and Chan [33] developed a modular route for the synthesis of PHOX ligands containing one (**L8a-d**) and two (**L8e-n**) chiral centers from amino acids and phosphino carboxylate building blocks. All these ligands (10 mol %) were tested in the asymmetric alkylation of 1, 3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (5 mol %), BSA (Scheme 6).

It was observed that the ligands derived from phenylalanine (**L8b**) and valine (**L8c**) gave the 87% and 86% *ee* in the presence of Bu_4N^+ -BSA cation base combination in DCM at 0 °C. While the highest enantioselectivity (90% *ee*) was attained using **L8d** in the presence of *n*-Hex₄N⁺-BSA. Moreover, except ligands **L8k** and **L8l**, all other ligands containing two chiral centers (**L8e-n**) with an *R*, *S* configuration were found to be the most selective ligand in acetonitrile. The 42% *ee* was obtained for the ligands **L8e** in allylation of 2-cyclopentenyl acetate with dimethyl malonate in acetonitrile. A chiral center next to the phosphine has little effect on the selectivity of the reaction.

Recently, carbohydrates have evoked much attention as the chiral backbone in many asymmetric transformations [34]. In the Pd-catalyzed asymmetric allylic substitution reactions, the D-Glucosamine-derived chiral PHOX ligands are the best as it provides a chiral environment for catalysts. [35]. In this regard, Kunz et al. [36] synthesized a new enantiomerically pure phosphine- α -D-glucopyrano-oxazoline ligand (**L9**) (Scheme 7). The ligand synthesis involves the N-acylation of glucosamine with 2-fluoro benzoyl chloride, which on O-pivaloylation gave α -D-glucopyranosyl bromide [37]. These on treatment with Et₄NBr

gave 2-(2-fluorophenyl)-oxazoline, which reacts with potassium diphenylphosphine and delivered the desired ligand **L9**. The $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2/\text{L9}$ (0.5:1.2 mol %) catalytic system gave high yields (up to 94%) and enantioselectivity (up to 98% *ee*) for reactions of substituted allyl acetates with dimethyl malonate in presence of BSA and KOAc in THF at 25 °C. The catalytic system also coupled substituted allyl acetates with Na-malonate in DCM at 20 °C with high yields (83%) and enantioselectivity (up to 92% *ee*).

Next, Uemura and coworkers reported the detailed work on novel chiral phosphinite–nitrogen PHOX ligands (**L10a–f**) derived from commercially available D-glucosamine hydrochloride [38]. When applied to Pd-catalyzed asymmetric allylic alkylation and amination reactions of 1,3-diphenyl 2-propenyl acetate, a high level of enantioselectivity was obtained in the presence of BSA, KOAc in toluene at 0 °C (Scheme 8).

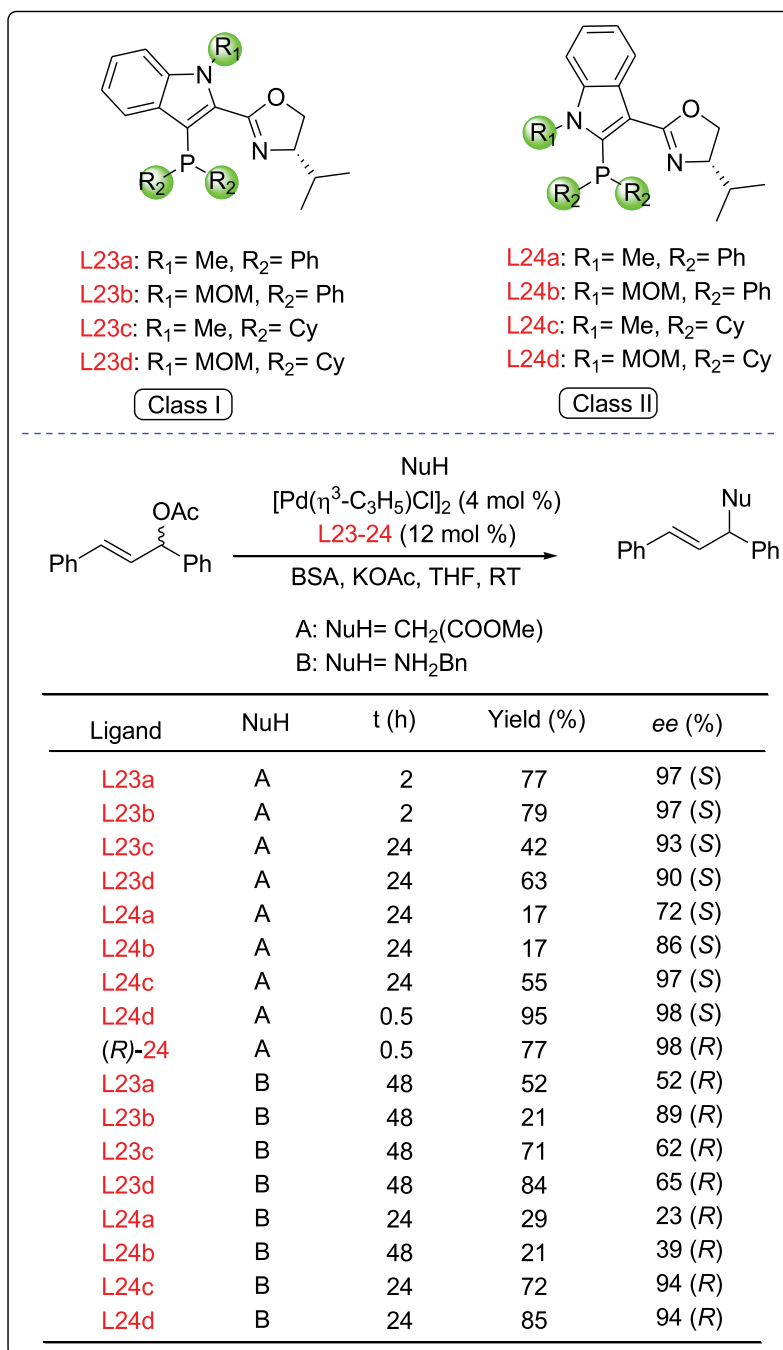
The activity was higher for less sterically demanding substituents present on oxazoline. The activity order was Me > Ph > *i*-Pr > *t*-Bu, and the ligand **L10a** (0.55 mol %) containing the smallest substituent on oxazoline furnished the highest *ee* (96%). Furthermore, the ligand **L10a** was also effective for the amination of ethyl 1,3-diphenylprop-2-enyl carbonate showing 94% *ee*. In the allylic alkylation of 3-acetoxycyclohexene, the $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2/\text{L10a}$ (1.1:0.5 mol %) catalytic system showed the highest enantioselectivity (74% *ee*) with dimethyl malonate in the presence of BSA, KOAc in THF at 0 °C.

The Pd-CarboPHOX (**C1**) containing amphiphilic ligand (**L10a**) having the diethylamino methyl group at the 4-position of the phenyl ring on the phosphorus atom [39] showed up to 92% *ee* in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethylmalonate in acetonitrile, 83% *ee* for acetylacetone in MeCN:H₂O (4:1), and up to 85% *ee* for benzylamine in MeCN:H₂O (4:1) mixture. The catalytic system could be separated easily by simple acid-base extraction and reused at least two times.

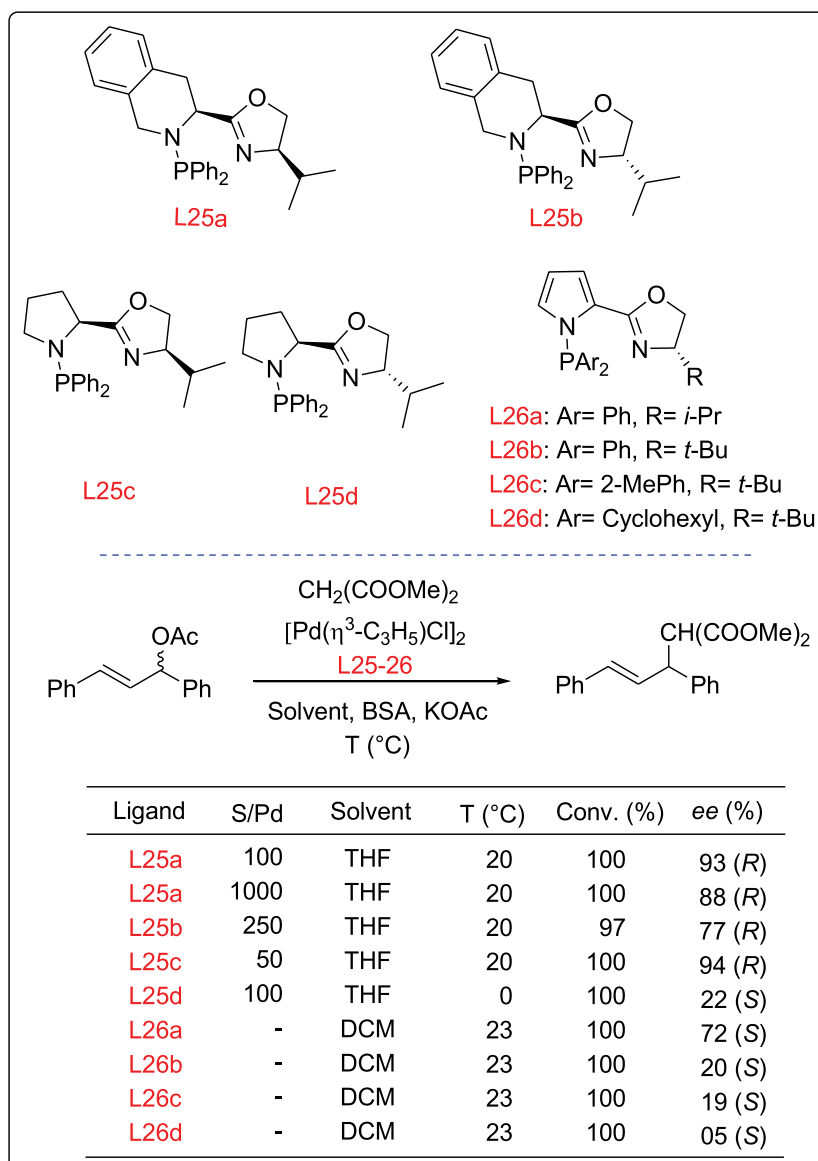
Though the ligands **L9** and **L10** are ineffective in allylic substitution of the hindered 1,3-diphenyl-2-propenyl acetate, they showed low enantioselectivity for unhindered cyclic and linear substrates. In this regard, Dieguez and coworkers [40] reported a new family of carbohydrate-based phosphite-oxazoline ligands (**L11-L15**) for allylic substitution reactions. These ligands contain a phosphite group instead of a phosphinite group.

The incorporation of a phosphite moiety into the ligand offers the following advantages. (i) The bulky biphenyl phosphite provides a larger bite angle than phosphinites. (ii) The biphenyl moiety offers more flexibility and can be used to fine-tune the chiral pocket formed upon the complexation of substrates. (iii) Incorporating a phosphite moiety also decreases substrate specificity because the chiral pocket in which the allyl moiety embedded is smaller than for ligands **L9**, but flexible enough to allow the perfect coordination of both hindered and unhindered substrates. (iv) These ligands provide more flexibility that can be finely tuned by phosphite and oxazoline substituents to explore their effect on the catalytic performance. (v) The rate of reaction increases because of the high π -acceptor capacity of the phosphite moiety. (vi) The regioselectivity towards the desired branched isomer in mono-substituted linear substrates increases because the π -acceptor capacity of the phosphite moiety enhances the S_N1 character of the nucleophilic attack.

The ligands were prepared by reacting the corresponding sugar oxazoline-alcohols obtained from D-glucosamine with the corresponding biaryl phosphorochloridite in the presence of pyridine [41]. All the ligands were analyzed by elemental analyses, ¹H, ¹³C, and ³¹P NMR spectroscopy. All the ligands (0.9 mol %) were screened for the allylic substitution reactions of 1,3-diphenyl 2-propenyl acetate with dimethyl



Scheme 13. IndPHOX ligands used in the asymmetric allylic alkylations.



Scheme 14. AminPHOX ligands for asymmetric allylic alkylations.

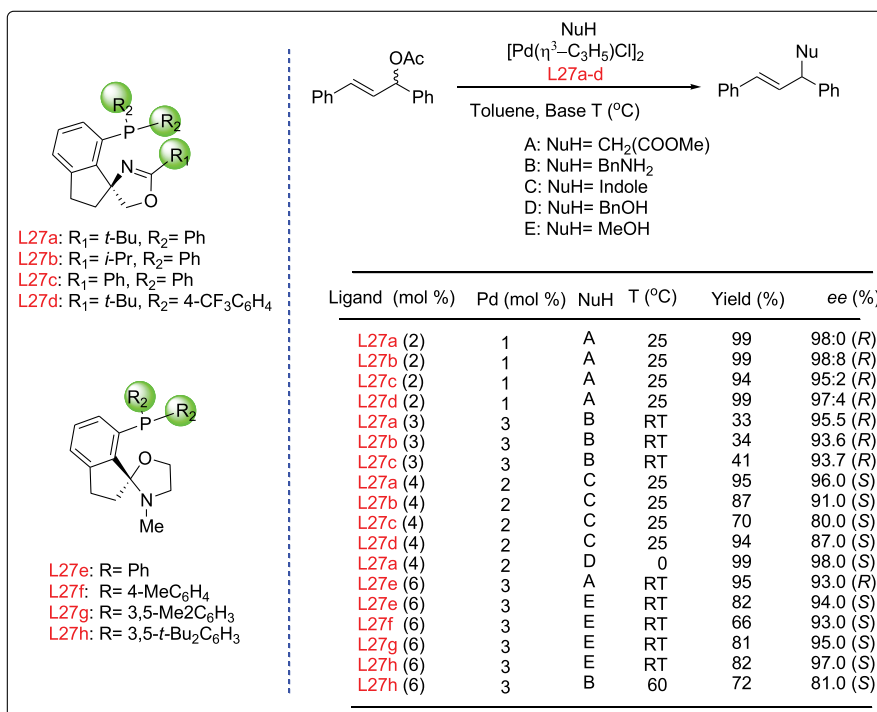
malonate and benzylamine in the presence $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (0.5 mol %), BSA, and KOAc as a base in DCM at 23 °C (Scheme 9).

Thus, the ligand **L11a** (0.9 mol %) containing the *t*-butyl group present at both the *ortho* and *para* positions of the biphenyl-phosphite moiety showed the highest activity (100% conversion) as well as enantioselectivity (92% *ee*) for dimethyl malonate in 30 min. The enantioselectivity of **L11a** was further increased by lowering the temperature to 0 °C (95%) and changing the solvent from DCM to toluene with increased Pd loading to 2.0 mol % (99%) in 6 h. When the excess of a ligand is present, the phosphite-oxazoline ligand acts as a monodentate ligand. Excellent enantioselectivity was acquired by using the ligand-to-palladium ratio of 0.9. Though, 94% enantioselectivity was observed for benzylamine the activity was lower (60%) than the alkylation reaction of dimethyl malonate in the presence of 1.0 mol % Pd and 1.8 mol % ligands. As compared to other substituents, a phenyl substituent on the oxazoline ring (**L11a**) showed the highest enantioselectivity (99% *ee*). These results are in contrast with the ligand **L10**, in which the oxazoline-containing methyl substituent showed the highest enantioselectivity. The activities of the ligands were affected by substituents at the *ortho*

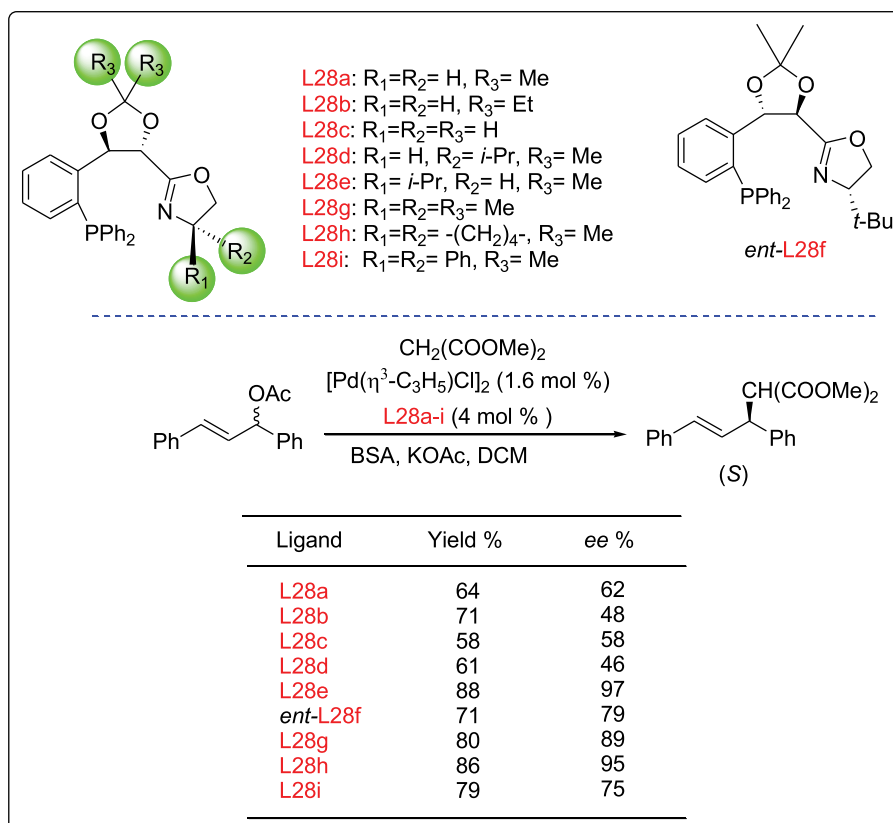
positions of the biphenyl moiety. While enantioselectivities were mainly affected by the substituents at the *para* positions of the biphenyl moiety. Thus, the best result was shown by ligand **L11a**, which contains the optimal combination of ligand parameters. The ligand **L14h** showed the best enantioselectivity (up to 89% *ee*) for less sterically demanding 1,3-dimethyl 2-propenyl acetate using dimethyl malonate as a nucleophile. Thus, ligand **L11a** (1.8 mol %) provided the best enantioselectivity (up to 91% *ee*) for the allylic substitution of unsymmetrical 1,1-diphenyl-1-hepten-3-yl acetate using dimethyl malonate as a nucleophile at 23 °C using 1 mol % $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$.

Ziegler and coworkers synthesized spiro-fused ligands containing diphenylphosphino groups (**L16a-i**) from D-fructose [42]. All the ligands showed good to excellent conversion of 1,3-diphenyl 2-propenyl acetate and dimethyl malonate into desired products with an *er* ranging from 76:24 to 84:16 in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$, BSA, and KOAc in various solvents at 0 °C to room temperature (Scheme 10).

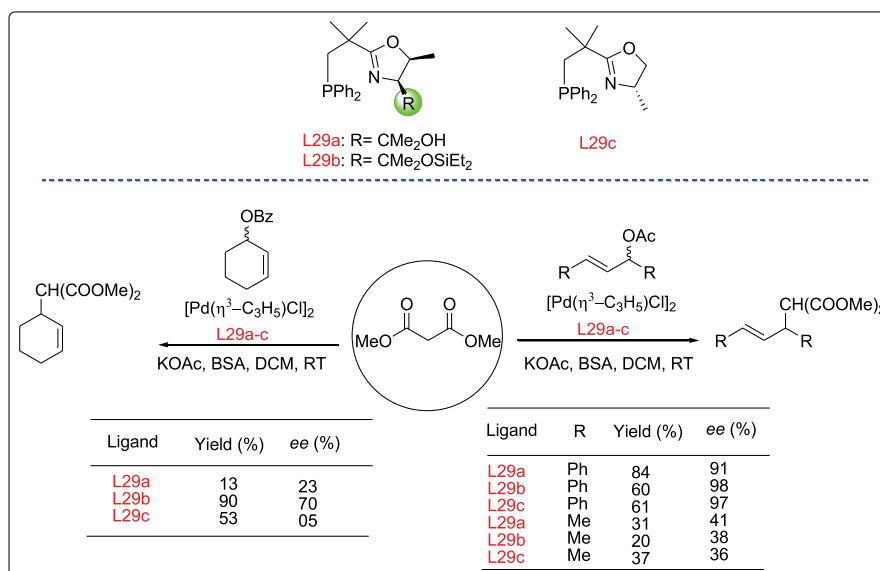
It was observed that the bulky substituent at the 3 position and small substituents at the 4 and 5 positions of fructose positively influence the stereoselectivity. The ligand containing pivaloyl groups (**L16e**) provided



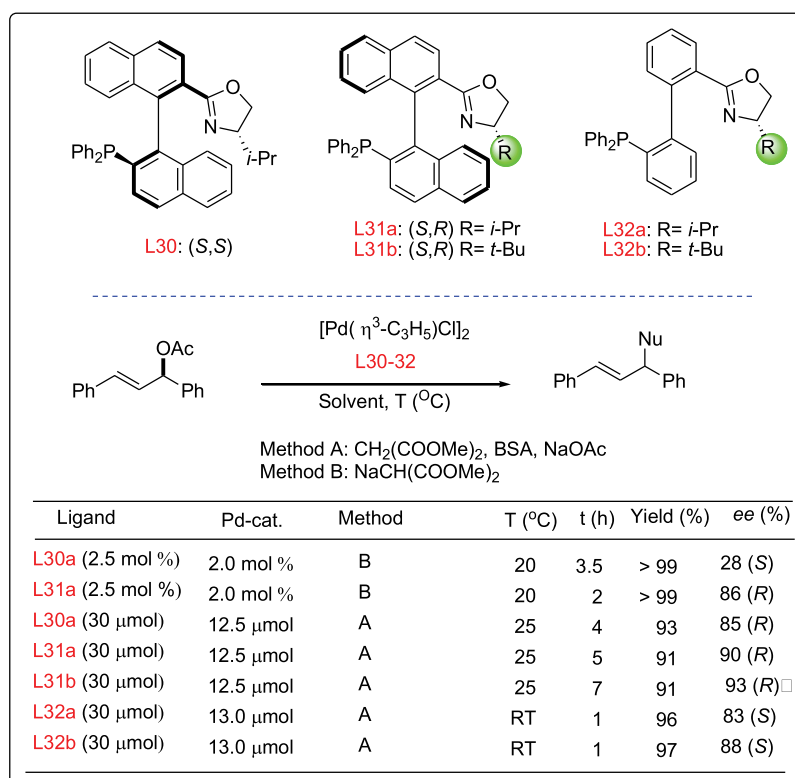
Scheme 15. SMIPHOX ligands used for asymmetric allylic amination, etherification reactions.



Scheme 16. StePHOX for asymmetric allylic alkylations.



Scheme 17. NeoPHOX ligands for asymmetric allylic alkylations.



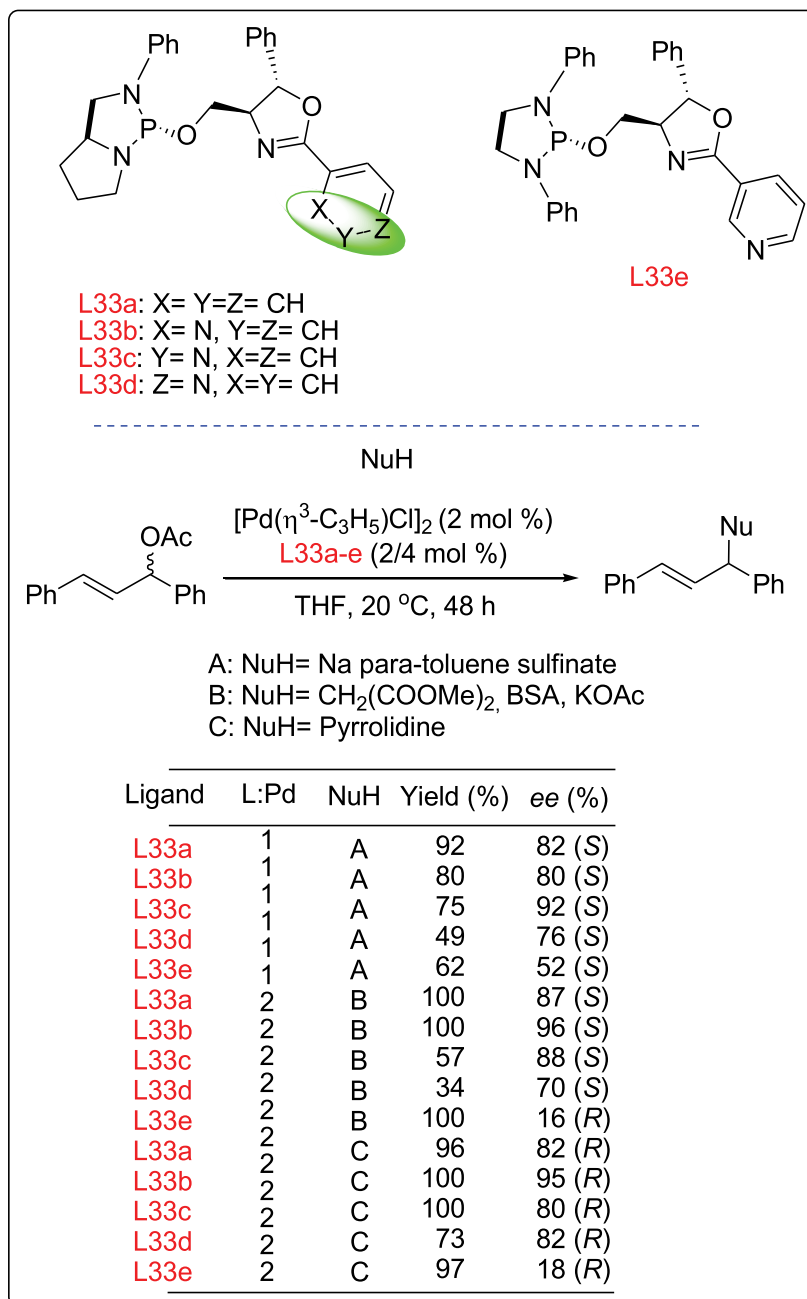
Scheme 18. BinaPHOX ligands for asymmetric allylic alkylations.

an *er* ratio of 89:11 in toluene at room temperature. They developed a second-generation D-fructose-based spirofused PHOX ligands (**L17a-b**) having bulky benzyl groups at 3 positions of the fructose moiety. The ligands attained high *er* up to 93:7.

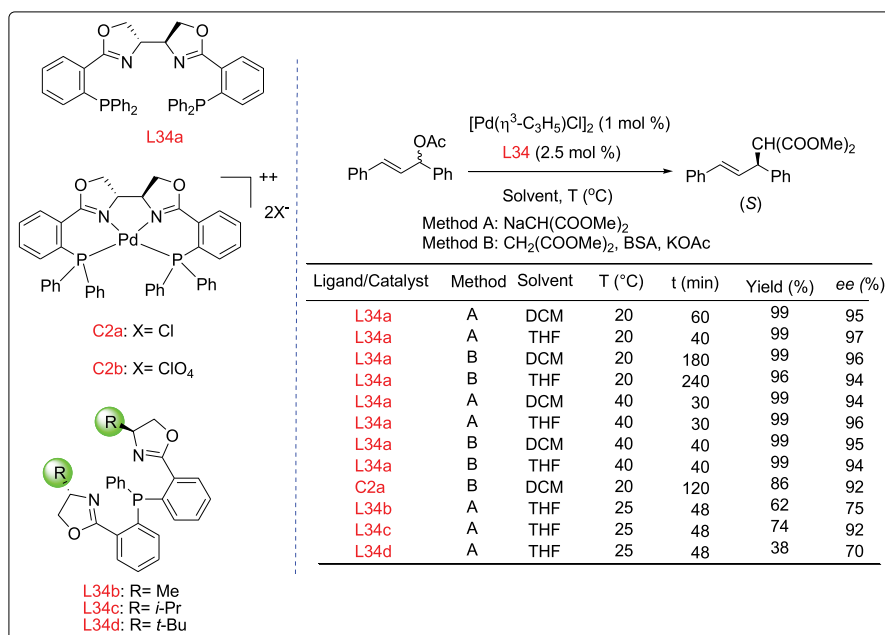
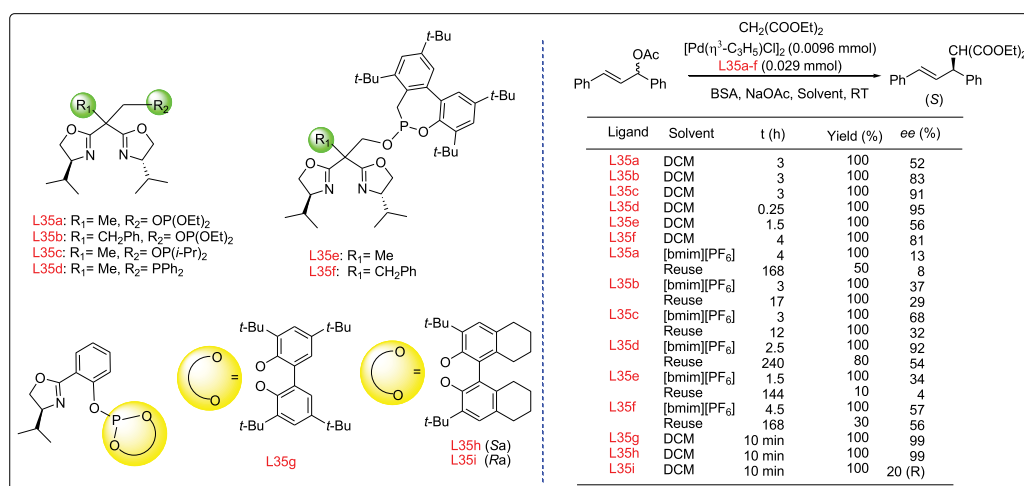
P-N ligands containing heterocycle PHOX (HetPHOX) such as ThioPHOX, IndanolPHOX, IndPHOX, and AminPHOX have emerged as modular ligands in asymmetric catalysis. HetPHOX allows the easy introduction of the required diphenylphosphino groups on the ligand

backbone. A series of HetPHOX based on thiophene and benzothiophene-oxazoline-containing ligands were reported for Pd-catalyzed asymmetric catalysis.

Tietze et al. [43a] and Cozzi et al. [43b] synthesized various chiral thiophene, benzothiophene, and benzofuran-based HetPHOX ligands (**L18-21**). In these ligands, substituents can be introduced at different positions either by direct metallation or by bromination followed by successive halogen metal exchange reactions. All the ligands (6 mol %)



Scheme 19. N, N-diamidophosphite–oxazoline ligands for asymmetric allylic alkylations.

Scheme 20. C₂-symmetric Phos-Biox ligand for asymmetric allylic alkylations.

Scheme 21. Bis(oxazolanyl)-type PHOX ligands for asymmetric allylic alkylations.

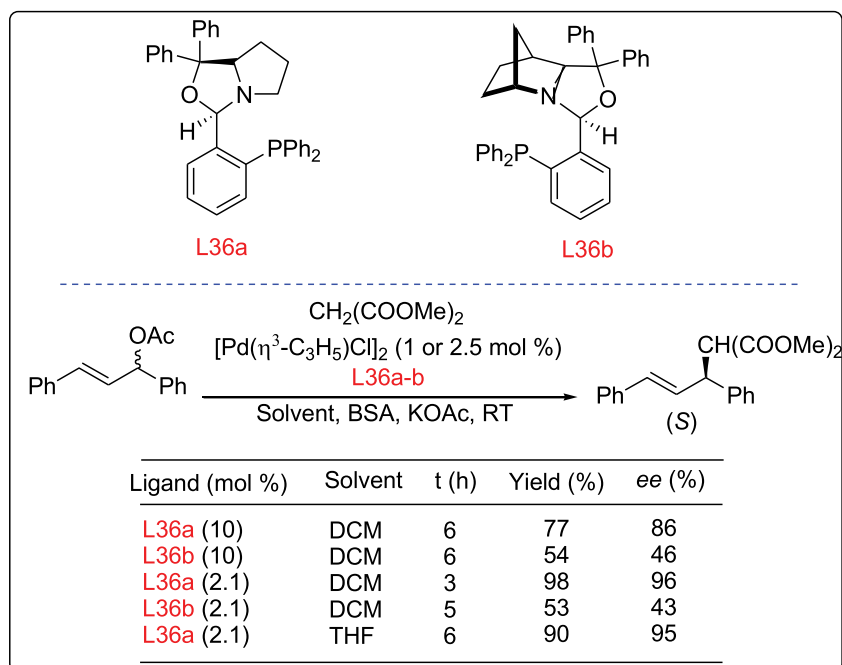
were applied in the asymmetric allylation of 1,3-diphenyl 2-propenyl acetate with Na-dimethyl malonate or dimethyl malonate using [Pd(η^3 -C₃H₅)Cl]₂ (2.5 mol %) catalyst in THF. It was observed that the ligand **L20a** with an isopropyl group at the oxazoline moiety gave 92% allylation product with 97% *ee* at 0 °C in 2 h, while the ligand **L20b** containing the *t*-butyl group led to the 91% desired product with 94% *ee* at -10 °C (Scheme 11). Similarly, these ligands (3 mol %) also gave up to 90% yield and 99% *ee* for dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (1.5 mol %) in DCM at room temperature.

Saigo and coworkers [44] introduced a bulky indanol-based PHOX ligand (**L22**) derived from non-natural chiral amino alcohol and *cis*-2-amino-3,3-dimethyl-1-indanol. The ligand **L22** (3.6 mol %) was found to be an efficient ligand for enantioselective allylic amination of different 1,3-alkyl/aryl 2-propenyl acetates with benzylamine or potassium phthalimide in presence of Pd₂(dba)₃ (1.5 mol %) in THF (Scheme 12). In the amination reaction of 1,3-diphenyl-2-propenyl acetate, the ligand

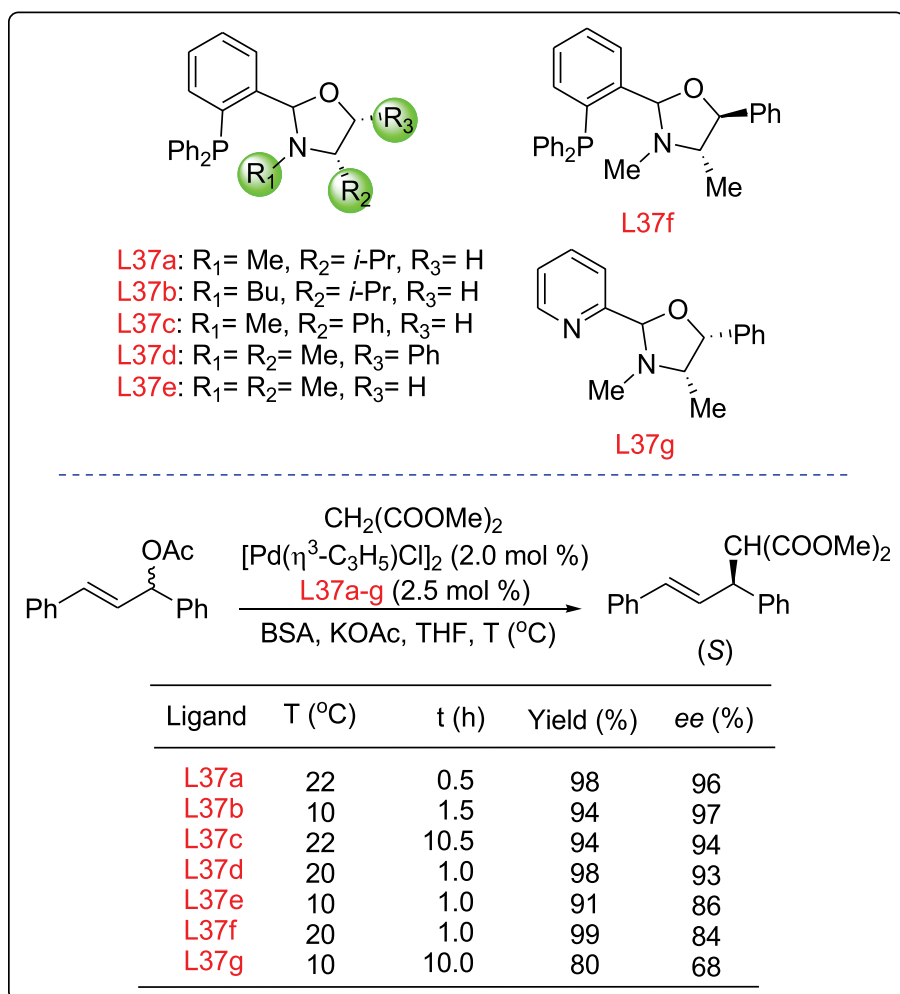
L22 was more efficient than similar ligands.

Franzén and coworkers utilized two types of indole-based HetPHOX ligands called 'IndPHOX'. Class I ligands **L23a-d**, and class II ligands **L24a-d** (12 mol %) were applied to the asymmetric allylic alkylation of 1,3-diphenyl-2-propenyl acetate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (4 mol %), KOAc, and BSA in THF at room temperature [45]. In amination, the ligands containing diphenylphosphine group at 2-position and oxazoline moiety at 3-position of the indole-moiety **L24a-d** (12 mol %) showed the best results (up to 98% yield and 99% *ee*) [46] (Scheme 13).

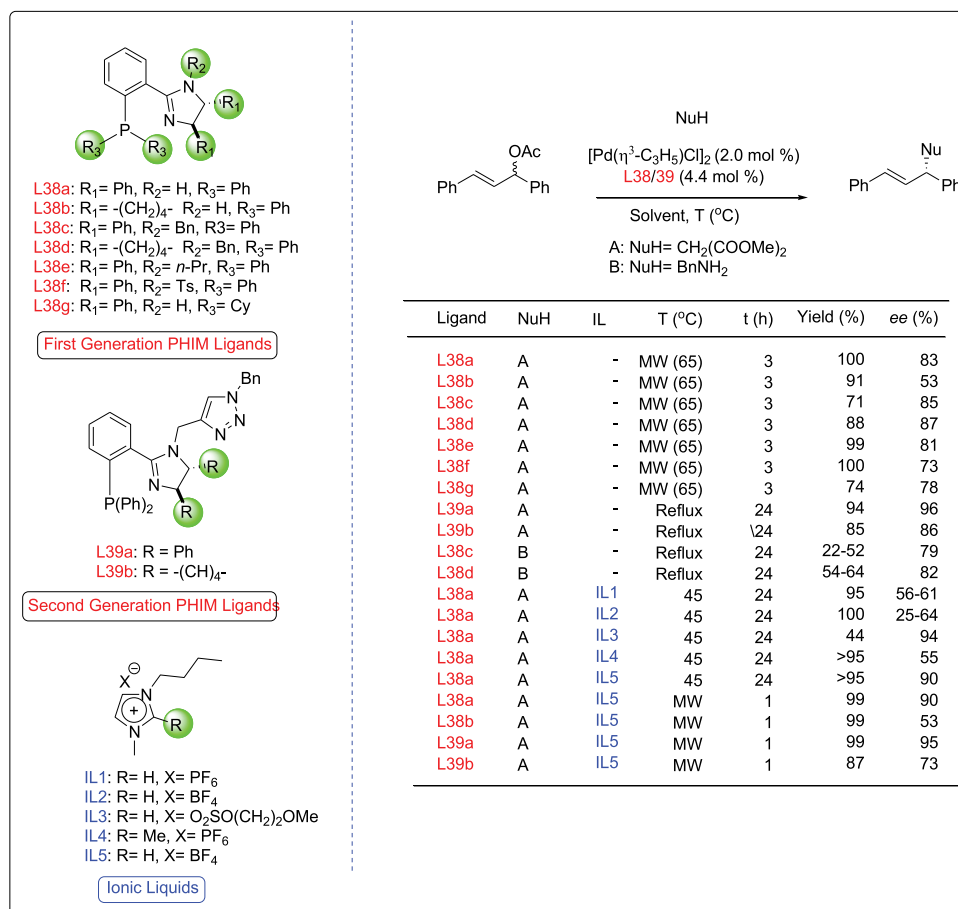
Two-fold amino acid-based chiral aminophosphine-oxazoline ligands (AminPHOX) based on tetrahydroisoquinoline carboxylic acid (**L25a-b**) and proline (**L25c-d**) was reported by Niedercorn et al. [47a]. The ligand **L25a** showed 93% *ee* while ligand **L25c** showed 94% *ee* in the allylation of 1,3-diphenyl-2-propenyl acetate using [Pd(η^3 -C₃H₅)Cl]₂, BSA, and KOAc in THF at 20 °C (Scheme 14). Even improved enantioselectivity was shown in MeCN at 0 °C. Similarly, chiral



Scheme 22. Pyrrolidinyl and 2-azanorbonylphosphinooxazolidine ligands for asymmetric allylic alkylations.



Scheme 23. Ephedrine-derived chiral phosphinooxazolidine and pyridinoxazolidine ligands for asymmetric allylic alkylations.



Scheme 24. Phosphine-imidazoline (PHIM) ligands for asymmetric allylic alkylations.

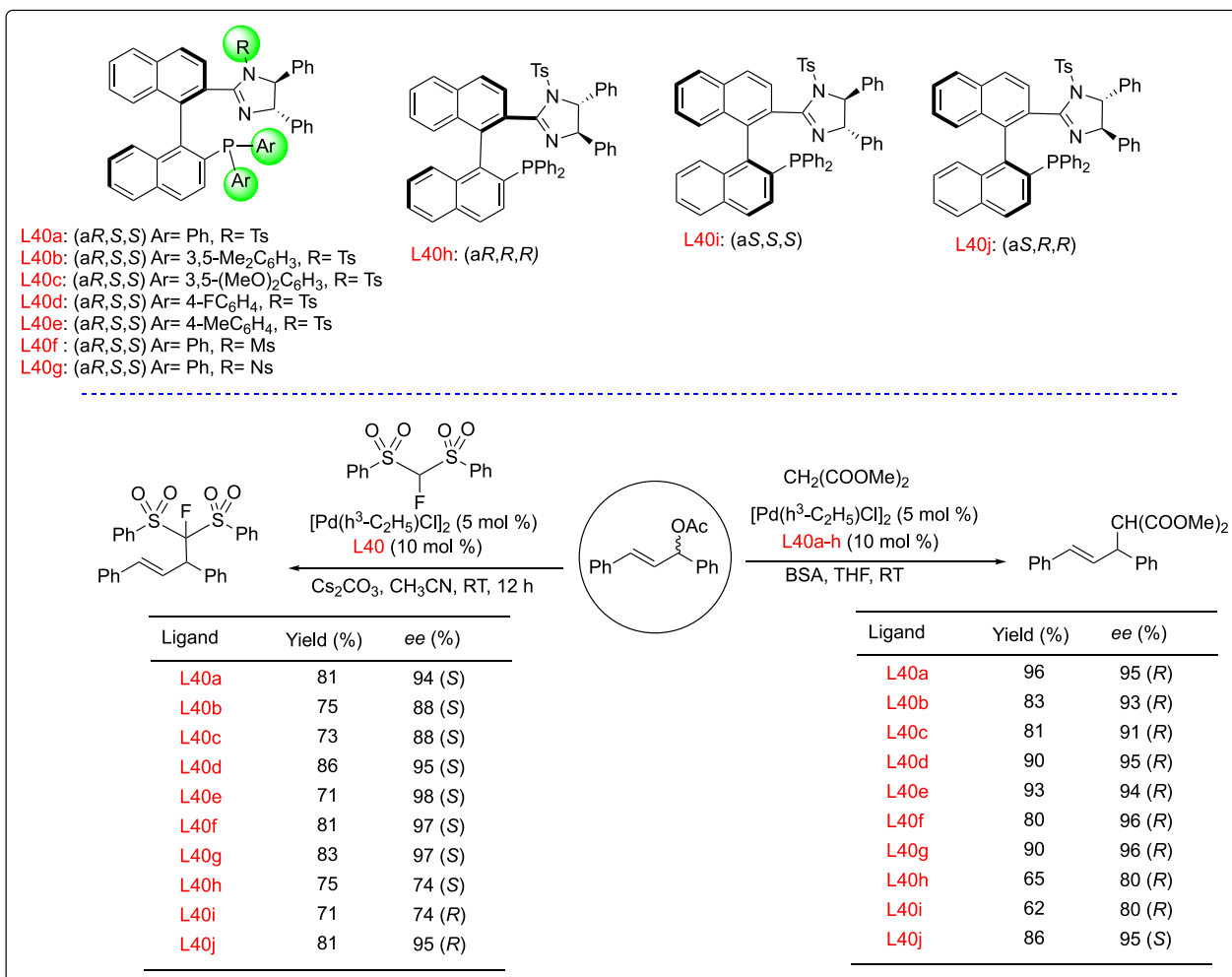
phosphinopyrrolyl-oxazolines (PyroPHOX) ligands (**L26a-d**) (2.4 mol %) [47b] also showed moderate *ee* in the allylation of 1,3-diphenyl-2-propenyl acetate using [Pd(η^3 -C₃H₅)Cl]₂ (1 mol %) in DCM at 23 °C.

The ligands containing the 'spiro' backbone have been recognized as a privileged motif in many catalytic reactions. Such a rigid scaffold of the ligands could reduce the conformational obscurity of the catalyst and create an effective asymmetric environment around the central metal. In this context, a new type of HetPHOX ligands called 'SIPHOX' (SpirobiIndane PHOX), 'SpinPHOX' (Spiro[4,4]-1,6 nonadiene PHOX), and 'HMSI-PHOX' (HexaMethyl-1,10-SpirobiIndane PHOX) have been developed for Ir-catalyzed asymmetric hydrogenation reactions and Ni-catalyzed asymmetric arylations [48]. In this connection, Teng and co-workers introduced 'SMIPHOX' (Spiro Mono-Indane PHOX) ligands (**L27a-d**) for various Pd-catalyzed asymmetric reactions [49]. The ligands were synthesized from commercially available 7-bromo-1-indanone and (*R*)-phenylglycinol and applied in the reaction of 1,3-diphenyl-2-propenyl acetate and dimethyl malonate in the presence of BSA and KOAc in toluene at different temperatures (-10 to 25 °C) (Scheme 15). The [Pd(η^3 -C₃H₅)Cl]₂/L27a-d (1:2 mol %) catalytic system afforded the desired product in high yields (94–99%) and enantioselectivities (95.2–98.8 % *ee*). All the ligands showed good yields with excellent *ee* for various substrates. They extended this protocol for enantioselective allylic alkylation of indoles and allylic etherification [50a]. The ligands **L27a-d** (4.0 mol %) showed 87–95% yields and 87–96% enantioselectivity for the reaction of indole with 1,

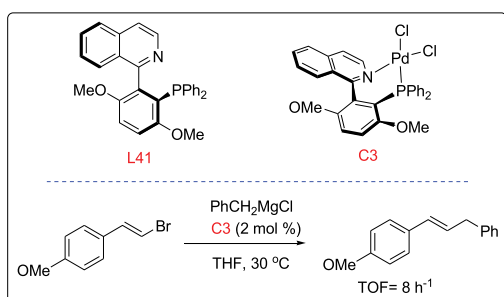
3-diphenyl-2-propenyl acetate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2 mol %) using Cs₂CO₃ as a base in toluene at room temperature. However, yield and enantioselectivity are slightly low for the ligand **L27c**. The ligand **L27a** also worked efficiently in the allylic etherification of primary and secondary alcohols. The desired products formed with excellent enantioselectivity (93–99% *ee*) in the presence of Cs₂CO₃ in DCM at 0 °C. Similar type of amino phosphine ligands (**L27e-h**) bearing Spiro[indane-1,2'-pyrrolidine] backbone synthesized in gram scale [50b]. The ligand **L27h** (6 mol %) gave the highest *ee* of 97% with 82% yield in allylic alkylation of 1,3-diphenyl-2-propenyl acetate with methanol using [Pd(η^3 -C₃H₅)Cl]₂ (3 mol %) and Cs₂CO₃ as the base in toluene at room temperature.

Trudeau and Morken [51] developed a new class of optically active PHOX ligands called 'StePHOX' (**L28a-h**) containing different ligand backbones as well substitution patterns on the oxazoline and dioxolane moiety (Scheme 16). The dioxolane linkage offers additional chirality and tuning element to the PHOX ligands. The ligands **L28a-d** lead to moderate enantioselectivity (46–62% *ee*) in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂, KOAc, and BSA in DCM. While ligands **L28e** and **L28h** attained excellent enantioselectivity (97 and 95% *ee*).

In 2016, Pfaltz and coworkers [52] introduced serine and threonine-derived neopentyl backbone containing ligands (**L29a-c**) called 'NeoPHOX' for the allylic substitution of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂,



Scheme 25. Chiral imidazoline-phosphine ligands for asymmetric allylic alkylations.

Scheme 26. Pd-QUINAP catalyst used in coupling reaction of vinyl halide with PhCH₂MgCl.

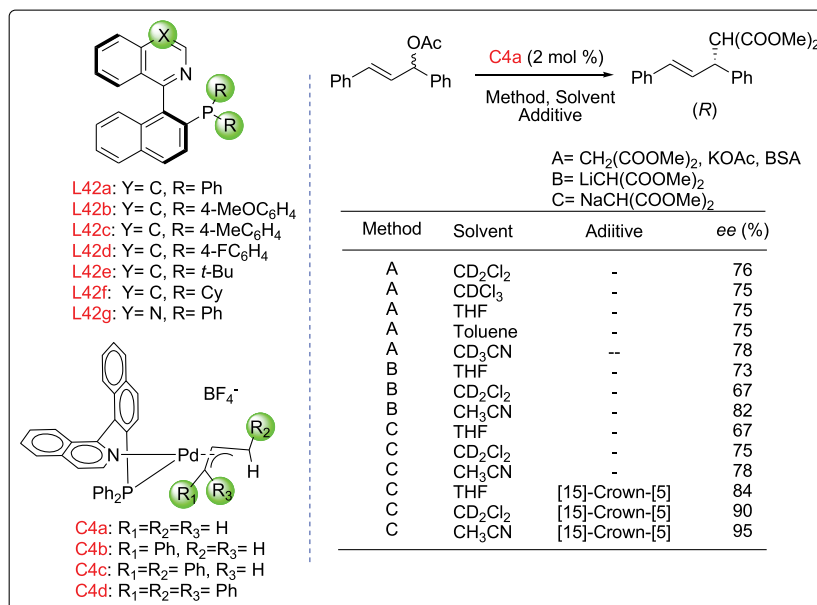
KOAc, and BSA in DCM (Scheme 17). The ligands **L29b** and **L29c** with a free hydroxyl group and the trimethylsilyl-protected derivative afforded 98% and 97% *ee*, respectively, while **L29a** provided 91% *ee*. Unfortunately, all the tested ligands were less efficient for 1,3-diphenyl-2-propenyl acetate. On the other hand, ligand **L29b** achieved 70 % *ee* for cyclohex-2-en-1-yl benzoate.

A novel axially chiral PHOX ligands **L30** [53], and **L31a-b** [54] with binaphthyl backbone as well as axis-unfixed biphenyl backbone

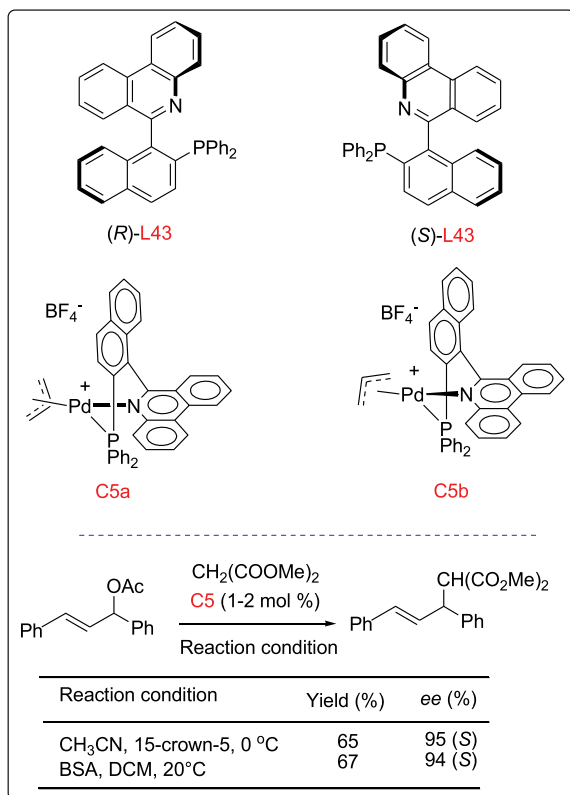
(**L32a-b**) [55] were applied for Pd catalyzed allylic substitution reactions (Scheme 18). All the ligands gave high chemical yields (up to 99%) and enantioselectivity (up to 93% *ee*) for the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of [Pd(η³-C₃H₅)Cl]₂ in different solvents at different temperatures.

Gavrilov et al. [56] developed a small library of readily available N, N-diamidophosphite-oxazoline ligands (**L33a-e**) containing 1,3,2-diazaphospholidine and a pyridine moiety for Pd-catalyzed asymmetric allylation reactions (Scheme 19). The ligand **L33c** (2 mol %) provided up to 92% *ee* (*S*-product) for allylations of 1,3-diphenyl-2-propenyl acetate with sodium *para*-toluene sulfinate in the presence of [Pd(η³-C₃H₅)Cl]₂ (2 mol %) in THF at 20 °C. The ligand **L33b** (4 mol %) provided 96% *ee* for dimethylmalonate in BSA-KOAc and DCM at 20 °C. The 95% *ee* was attained for the amination of 1,3-diphenyl-2-propenyl acetate with pyrrolidine using **L33b** (4 mol %) in THF at 20 °C. The ligands bearing P and C stereocenters in the 1,3,2-diazaphospholidine rings were more efficient. However, the ligands **L33b-c** with pyridin-2-yl or pyridin-3-yl substituents were the most efficient, while the ligand **L33e** containing achiral diamidophosphite showed much lower enantioselectivities.

Lee and coworkers [57a] reported the C₂-symmetric conformationally rigid bisphosphinobioxazoline [(*S*, *S*)-Phos-Biox] ligand (**L34**). In the presence of [Pd(η³-C₃H₅)Cl]₂ (1 mol %), the ligand (2.5 mol %) exhibited very high efficiency (94–97% *ee*) for the reaction of 1,3-diphenyl-2-propenyl acetate and dimethyl malonate or its Na salt with almost quantitative chemical yields of (*S*) product in THF or DCM (Scheme 20).



Scheme 27. Applications of QUINAP in asymmetric allylic alkylations.



Scheme 28. Applications of PHENAP in asymmetric allylic alkylations.

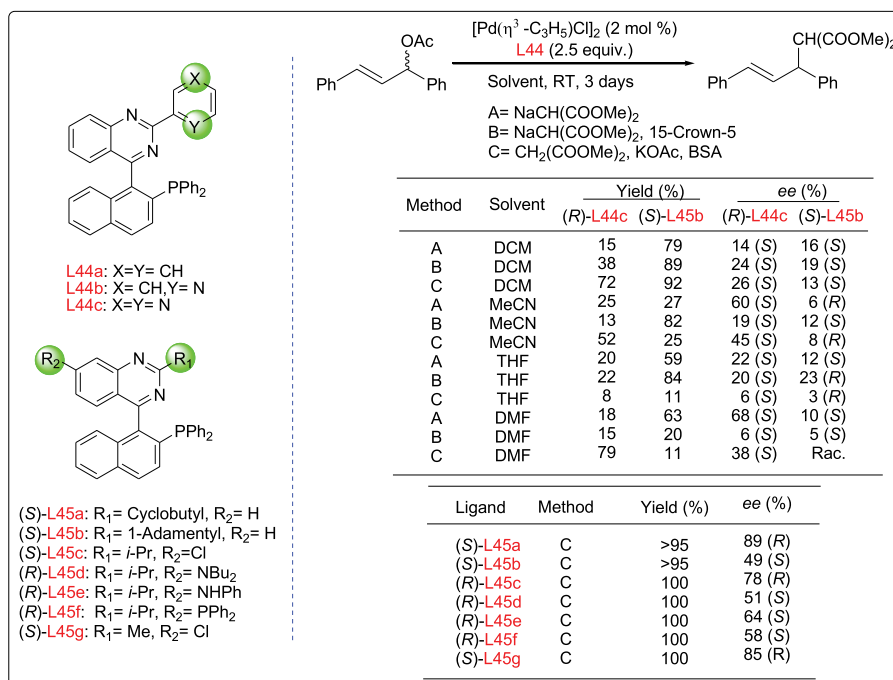
Moreover, the P, N, N, P-chelated complex (**C2a-b**) obtained from **L34** also exhibited high enantioselectivity (92% *ee*) and good yield (86%) under optimized reaction conditions. The enantio discrimination capacity of **L34** was also examined for meso-dibenzoate of *cis*-2-cyclopenten-1,4-diol with Na salt of dimethyl malonate and *N*-benzyl-*N*-methylamine. The catalytic system also carried Pd-catalyzed intramolecular cyclization of biscarbamate of

cis-2-cyclopenten-1,4-diol with *p*-toluenesulfonyl isocyanate (TsNCO) in THF at 50 °C giving the corresponding biscarbamate mixture (1:3) in 90% yield with 50% *ee*. Similarly, N-P-N type PHOX ligands (**L34c-d**) also showed excellent enantioselectivity in asymmetric allylic substitution reactions [57b]. These ligands coordinated to Pd(II) as bidentate ligands to afford Pd(η^3 -C₃H₅)(**L34c-d**)]PF₆ complex. The complex obtained from **L34c** showed high enantioselectivity in the reaction of 1,3-diphenyl-2-propenyl acetate and Na salt of dimethyl malonate in THF.

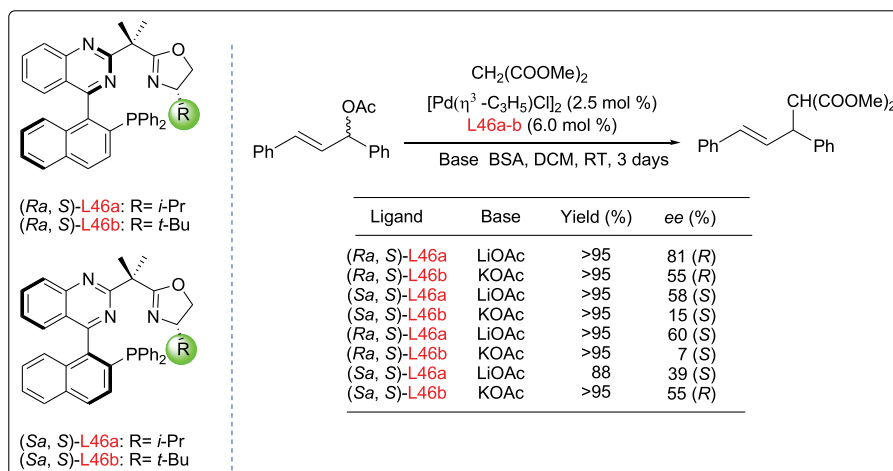
Chiral tridentate ligands have a large diversity of binding modes for metals and allow fine-tuning of the catalytic system. In this regard, Castillo and coworkers [58a] introduced tridentate chiral NPN-based PHOX-type ligands (**L35a-f**) for the asymmetric allylic substitution reactions (Scheme 21). The reaction involves the alkylation of 1,3-diphenyl-2-propenyl acetate with diethyl malonate in either DCM or [bmim]PF₆ in the presence of BSA and NaOAc. The phosphorus-based ligands showed increased catalytic activity, leading to total conversions in 4 h. While, analogous bis(oxazoline) and azabis(oxazoline) ligands required 2 to 7 days. Unfortunately, complete recovery of the catalyst and ionic liquid was impossible, and decreased catalytic activity with a drop in enantioselectivity was observed with longer reaction times. A similar type of bidentate PHOX ligands (**L35g-i**) (1.1 mol %) composed of chirally flexible biaryl phosphite showed excellent enantioselectivities in allylic substitutions of a wide range of substrates and nucleophiles in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (0.5 mol %) in DCM at room temperature [58b].

Hongo and coworkers [59] synthesized pyrrolidiny and 2-azanorbornyl phosphinooxazolidines (**L36a-b**) from 2-(diphenylphosphino) benzaldehyde and corresponding amines in refluxing toluene. The effectiveness of the ligands was examined in asymmetric allylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate (Scheme 22). The ligand **L36a** (2.1 mol %) gave an excellent yield (98%) and *ee* (96%) in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (1 mol %) and BSA-KOAc in DCM at room temperature. Even high enantiomeric excess (95%) was observed in THF. Ligand **L36b** was less effective (53% yield and 43% *ee*) under similar reaction conditions.

Jin and coworkers [60] developed ephedrine-derived chiral phosphinooxazolidine (**L37a-f**) and pyridinoxazolidine (**L37g**) ligands for the asymmetric allylic substitution reactions of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate, in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2.5 mol %) in BSA-KOAc and THF at 0–20 °C (Scheme 23). The ligand



Scheme 29. Applications of Quinazolinap ligands in asymmetric allylic alkylations.



Scheme 30. Applications of Quinazox in asymmetric allylic alkylations.

L37e having a methyl group at C4 showed lower enantioselectivity (86% *ee*) than L37a and L37b containing a bulky *i*-Pr group (98% and 97% *ee*). Compared to ligand L37e, the ligand L37d, having a phenyl group at C5, enhanced the *ee* up to 93 %, while ligand L37f showed lower enantioselectivity (84% *ee*). Even the ligand L37g showed low *ee* (68%) in THF. The ligand L37a afforded the alkylation product with 98% *ee*, while moderate enantioselectivity (60% *ee*) and yield (51%) were observed in the amination with benzylamine. Predominantly (S)-product was formed due to the favorable nucleophilic attack, to a carbon atom of the π -allyl moiety at the trans position to the oxazolidine nitrogen.

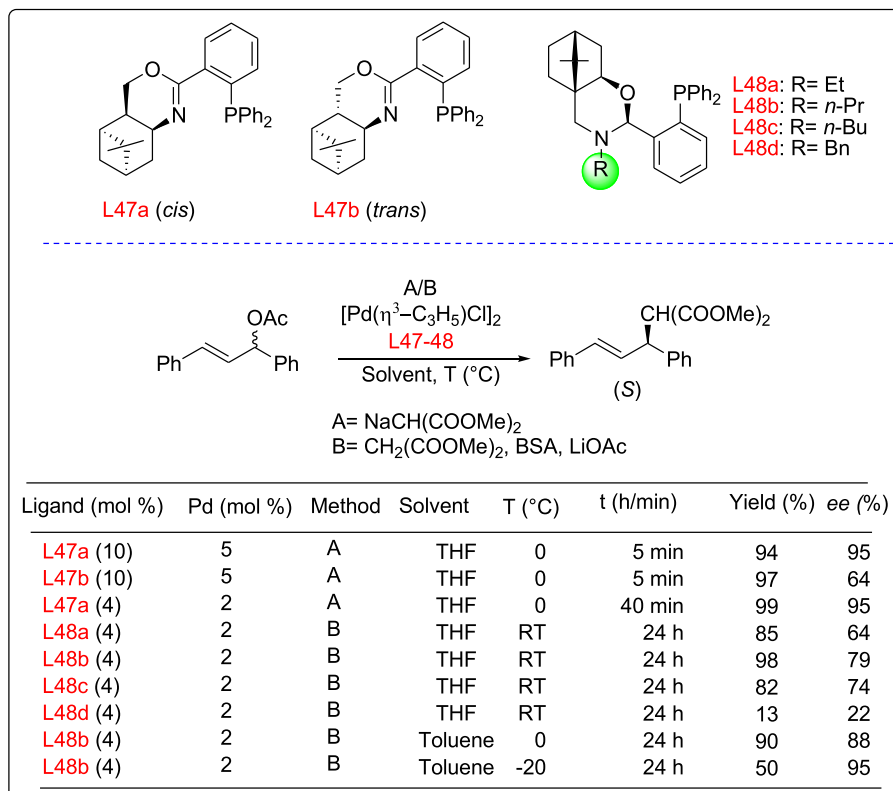
Phosphine-imidazoline (PHIM) ligands analogous to that BIPI-ligands have been acting as interesting ligands in asymmetric catalysis [61]. These ligands can be synthesized easily from readily available C₂-symmetric diamine fragments. The ligand contains a second nitrogen atom that can work as an additional source of molecular diversity, as it allows

the adjustment of the electronic properties of the coordinating nitrogen atom. Furthermore, it is possible to immobilize it on a suitable solid support.

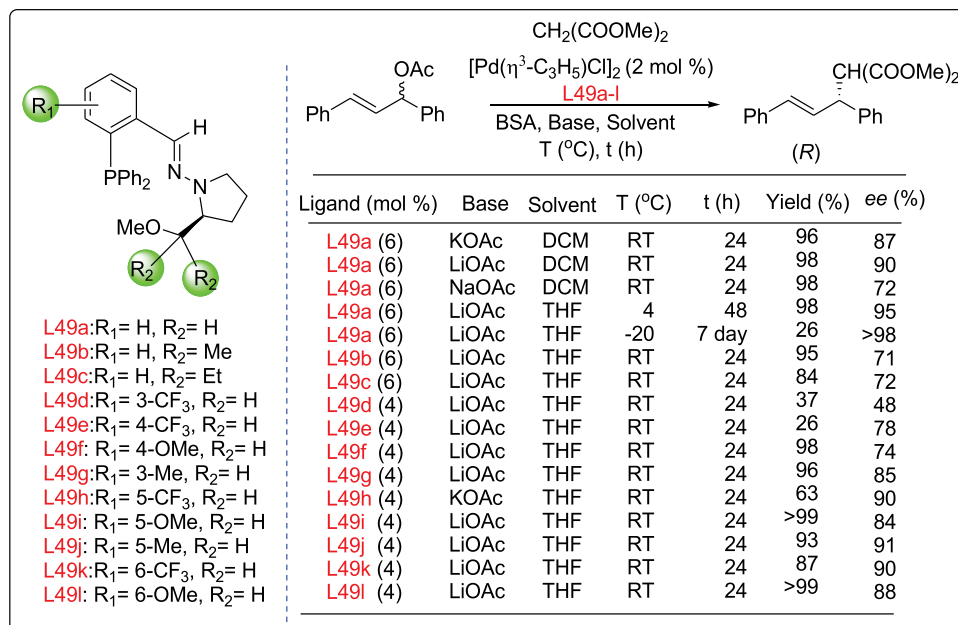
These ligands are of two types;

- First-generation PHIM ligands (L38a-g): It involves various substituents on the C4 and C5 (R₁) of imidazoline moiety, the non-coordinating nitrogen atom of the imidazoline ring (R₂), and the phosphine moiety (R₃).
- Second generation PHIM ligands (L39a-b): These ligands result from the introduction of triazolymethyl moieties as nitrogen substituents (R₂).

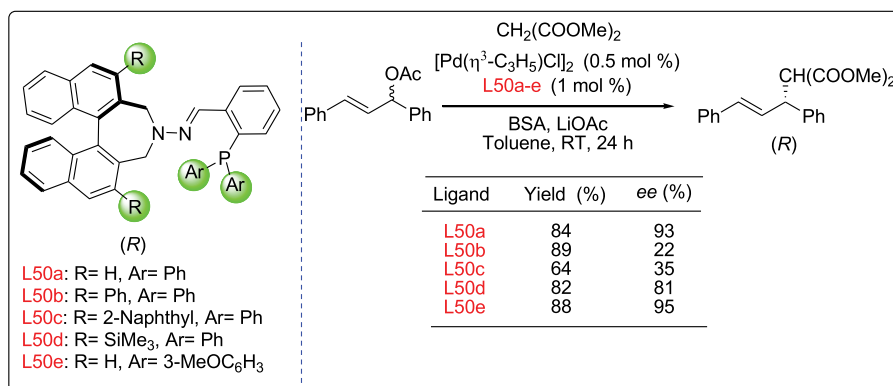
All these ligands were applied for asymmetric alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate or benzylamine [62] (Scheme 24). The ligands L38c-d gave the best enantioselectivity



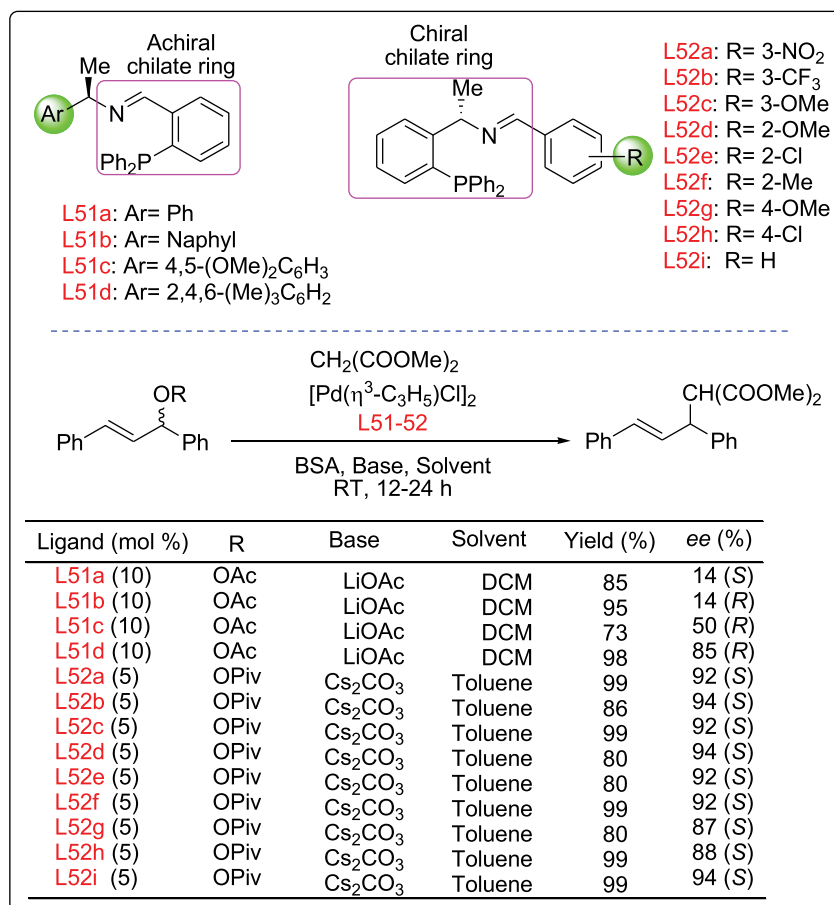
Scheme 31. Phosphine-oxazinanone ligands for asymmetric allylic alkylations.



Scheme 32. DPPB-SAMP ligands for asymmetric allylic alkylations.



Scheme 33. Phosphino hydrazone with a pendant binaphthyl unit as a chiral modifier used in asymmetric allylic alkylations.



Scheme 34. Phosphine-imine ligands for asymmetric allylic alkylations.

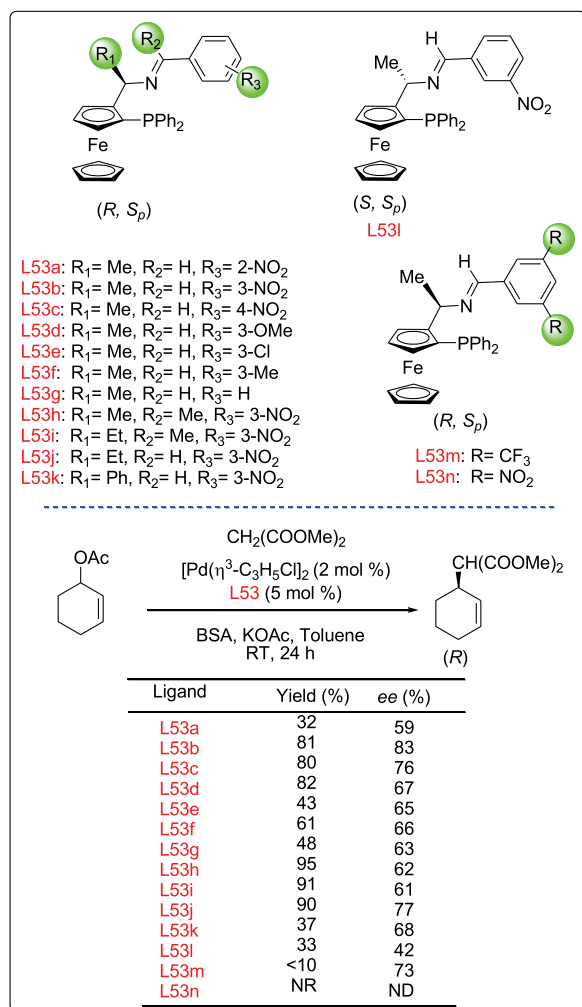
(85% and 87% ee, respectively) in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5\text{Cl})_2]$ (2 mol %) and KOAc-BSA in DCM, under MW at 65 °C.

The activity and enantioselectivity of ligands **L38a-b** and **L39a-b** were also studied in different ionic liquids (**IL1-5**) under conventional as well as MW heating. The complete conversion with 90% enantioselectivity was observed in **IL5** using **L38a** (4.4 mol %) at 45 °C. A comparison of ligands showed that ligand **L39a** in **IL5** achieved 95% ee under MW irradiation. Moreover, **IL5** and ligand **L38a** or **L39a** could be successfully recycled at least four times under MW irradiation. The conversion and enantioselectivity dramatically decreased after the third

run.

Similarly, Shi et al. [63] used chiral imidazoline-phosphine ligands (**L40a-j**) with a 1,1'-binaphthalene framework for allylic alkylation (Scheme 25). For the reaction of 1,3-diaryl-2-propenyl acetate with dimethyl malonate, excellent enantioselectivity (up to 96%) was observed except for **L40h** and **L40i** ligands. In addition, except **L40b-c** and **L40h-i** all the ligands showed up to 98% enantioselectivity in monofluoromethylation of 1,3-diphenyl-2-propenyl acetate with 1-fluorobis(phenylsulfonyl)methane.

The 'Atropos P-N' ligands (APN) are a specific type of biaryls with



Scheme 35. Ferrocene-based phosphine-imine ligands for asymmetric allylic alkylations of cycloalkenyl esters.

one component carrying a pendant diaryl substituted phosphine, and the other bears an sp^2 N-atom adjacent to the biaryl link. The substituents in the biaryl inhibit rotation about the linking bond and form a stable, non-racemizing six-membered chelate ring. In their seminal reports, Guiry et al. [64] and other workers [65] described the evolution of APN type ligands like 'QUINAP', PINAP, and QUINAZOLINAP' as privileged chiral ligands for many asymmetric transformations.

The first new class of hybrid, axially chiral (1-(2'-

diphenylphosphino-3',6'-dimethoxy phenyl)isoquinoline) (**L41**) ligand that creates a six-membered chelate ring was reported by Alcock and coworkers [66a]. The reaction of the **L41** with $(MeCN)_2PdCl_2$ in DCM gave the corresponding racemic chelate Pd-complex (**C3**) having the *cis*-chelate square planar structure as analyzed by X-ray crystallography. The complex **C3** (2 mol %) effectively catalyzed the coupling of vinyl halide with $PhCH_2MgCl$ leading to the formation of stilbene in THF with a TOF = $8 h^{-1}$ at 30 °C (Scheme 26).

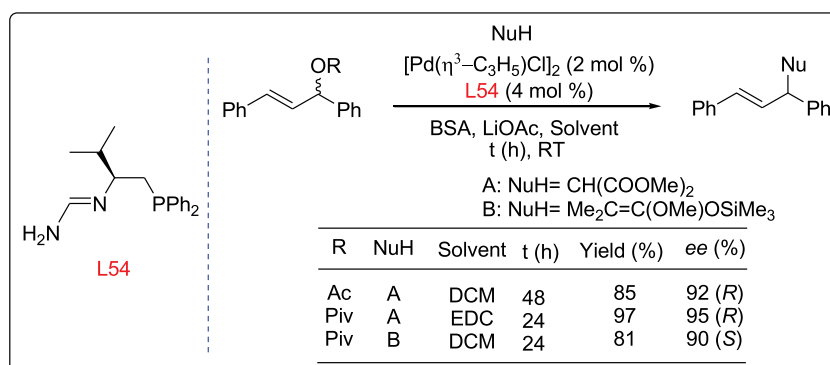
The difference in the electronic nature of N and P atoms would give a potentially more labile intermediate showing different reactivity in asymmetric catalysis. But this ligand racemizes at ambient temperature, which limits its application in asymmetric catalysis. To increase the barrier to rotation around the chiral axis Alcock and coworkers reported 'QUINAP' **L42a** containing a naphthalene-isoquinoline backbone by analogy to 'BINAP' for the asymmetric catalysis (Scheme 27) [66b].

The allylpalladium tetrafluoroborate complexes (**C4a-d**) of Quinap ligand were prepared by reacting $[Pd(\eta^3-C_3H_5)Cl]_2$ with mono, 1,3-di, and 1,1,3-triphenyl analogs [67]. The structures of complexes **C4a-c** were confirmed by electrospray MS. Quinap has continued to be successfully employed in several allylic alkylation reactions. Most of the reactions were carried out by using **C4a** (2 mol %) as well as preparing it *in situ* by reacting **L42a** and $[Pd(\eta^3-C_3H_5)Cl]_2$. Initial screening of **C4a** was carried out with 1,3-diphenyl-2-propenyl acetate and dimethyl malonate in BSA or its Li/Na salts in various solvents. At ambient temperature, asymmetric induction was attained between 67% and 82%. The addition of 15-Crown-5 in DCM significantly enhanced the enantioselectivity to *ee* 90%. Furthermore, *ee* was 95% in acetonitrile at ambient temperature in the presence of 15-Crown-5, which is get enhanced to 97.8% at 0 °C, and 98.2% at -13 °C.

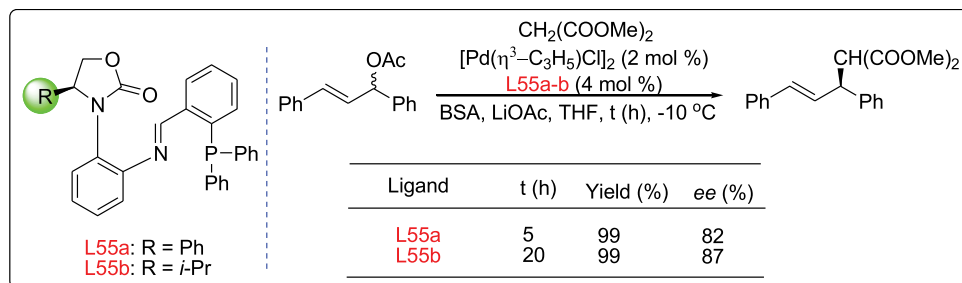
Brown et al. [68] synthesized a new (*R*) and (*S*)-6-(2'-diphenylphosphino-1'-naphthyl) phenanthridines called phosphinamine (PHENAP) (**L43**). It involves additional fused arene rings. It plays a critical part in defining the coordination sphere through its steric pressure. The ligand was synthesized by starting the Suzuki coupling of 6-chlorophenanthridine with 2-methoxy-1-naphthaleneboronic acid. Excellent enantioselectivity was obtained for allylic alkylation of 1,3-diphenyl-2-propenyl acetate (Scheme 28). The enantioselectivity is similar to the QUINAP ligand. (*R*)-**L43** derived catalyst (**C5a-b**) gives rise to *S*-product.

The 1H NMR spectroscopic studies of Pd-QUINAP complexes showed that in allylic alkylation, the 3-position of the ligand of the isoquinoline unit takes up a position in space. That leads to ligand-reactant steric interactions, which could be important for asymmetric induction. This was the principle behind the design and the preparation of PHENAP (**L43**), which in turn gave high enantioselectivity in Pd-catalyzed allylic substitutions.

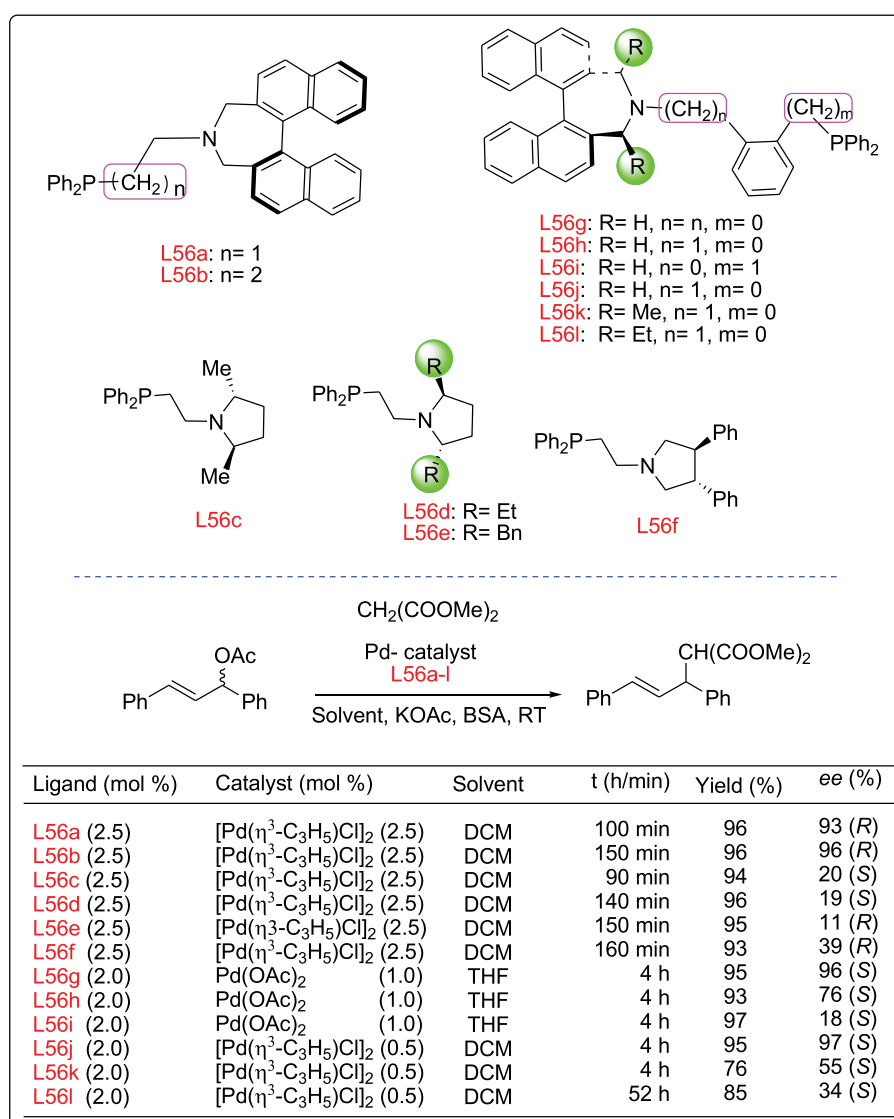
In this connection, Guiry and coworkers introduced a new class of axially chiral phosphinamine ligand called 'QUINAZOLINAP' (**L44a**), containing a sterically bulky group at the 2-position of the quinazoline ring [69]. The ligand was synthesized by the Suzuki coupling of 4-chloro-2-phenylquinazoline and 2-methoxy-1-naphthylboronic acid in high

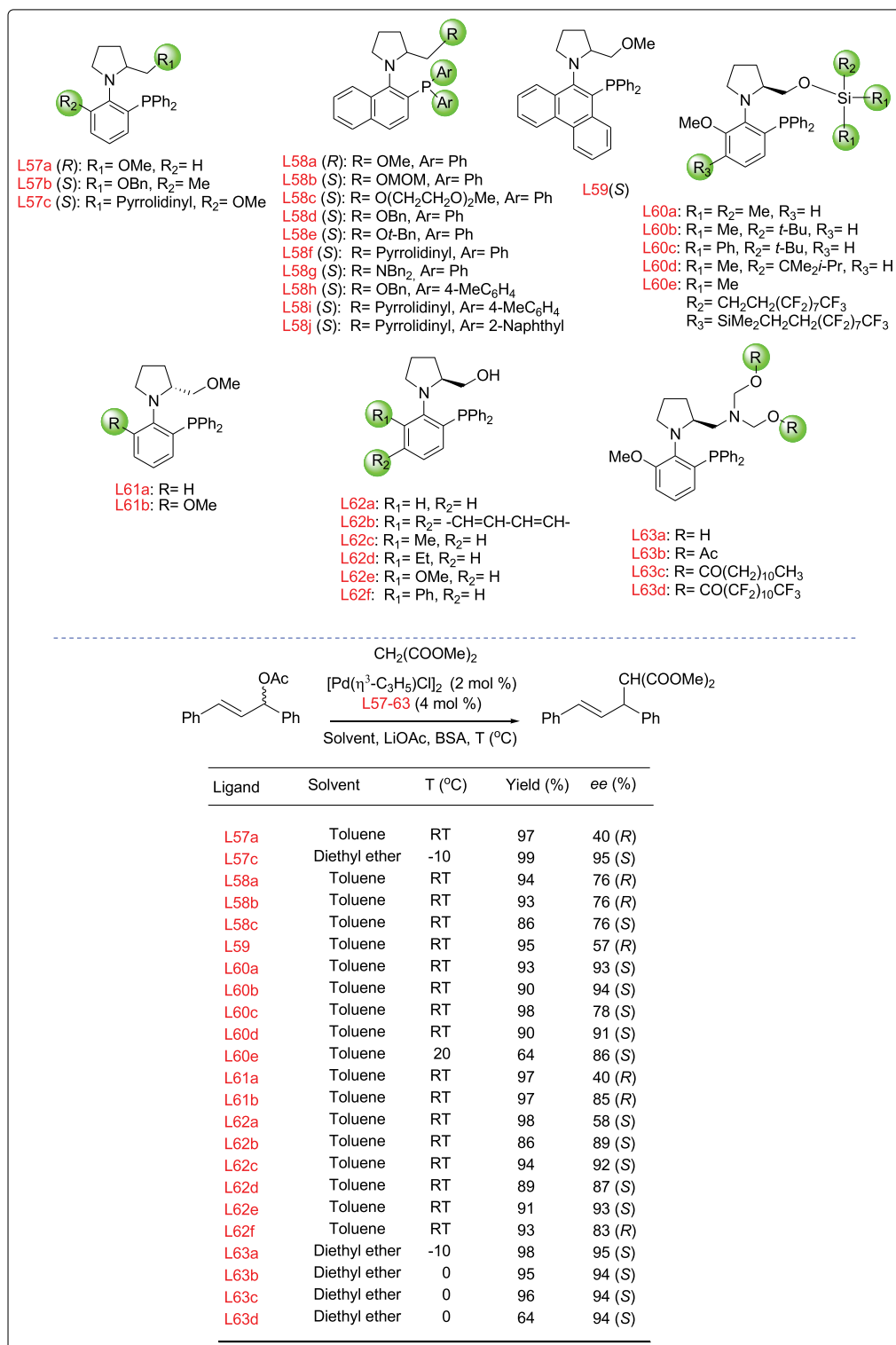


Scheme 36. Phosphorus-containing chiral amidine ligand in the asymmetric allylic alkylations.

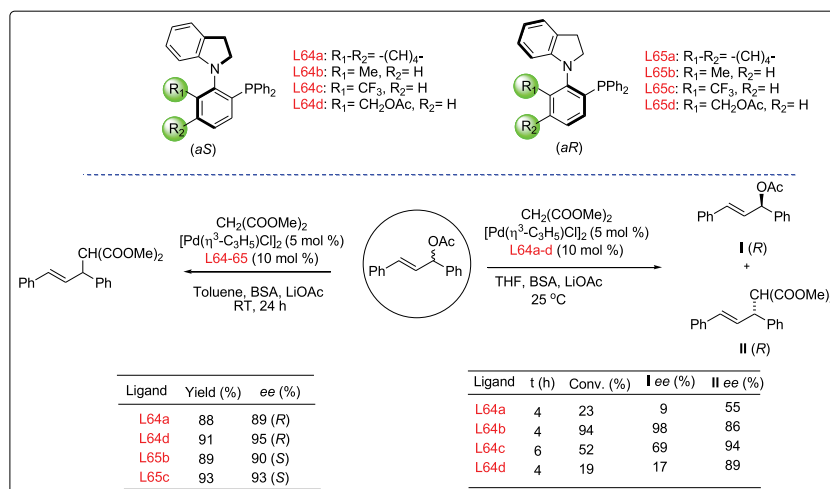


Scheme 37. Evans-type P, N ligands for asymmetric allylic alkylations.

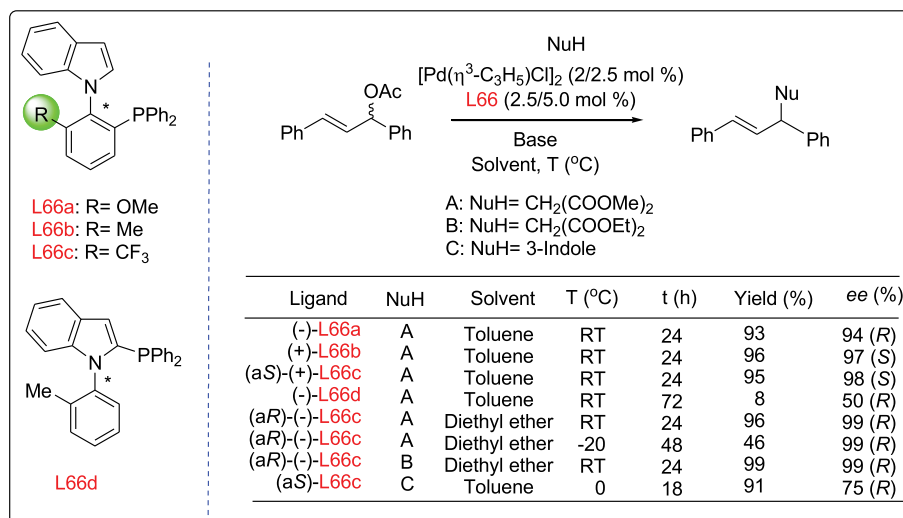
Scheme 38. Asymmetric allylic alkylations using C_2 -symmetric aminophosphines.



Scheme 39. Asymmetric allylic alkylations using pyrrolidine-derived chiral aminophosphines.



Scheme 40. Indoline-derived aminophosphines for asymmetric allylic alkylations and kinetic resolution of esters.



Scheme 41. Indole-derived aminophosphines for asymmetric allylic alkylations.

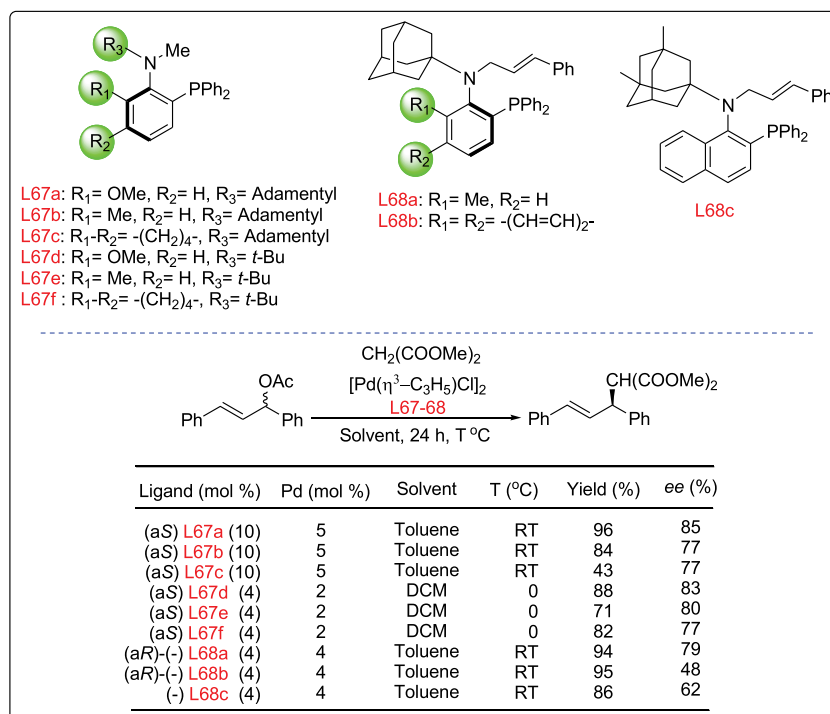
yield (83%) [70]. Similarly, they used two new atropisomer phosphinamine ligands, 2-(2-pyridyl)-Quinazolinap (*S*)-L44b and 2-(2-pyridyl)-Quinazolinap (*R*)-L44c for asymmetric allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate under two standard methods (Scheme 29) [71]. In some cases, 15-Crown-5 has been added to aid the dissolution of the preformed sodium malonate. Both the ligands showed low conversions and enantioselectivities, possibly due to the hemilabile nature of the nitrogen atom of the 2-substituent.

Similarly, the Quinazolinap ligands containing a 2-cyclobutyl ring (*S*)-L45a and 2-adamantyl ring (*S*)-L45b (4.3 mol %) were applied for allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate using [Pd(η³-C₃H₅)Cl]₂ (2.5 mol %), BSA and various bases and solvents at room temperature [72]. The ligand (*S*)-L45a furnished the alkylated (*R*) product with 95% conversion and 89% ee in KOAc and DCM, while (*S*)-L45b provided the alkylated (*S*) product with 95% conversion with 49% ee in Cs₂CO₃ and DCM. The ligand (*S*)-L45a was superior to (*S*)-L45b in enantioselectivity. The stereochemical outcome of the reaction depends on the alkali metal salt used. The ligand (*S*)-L45a containing the larger counterions (K⁺ and Cs⁺) provided the product with high optical purity (89% ee and 87% ee) than smaller counterions Na⁺ and Li⁺ (82% ee and 40% ee).

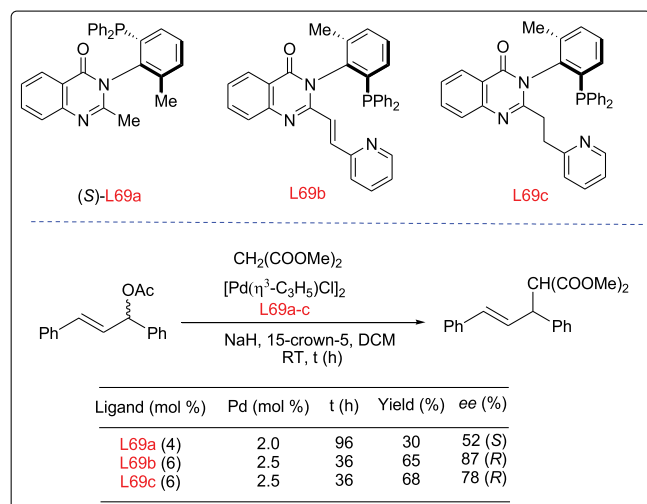
They [73] also developed four new members of the Quinazolinap ligands L45c-g for allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate. Though all the ligands (6 mol %) showed 100% conversions, ligand (*S*)-L45g showed 85% ee of (*R*) product in BSA-KOAc and DCM in the presence of [Pd(η³-C₃H₅)Cl]₂ (2.5 mol %) at room temperature.

'Quinox' (L46a-b) is an analog of 'Quinazolinap' and contains the oxazoline moiety. The ligands were used in Pd-catalyzed asymmetric allylic alkylation. Though all the ligands (6 mol %) showed > 95% conversions, ligand (*Ra*, *S*)-L46a showed 81% ee of (*R*) product in BSA-LiOAc and DCM in the presence of [Pd(η³-C₃H₅)Cl]₂ (2.5 mol %) at room temperature. (Scheme 30) [74].

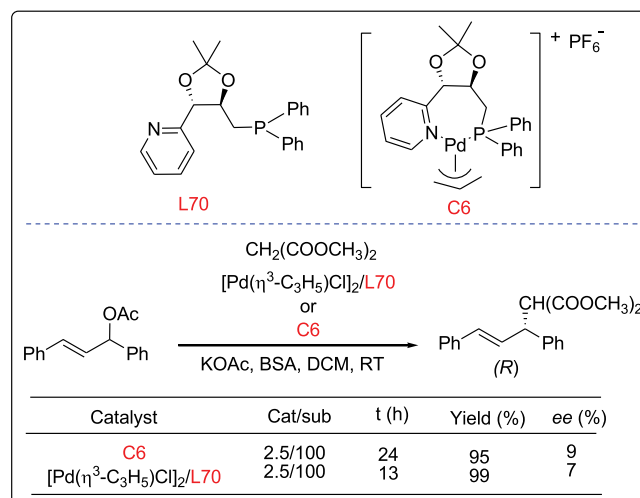
Evans and Brandt [75] obtained *cis* and *trans*-phosphine-1,3-oxazines (L47a and L47b) from readily available materials. The preformed catalyst prepared from [Pd(η³-C₃H₅)Cl]₂ and the L47a-b (ratio: 1:2) was applied for asymmetric allylic alkylation of 1,3-diphenyl-2-propenyl acetate and Na salt of dimethyl malonate (Scheme 31). The ligand L47a (4 mol %) attained 95% ee with a 99% yield of the desired product in DCM at 0 °C. Under similar reaction conditions, chiral phosphine-oxazinane ligands (L48a-d) are also used in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate [76]. The ligand showed 95% ee for L48b.



Scheme 42. Asymmetric allylic alkylations using aniline-based aminophosphines.



Scheme 43. Atropisomeric monodentate and chelate quinazolinone phosphine ligands for asymmetric allylic alkylations.



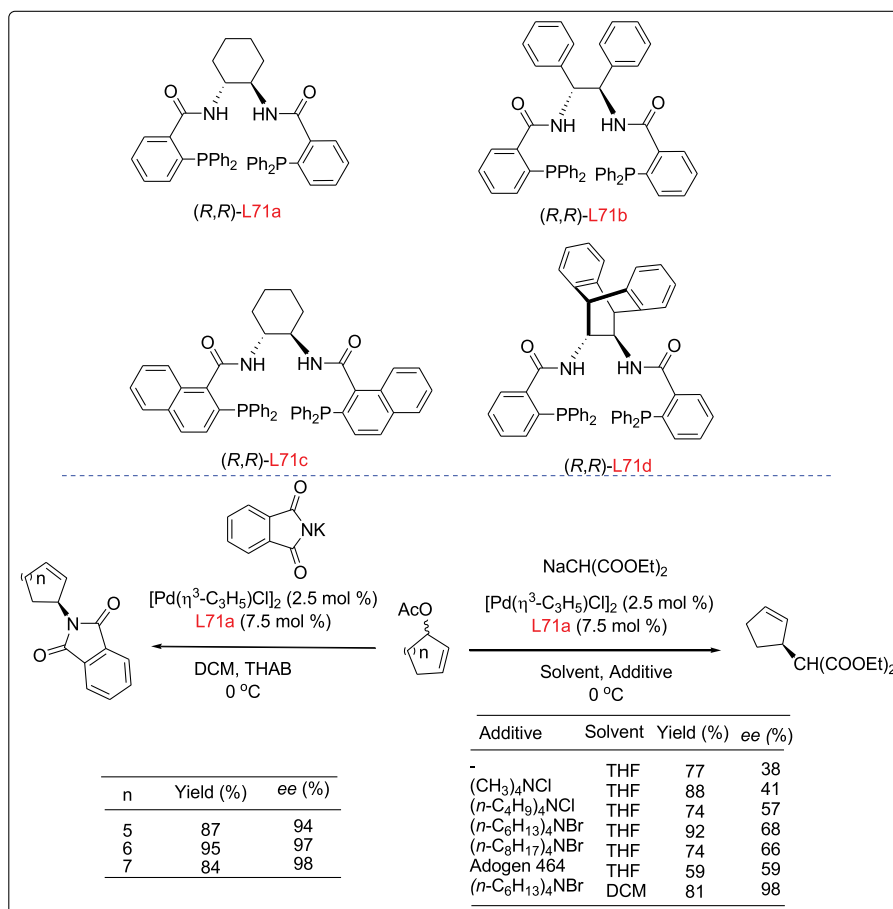
Scheme 44. Homochiral pyridyl phosphino dioxolane derivatives for asymmetric allylic alkylations.

(S)-1-Amino-2-methoxymethyl pyrrolidine (SAMP) and their analogs derived from chiral phosphine hydrazones have been developed as efficient ligands for many Pd-catalyzed reactions [77]. In this regard, extensive work on the Pd-catalyzed asymmetric alkylation reactions using various SAMP hydrazones was reported by Mino and coworkers.

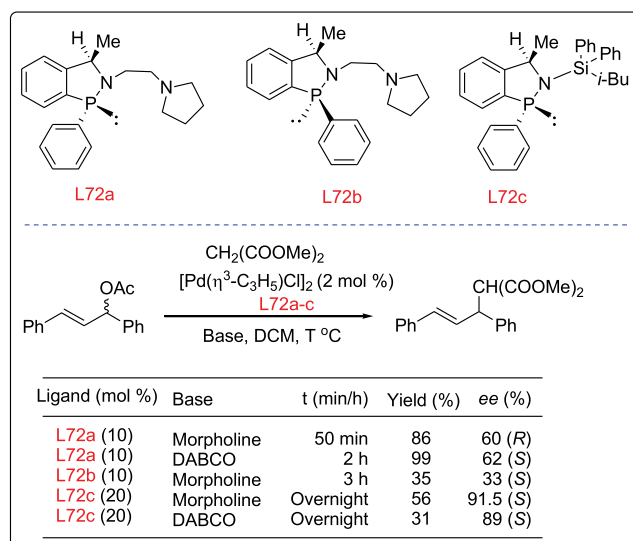
The 2-diphenylphosphinobenzaldehyde-SAMP hydrazone (DPPB-SAMP) ligands were prepared easily from 2-(diphenylphosphino)benzaldehyde and corresponding chiral hydrazines. The ligands **L49a-c** (6 mol %) were applied for asymmetric allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate, BSA-LiOAc using [Pd(η³-C₃H₅)Cl]₂ (2 mol %) in different solvents at room temperature (Scheme 32) [78]. The ligand **L49a** gave predominantly, the (R)-enantiomer with > 98% enantioselectivity. Though superior enantioselectivity (87% ee)

was observed in KOAc, the enantioselectivity increased to 90% in LiOAc. However, enantioselectivity decreased to 72% in NaOAc. In addition, the SAMP hydrazone ligands (**L49d-l**) based on ortho fluoro-benzaldehyde also showed high enantioselectivity [79]. It was observed that chiral hydrazone ligands **L49j** attained high yields with the highest enantioselectivities (91% ee) for the allylic alkylation of 1, 3-diphenyl-2-propenyl acetate with dimethyl malonate in THF at room temperature.

A group of five phosphino hydrazone ligands (**L50a-e**) with a pendant binaphthyl unit as a chiral modifier was synthesized from non-racemic 2,2'-bis(bromomethyl)-1,10-binaphthyl and 3,3'-diiodo-2,20-bis(bromomethyl)-1,10-binaphthyl (Scheme 33) [80]. The [Pd(η³-C₃H₅)Cl]₂/**L50e** (0.5:1.0 mol %) system showed up to 95% ee in allylic



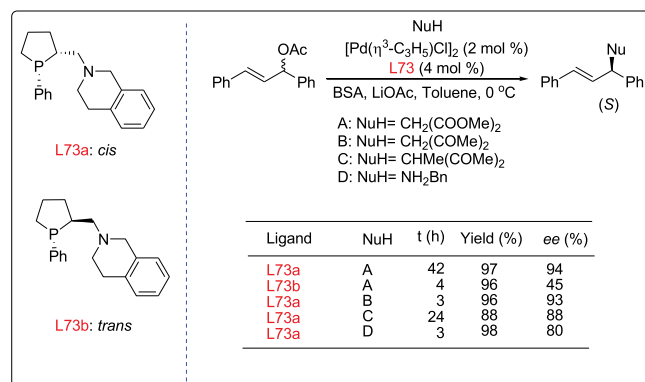
Scheme 45. Trost's chiral amino phosphine ligands for asymmetric allylic alkylations.



Scheme 46. Mono-substituted phosphinuous amide ligands for asymmetric allylic alkylations.

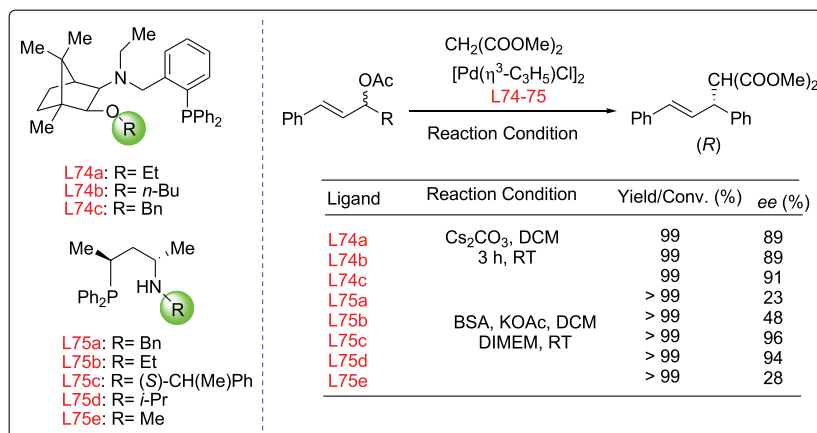
alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate using BSA, LiOAc as a base in toluene at room temperature. The cyclic acetates yielded largely racemic products, and only cycloheptenyl acetate showed good asymmetric induction.

A chiral phosphine-imine ligands derived from 1-mesityl ethylamine (L51a-d) [81a] and α -phenylethylamine (L52a-i) [81b] obtained from

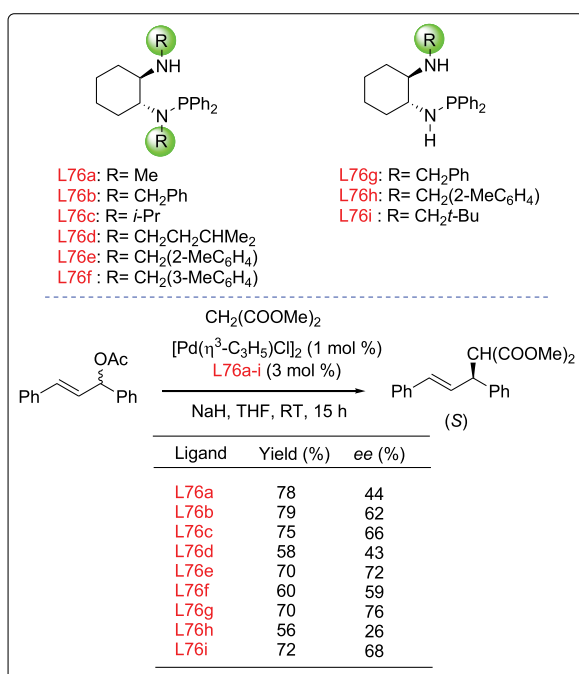


Scheme 47. P, N-ligands containing a tetrahydroisoquinoline skeleton for asymmetric allylic alkylations.

phenylethylamine also applied for Pd catalyzed asymmetric allylic alkylation (Scheme 34). The ligand L51a-b resulted in only poor enantioselectivity, while L51c showed moderate enantioselectivity in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2.5 mol %) BSA, LiOAc in DCM at room temperature. However, due to the steric effect of the substituents on the aromatic ring, L51d showed good enantioselectivity. Similarly, the readily available phosphine-imine ligands (L52a-i) from α -phenylethylamine act as highly efficient ligands for alkylation of 1,3-diphenyl-prop-2-en-1-yl pivalate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2 mol %) BSA, Cs₂CO₃ in toluene at room temperature. All the ligands showed up to 94% ee and 99% conversions. It was



Scheme 48. Camphene-derived aminophosphines for asymmetric allylic alkylations.



Scheme 49. Homochiral amino phosphine ligands for asymmetric allylic alkylations.

observed that the chirality on the chelated ring of ligand **L52** more effectively transfers stereochemical information than ligand **L51**, in which the chirality lay outside the chelate ring.

Zeng et al. [82] used chiral ferrocene-based phosphine-imine ligands (**L53a-n**) for allylic alkylation of cycloalkenyl esters. The ligands substituents of the aryl, ferrocenylmethyl, and benzyliene positions strongly influence the enantioselective induction. Thus the ligand **L53b** (5 mol %) with a nitro group at the *meta* position of the phenyl ring showed the highest enantioselectivity (83% *ee*) in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2 mol %), BSA and KOAc in toluene at room temperature for 24 h, which get increased to 91% using LiOAc at 0 °C for 48 h (Scheme 35).

Phosphorus-containing chiral amidine ligand (**L54**) showed excellent yield (up to 99%) and enantioselectivity (up to 95% *ee*) in the allylic alkylation of 1,3-diphenyl-2-propenyl acetate or pivalate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (0.025-0.05 mol %), BSA-LiOAc in various solvents such as THF or DCM or DCE at room

temperature (Scheme 36) [83]. In addition, this catalytic system also showed excellent yield (up to 95%) and enantioselectivity (up to 93% *ee*) in the alkylations with ketene silyl acetals [84].

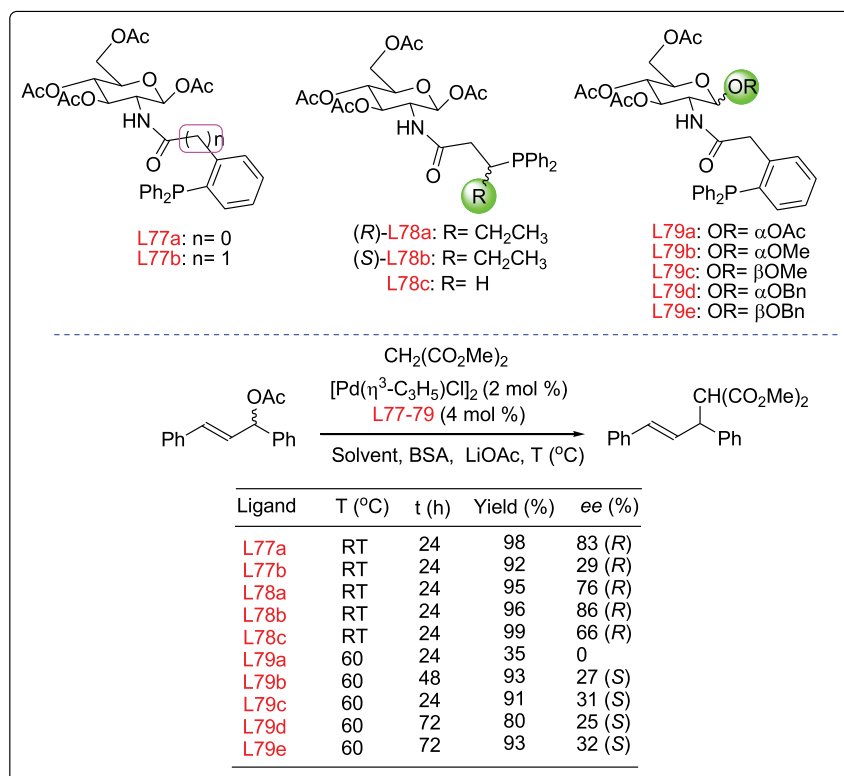
Jiang and coworkers [85] applied Evans-type P, N ligand (**L55a-b**) for the asymmetric allylic alkylation (Scheme 37). These ligands coupled 1,3-diphenyl-2-propenyl acetate and several nucleophiles under optimal reaction conditions. The reactions were smoothly carried out at -10 °C providing the desired product with up to 87% *ee* and excellent yield using [Pd(η^3 -C₃H₅)Cl]₂ (2 mol %).

It is shown that the length and rigidity of the P-N bond connecting the carbon backbone determine the "bite-angle" of the ligand, which influences the catalytic activity and stereoselectivity of catalysts. In this regard, aminophosphines are considered versatile ligands in which the N centers can impart additional reactivity features. The first amino-phosphine was reported by Michaelis in 1903 [86]. Though amino-phosphines are sensitive to air and moisture, research in aminophosphine chemistry is gaining much attention in catalysis, especially in heterocyclic chemistry [87].

The phosphorus-containing C₂-symmetric chiral amine ligand is highly effective for Pd-catalyzed enantioselective allylic substitution reactions. In these connections, Kubota and Koga [88] reported C₂-symmetric chiral ligands (**L56a-f**) for allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2.5 mol %), KOAc-BSA, in DCM at room temperature (Scheme 38). The ligand **L56a** (2.5 mol%) having chiral binaphthyl rings in the axial orientation to the homo piperidine ring would create a better chiral environment as it obtained high *ee* (up to 93%) than the chiral pyrrolidine having a pseudo-equatorial substituent. In addition, the ligand **L56b** having an expanded ring attained up to 96% enantioselectivity as compared to ligand **L56a**.

Widhalm and coworkers [89] reported various chiral amino-phosphine ligands for allylic alkylation reactions. Though all the ligands **L56h-i** (2 mol %) gave excellent chemical yield, while, only **L56g** exhibited excellent enantioselectivity (96%) for 1,3-diphenyl-2-propenyl acetate in the presence of Pd(OAc)₂ (1 mol %) in THF at room temperature. The ligand **L56j-l** was also used in the allylic substitution of 1,3-substituted propenylacetates with dimethyl malonate [90]. The ligand **L56j** also attained up to 97% *ee* in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (0.5 mol %), KOAc, BSA, in DCM at room temperature.

Extensive work on the DPPBA-SAMP-derived chiral aminophosphine without hydrazone moiety was reported by Mino and coworkers. Their initial report [91] introduced chiral aminophosphine ligands **L57a**, **L58a-c**, and **L59** for asymmetric allylic alkylation reaction (Scheme 39). Though all the ligands (4 mol %) showed good to excellent yields, enantioselectivity was moderate (maximum up to 76% *ee*) for the



Scheme 50. D-Glucosamine diphenylphosphino ligands for asymmetric allylic alkylations.

reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of $[Pd(\eta^3-C_3H_5)Cl]_2$ (2 mol %) using BSA-LiOAc system in toluene at room temperature. Compared to **L58a** (94% yield with 76% *ee*), a slight increase in yield and *ee* was observed for **L58b** (96% with 79% *ee*) in THF. This indicated that a methoxy ethoxy methyl group of the pyrrolidine side chain of ligand **L58b** interacts slightly more with the incoming nucleophile.

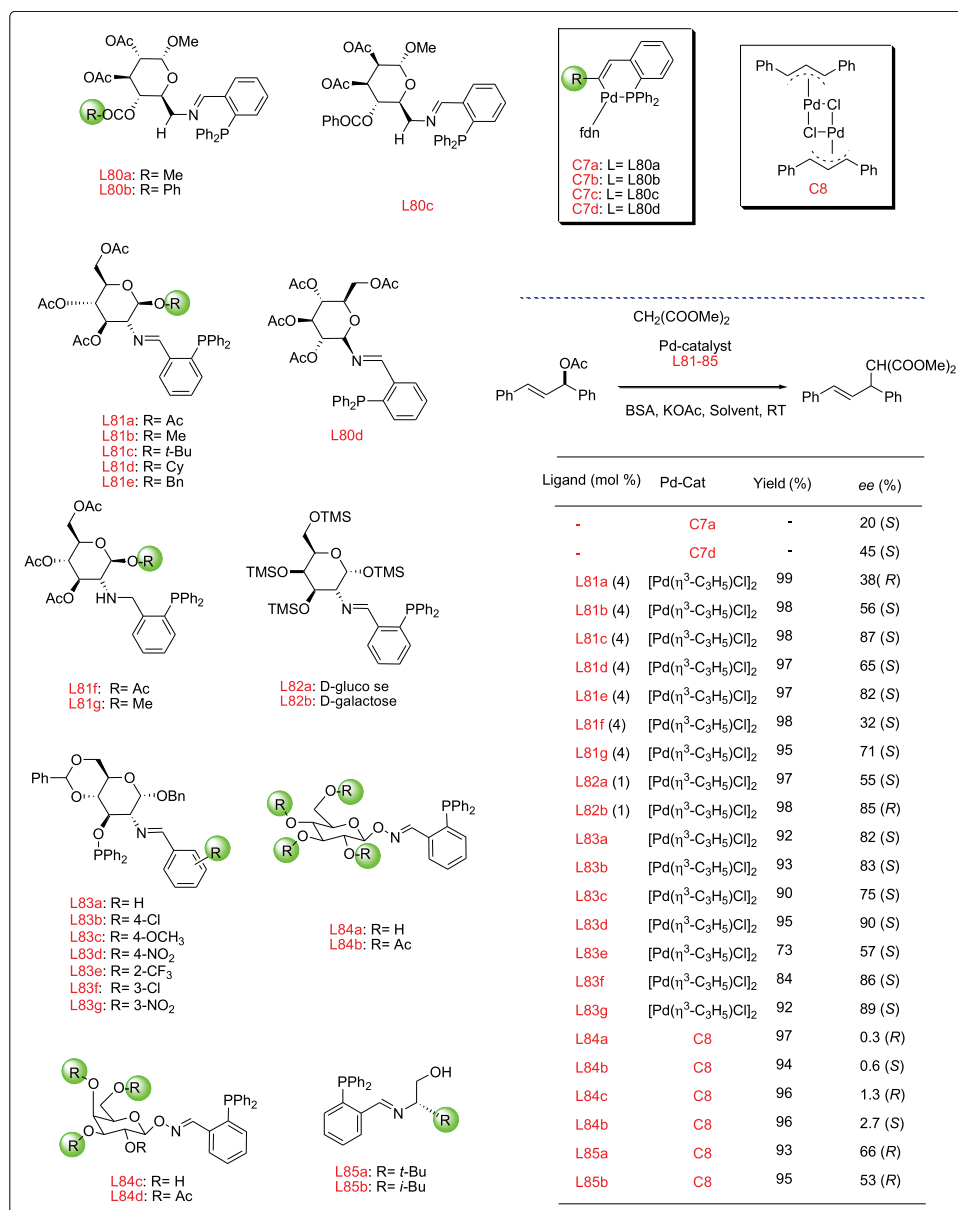
The reactivity and selectivity of pyrrolidinyl-containing amino-phosphines **L60a-d**, **L61a-b**, and ligands **L62a-f** (4 mol %) were studied for asymmetric allylic alkylation [92]. All other ligands, except **L62a** (58% *ee*), brought high enantioselectivity (83–93% *ee*) for the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate by using $[Pd(\eta^3-C_3H_5)Cl]_2$ (2 mol %), BSA-LiOAc in toluene at room temperature. The enantioselectivity was further improved to 96% by decreasing the reaction temperature to $-20^{\circ}C$. Under the same reaction condition, the more sterically hindered silylated chiral aminophosphine ligands (**L60a-d**) worked efficiently. Except for **L60c** (78% *ee*), all the ligands brought high enantioselectivity (91–94%) at room temperature in toluene. The enantioselectivity was further improved to 97% by decreasing the reaction temperature to $-20^{\circ}C$ with only 69%. Under similar reaction conditions, the chiral (*S*)-prolinol-derived amino-phosphine ligands **L63a-c** and ligand **L63d** containing two fluororous ponytails showed excellent yield and enantioselectivities (96% yield and 94% *ee* for **L63d**) for alkylation of 1,3-diphenyl-2-propenyl acetate with dialkyl malonate in diethyl ether at various temperatures [93]. The fluororous Pd catalyst (10 mol %) prepared *in situ* from $[Pd(\eta^3-C_3H_5)Cl]_2$ and **L63d** can be separated easily by simple solid/liquid separation from the reaction mixture and reused up to five times. The catalyst showed consistent four times reusability for the reaction of 1,

3-diphenyl-2-propenyl acetate with dimethyl malonate, while six times reusability for diethyl malonate in diethyl ether at $30^{\circ}C$.

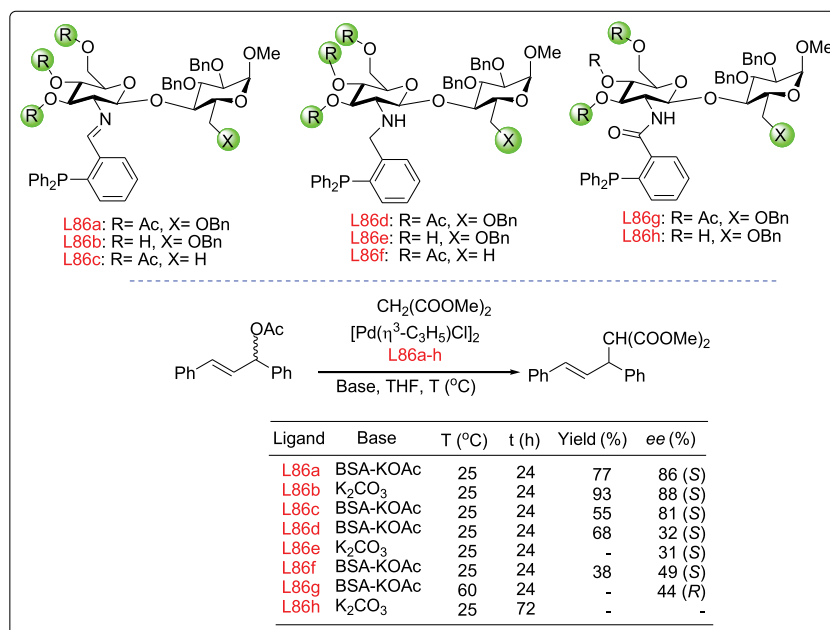
Mino and coworkers synthesized *N*-aryl indoline-type amino-phosphines by the optical resolution of their diastereomeric Pd complexes. The ligands (**L64a-d** and **L65a-d**) (10 mol %) showed excellent enantioselectivity in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of $[Pd(\eta^3-C_3H_5)Cl]_2$ (5 mol %), BSA-LiOAc in toluene at room temperature (Scheme 40) [94]. The ligand **L64d** attained 95% *ee* with a 91% yield of (*R*) product. The ligands (10 mol %) also carried out the kinetic resolution of starting esters by using Pd-catalyzed asymmetric allylic alkylation of a 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of LiOAc-BSA and $[Pd(\eta^3-C_3H_5)Cl]_2$ (5 mol %) in various solvents at $25^{\circ}C$ [95]. Ligand **L64c** was an effective ligand for the kinetic resolution with good enantioselectivities (Scheme 40).

A new *N*-aryl indole-derived chiral phosphine ligands (**L66a-d**) (4 mol %) were also achieved up to 99% *ee* for asymmetric allylic alkylation of 1,3-diphenyl-2-propenyl with dimethyl malonate under standard reaction conditions (Scheme 41) [96]. Though all the ligands were efficient for the alkylation of indoles with 1,3-diphenyl-2-propenyl acetate [97], the ligand **L66c** (5 mol %) gave the desired products in good to excellent yields with up to 90% *ee* for the reaction of substituted indoles in the presence of $[Pd(\eta^3-C_3H_5)Cl]_2$ (2.5 mol %), and K_2CO_3 in toluene at room temperature under argon.

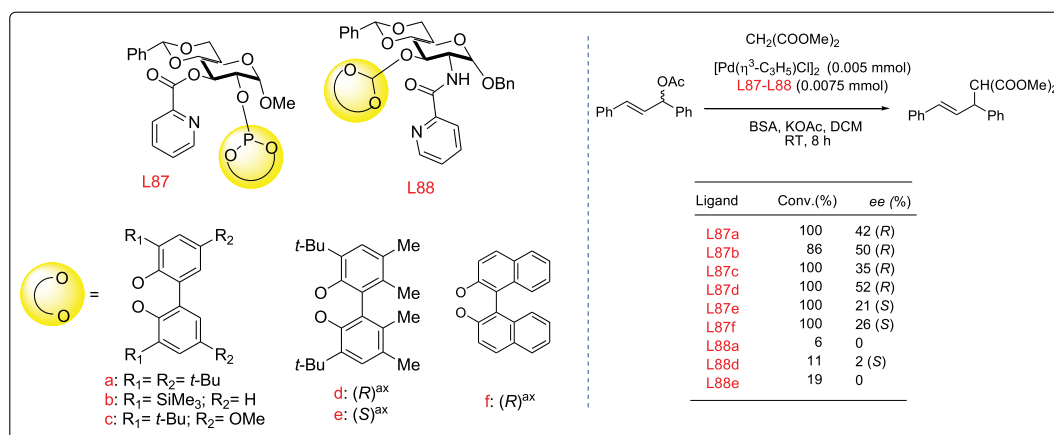
The C(aryl)-N(amine) bond atropisomers of acyclic amino-phosphines based on substituted anilines (**L67a-c**) [98], **L67d-f** [99], and **L68a-c** [100] was developed for asymmetric allylic alkylation reactions (Scheme 42). The chiral acyclic amine **L67a** can be resolved by crystallization without any outside chiral source. While ligands **L67d-f**



Scheme 51. Carbohydrate-based iminophosphinites for asymmetric allylic alkylations.



Scheme 52. Phosphines-based disaccharides for asymmetric allylic alkylations.



Scheme 53. Pyranoside phosphite-pyridine ligands for asymmetric allylic alkylations.

can be resolved by a chiral Pd resolving agent or a chiral HPLC method. The ligand (*aS*)-**L67a** could induce higher enantioselectivity (85% *ee*) than ligand **L67b-c** (up to 77% *ee*) with a 96% yield for the reaction of 1, 3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of BSA and LiOAc in toluene at room temperature. The yield was slightly improved to 98 % with 88% *ee* using MeCN as a solvent at 0 °C. The [Pd(η³-C₃H₅Cl)₂]/**L67d-f** (2:4 mol %) system gave 71–88% yield with 77–83% *ee* for dimethyl malonate in 24 h in DCM at 0 °C. The ligand (*aS*)-**L67d** showed higher enantioselectivity (83% *ee*) than the (*aS*)-**L67e** and (*aS*)-**L67f** in DCM. The enantioselectivity further increased to 90.2% in ether. It was demonstrated that N, N-disubstituted allylic amine type aminophosphines **L68a-c** can also be effective ligands for the allylic alkylation of 1,3-diphenyl-2-propenyl acetate and malonates.

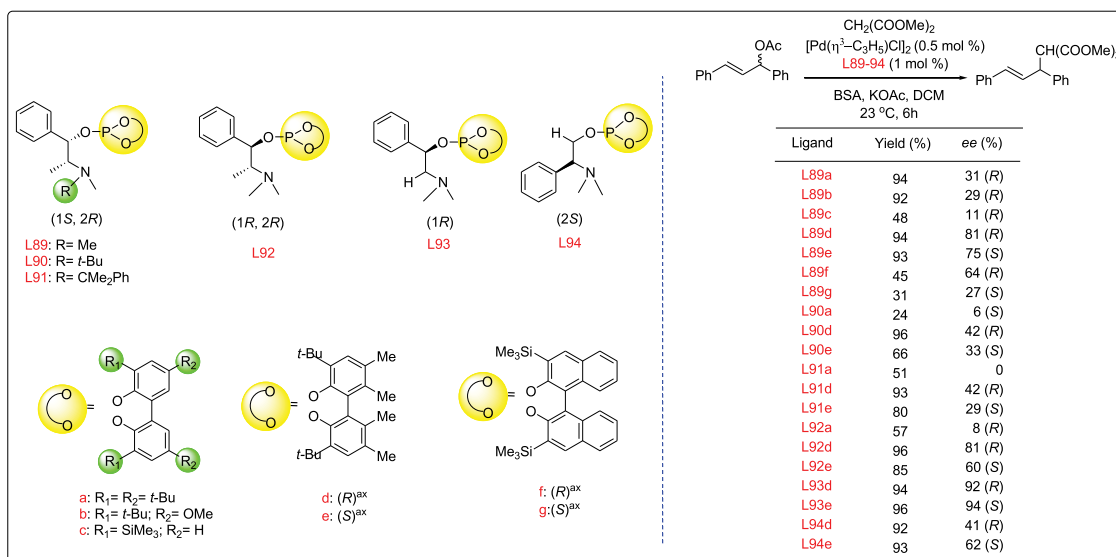
Dai and Virgil [101] achieved asymmetric inductions up to 87% *ee* in the allylic alkylation reactions using a new class of atropisomers monodentate (**L69a**) and its chelate quinazolinone phosphine ligands (**L69b-c**) (Scheme 43).

A homochiral pyridyl, and phosphino derivatives of 2,2-dimethyl-1,3-dioxolane (**L70**) were prepared for allylic alkylation of 1,3-

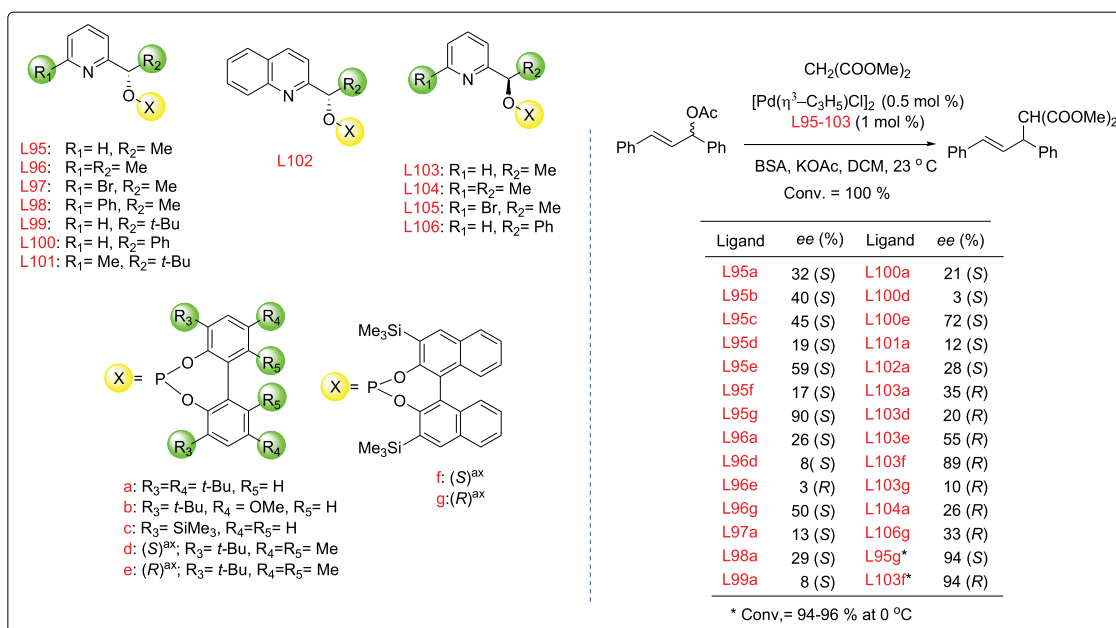
diphenyl-2-propenyl acetate with dimethyl malonate (Scheme 44) [102]. The reaction was carried out using either a preformed **C6** complex obtained from pydiphos [Pd(η³-C₃H₅Cl)₂] with a silver salt or complexes formed *in situ* between precatalyst and ligand **L70**. Though in both cases, quantitative yields were obtained in BSA, and KOAc in DCM enantioselectivity was very low at room temperature.

Trost and co-workers introduced four different ligands (**L71a-d**) for Pd-catalyzed asymmetric reactions (Scheme 45). Initially, they studied the role of the nature of the ion pair in the reaction of 3-acetoxycyclopentene with the Na salt of dimethyl malonate. The nature of tetraalkylammonium salts played an important role in asymmetric induction utilizing ligand **L71a** [103]. The limited solubility of the Na salt of dimethyl malonate leads to only 38% *ee*, which was increased to 86% *ee* using THAB, [Pd(η³-C₃H₅Cl)₂] (2.5 mol %), and **L71a** (7.5 mol %) in THF at 0 °C. As the nature of the ion pair varies with the solvent, the enantioselectivity jumped to 98% in DCM. An excellent result was shown for six and seven-membered ring substrates using a potassium salt of phthalimide under optimized reaction conditions.

Specifically, mono-substituted phosphinous amides (**L72a-c**) (5–20



Scheme 54. Amino phosphite ligands for asymmetric allylic alkylations.



Scheme 55. Phosphite-pyridine ligands for asymmetric allylic alkylations.

mol %) in which the asymmetric P atom is incorporated in a heterocyclic ring were applied for asymmetric allylic alkylations. The ligand **L72a** showed excellent activity (yield 99%) while ligand **L72a** showed 91.5% *ee* in the alkylation in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (2 mol %) using morpholine or DABCO as a base in DCM at room temperature (Scheme 46) [104].

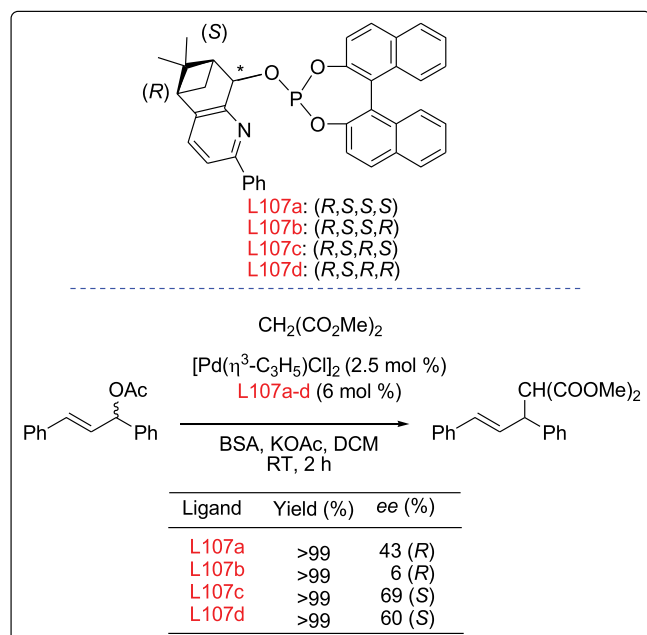
P, N-ligands containing a tetrahydroisoquinoline skeleton (**L73a-b**) act as excellent ligands in asymmetric allylic reactions of 1,3-diphenyl-2-propenyl acetate with several nucleophiles in the presence of [Pd(η^3 -C₃H₅)Cl]₂ [105]. The reaction provided high yields and enantioselectivity. These ligands did not serve as P, N-bidentate but as P-monodentate ligands (Scheme 47).

The camphene-derived aminophosphine ligands (**L74a-c**) also showed excellent enantioselectivity (up to 91%) for allylic alkylation of

1,3-diphenyl-2-propenyl acetate (Scheme 48) [106]. Similarly, Bakos and coworkers [107] reported the pentane-2,4-diyl-based P, N-ligands (**L75a-e**) for asymmetric allylic alkylation reactions. The substituent on N directs the substrate coordination which, influences the reactivity of the catalytically active species. Hence, ligands **L75c-d** provided high enantioselectivity (up to 96%) (Scheme 48)

A series of homochiral amino phosphine ligands (**L76a-f**) showed up to 76% *ee* in allylic alkylation of 1,3-diphenyl-2-propenyl acetate with Na salt of dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ as the pre-catalyst using NaH in THF at room temperature (Scheme 49) [108]. The enantioselectivity was affected mostly by the nature of the substituent on nitrogen.

Framery *et al.* [109] reported various carbohydrate-based amino-phosphine ligands for Pd-catalyzed allylic alkylation reactions (Scheme



Scheme 56. Pyridine-phosphite ligands for asymmetric allylic alkylations.

50). The flexible **L77b** and **L78a-c** ligands [110] gave the desired alkylation product with the (R)-configuration. The ligand **L77b** (4 mol %) gave the lowest enantioselectivity (29%), while ligand **L78b** gave 86% *ee* in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in THF. The ligands **L78a-b** bearing a stereogenic center on the spacer showed similar or lower enantioselectivity than **L77a** (83% *ee*). The ligand **L77a** (4 mol %) showed up to 97% enantioselectivity for the alkylation of 1,3-diphenyl-2-propenyl acetate with various carbon and aminonucleophiles in the presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (2 mol %) and BSA, LiOAc in THF. In this continuation, they [111] studied the influence of the enantioselectivity modifying the configuration and the nature of the substituent at the anomeric position of the phosphine-amide ligands (**L79a-e**). It is necessary to have an acetoxy group at the anomeric center in β -position to obtain enantioselectivity, while the size of the alkyl groups (methyl or benzyl) has not influenced the enantioselectivity.

The P, N-ligands (**L80a-d**) derived from D-glucose and D-mannose on treatment with $[\text{Pd}_3(\text{dba})_2]$ and Fumarodinitrile (a prochiral olefin) in dry toluene gave corresponding $[\text{Pd}(\text{L80a-d})(\text{fumarodinitrile})]\text{Pd}(0)$ complexes (**C7a-d**) [112]. Complex **C7a** showed only 20% enantioselectivity with (S) enantiomer in allylic alkylation of 1,3-diphenyl-2-propenyl acetate and sodium dimethyl malonate in dry THF. Better enantioselectivity (45% *ee*) could be attained by using complex **C7d** (Scheme 51).

Similarly, the monosaccharides imino and amino phosphine ligands (**L81a-g**) (4 mol %) based on alkyl β -D-glucopyranose also exhibited good to excellent activity and enantioselectivities (up to 87% *ee*) in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate using $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (4 mol %) and BSA-KOAc in THF at 25 or 60 °C [113]. Under similar reaction conditions, phosphine-imine ligands derived from D-glucose- and D-galactosamine ligands (**L82a-b**) (1 mol %) achieved a high level of enantioselectivity (up to 85% at 25 °C and 99% at 0 °C) [114].

A series of novel carbohydrate-based iminophosphinite ligands (**L83a-g**) was successfully applied in asymmetric allylic alkylation, and 90% *ee* was achieved by *p*-nitro-substituted ligand **L83d** at 25 °C [115].

A hydrolytically stable oxime (**L84a-d**) based on O- β -D-glucopyranosyl was applied for the Pd-catalyzed allylic alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate (Scheme 51)

[116]. Though all ligands exhibited good catalytic activity, the enantioselectivity was very poor under standard reaction conditions in DCM at room temperature. However, the ligands **L85a** and **L85b** induced significant stereoselectivity.

It was observed that the disaccharide aminophosphines (**L86a-h**) proved to be superior to the corresponding monosaccharide phosphine-amides in terms of enantiomeric excesses in the reaction of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in THF using $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (2 mol %). The ligand **L86b** (4 mol %) containing free hydroxyl groups on the D-glucosamine framework resulted in high reactivity (93% yield) and enantioselectivity (88% *ee*) (Scheme 52) [117].

Phosphite ligands are extremely attractive and widely used for transition metal catalysis. They are easy to prepare from readily available alcohol [118]. Also, they are less sensitive to air and other oxidizing agents than phosphines.

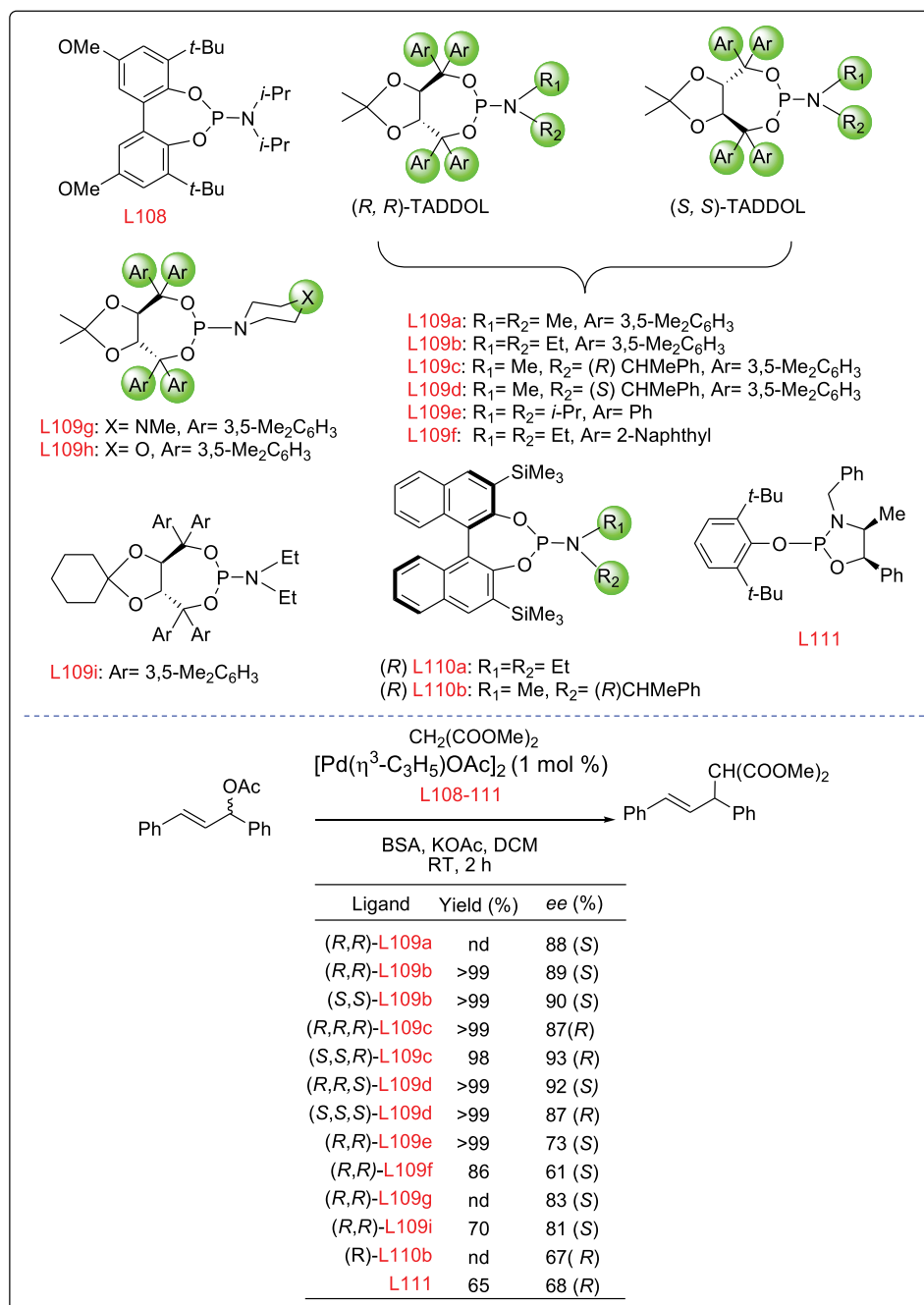
The pyranoside phosphite-pyridine ligands were developed for enantioselective allylic substitution of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate in the presence of the $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (Scheme 53) [119]. Two libraries of ligands were developed by the immediate functionalization of D-glucose and D-glucosamine. The ligands involve systematic variations of the positions of the phosphite group at both C-2 (**L87**) and C-3 (**L88**) positions of the pyranoside backbone. The ligands also consist of different substituents and configurations in the biaryl phosphite moiety which maximize the catalyst performance. It was indicated that enantioselectivity depends on the position of the phosphite moiety at C-2 or C-3 of the pyranoside backbone as well as the substituents and configuration of the biaryl phosphite moiety. The ligand **L88** with the phosphite group attached to the C-3 of the pyranoside backbone showed lower activity and enantioselectivity than ligand **L87**.

Similarly, a library of amino phosphite ligands (**L89-94**) obtained from enantiopure amino alcohols was applied for asymmetric allylic reactions (Scheme 54) [120]. Various factors, like substituents (**L89**, **L91**, and **L92**), configuration (**L89** and **L90**) at the ligand backbone, the amine substituents (**L89**), the substituents, and configuration of the biaryl phosphite moiety (**L92a-g**) affects on the activity of the ligands. The presence of trimethylsilyl groups at the ortho positions of the biaryl phosphite moiety negatively affects both the activity and the enantioselectivity.

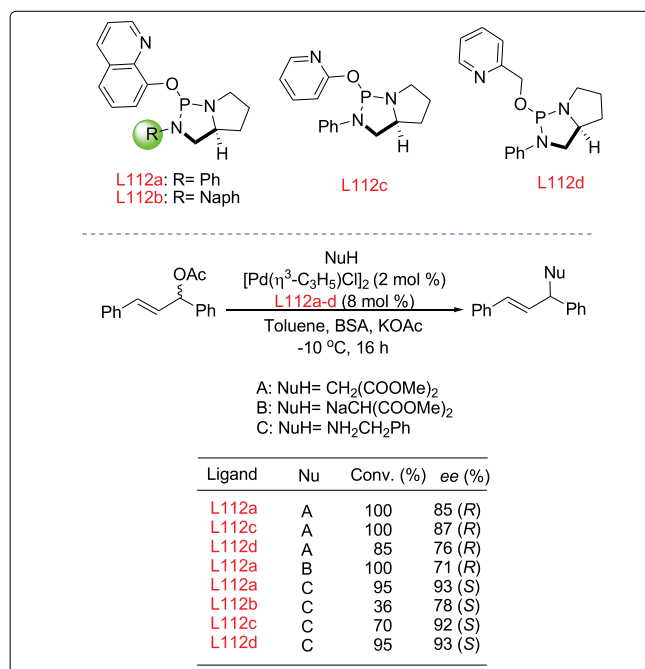
The best enantioselectivity was obtained for the ligands **L92d** and **L93e** containing enantiopure biaryl phosphite moiety containing the *t*-butyl group at the *ortho* positions. The effect of the amine substituents on **L89a-c** indicated that ligand **L89a** containing dimethylamine group showed better enantioselectivity than the **L89b-c**. The effect of the configuration of the ephedrine backbone was studied by comparing **L89a** and **L90**. Evaluation of **L93** and **L94** showed that the ligands, which do not have the methyl substituent at the stereogenic C-2 position of the ephedrine backbone provided the highest enantioselectivities (92% *ee* and 94% *ee* for **L93d** and **L93e**), while ligand **L94e** leads to similar enantioselectivity as that of **L92e**.

Similarly, a library of phosphite-pyridine ligands **L95-L106** synthesized from the corresponding easily accessible racemic hydroxyl-pyridine compounds were also applied for the allylic substitution reactions of various di- and trisubstituted substrates with C, N, and O nucleophiles (Scheme 55) [121]. High activity and enantioselectivity (up to 94%) were obtained for both **L95g** and **L103f** ligands in the reaction of 1,3-diphenyl-2-propenyl acetate in presence of $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (0.5 mol %) in DCM at 23 °C. The effect of the substituents at the ligand backbone (R_1 and R_2) showed that for **L95-L102a**, enantioselectivity decreases with steric hindrance at both positions. The best enantioselectivity was observed for **L95** containing hydrogen and a methyl substituent at R_1 and R_2 positions with the formation of (S) product, predominantly.

Meng et al. [122] synthesized a series of pyridine-phosphite ligands (**L107a-d**) from PCl_3 , chiral BINOL, and 4H-quinolyl alcohols and applied them to asymmetric allylic substitution reactions (Scheme 56).



Scheme 57. TADDOL-derived phosphoramidite for asymmetric allylic alkylations.



Scheme 58. C_2 -symmetric and non-symmetric chiral ligands for asymmetric allylic alkylations.

All the ligands (6 mol %) showed excellent yield (> 99%) with moderate enantioselectivity for the reaction of 1,3-diphenyl-2-propenyl acetate and dimethyl malonate in the presence of $[Pd(\eta^3-C_3H_5)Cl]_2$ (2.5 mol %), BSA, KOAc in DCM at room temperature. The ligands **L107a** and **L107b** gave a product with (R) configuration while **L107c** and **L107d** gave products with (S) configuration. Under optimized reaction conditions, moderate enantioselectivity was observed by the ligand **L107c** in the asymmetric allylic aminations with *p*-anisidine.

A series of bulky monodentate phosphoramidite ligands (**L108-L111**) based on TADDOL backbones was employed in the allylic alkylations of both symmetric and nonsymmetric substrates (Scheme 57) [123]. Thus, (*R,R*)-**L109b** showed 89% *ee*, with (S) configuration for the alkylation of 1,3-diphenyl-2-propenyl acetate with dimethyl malonate using $[Pd(\eta^3-C_3H_5)(OAc)]_2$ (1 mol %), BSA, KOAc at room temperature. It was observed that changing the ligand stereochemistry in the TADDOL backbone (*S,S*)-**L109b** resulted in the same enantioselectivity, with the formation of an opposite enantiomer (R). Ligands (*S,S,R*)-**L109c** and (*R,R,S*)-**L109d** also showed excellent enantioselectivity.

In most of the C_2 -symmetric and non-symmetric chiral ligands, the phosphorus atom is not stereogenic, while chirality is mainly induced by a chiral N atom. In contrast to this, Buono and coworkers introduced quinoline and pyridine-phosphines (**L112a-d**) as new nonsymmetric chiral ligands bearing the chiral phosphorus atom for Pd-catalyzed asymmetric alkylation reactions (Scheme 58). The ligands were synthesized from tris(dimethylamino)phosphine and (*S*)-(+)-2-anilinoethylpyrrolidine and pyridine. The ligand **L112c** (8 mol %) achieved up to 87% *ee* in the allylic substitution of 1,3-diphenyl-2-propenyl acetate in the presence of $[Pd(\eta^3-C_3H_5)Cl]_2$ (2 mol %) and BSA-KOAc in toluene at $-10^\circ C$ [124]. The ligand also attained up to 94% *ee* in the allylic amination of 1,3-diphenyl-2-propenyl acetate or 1,3-diphenyl-2-propenyl carbonate with primary or secondary amines such as benzylamine, veratrylamine, or morpholine [125].

Bravo and coworkers [126] reported a library of bis(diamidophosphite) ligands (**L113-122**) for asymmetric allylic substitution reactions (Scheme 59). Enantiopure bis(diamidophosphite) ligands (**L113-L115**) with a heterocyclic terminal fragment were derived from (*R*)- and (*S*)-*N,N'*-dimethyl-1,1'-binaphthyldiamine. The bridging fragments were derived from (*S,S*)-2,3-butanediol, (*4R,5R*)-4,5-di-

(hydroxymethyl)-2,2-dimethyl-1,3-dioxolane, and (*R*)- and (*S*)-1,1'-bi-2-naphthol. These ligands create a well-defined chiral pocket as analyzed by NMR spectra. The cationic allylic complexes **C9-C12** of general formula $[Pd(\eta^3-2-CH_3-C_3H_4)(P-P)](PF_6)$ were obtained by reaction of the organometallic precursor $[Pd(\eta^3-2-CH_3-C_3H_4)(\mu-Cl)]_2$ with the stoichiometric amount of the appropriate ligands at low temperature in the presence of $NaPF_6$.

The effect of the nature and absolute configuration of both the terminal and bridging fragments of the ligands were studied for the asymmetric allylic alkylation and amination reactions. All the catalysts were tested for reactions of 1,3-diphenyl-2-propenyl acetate with Na dimethyl malonate or benzylamine as a nucleophile in DCM at $25^\circ C$. The catalyst **C11a** derived from ligand **L115a** was best in terms of enantioselectivity (*ee* up to 85%) in both alkylation and amination reactions in DCM at $25^\circ C$. In addition, a decrease in the enantioselectivity was observed for the "in situ" generated catalysts by mixing $[Pd(\eta^3-C_3H_5)Cl]_2$ and the appropriate amount of the corresponding ligand. All the complexes were also applied for the allylic amination of 1,3-diphenyl-2-propenyl acetate by benzylamine. The ligand **L115a** was best in terms of enantioselectivity (*ee* up to 89%) and conversion (100%) with the formation of (S) product.

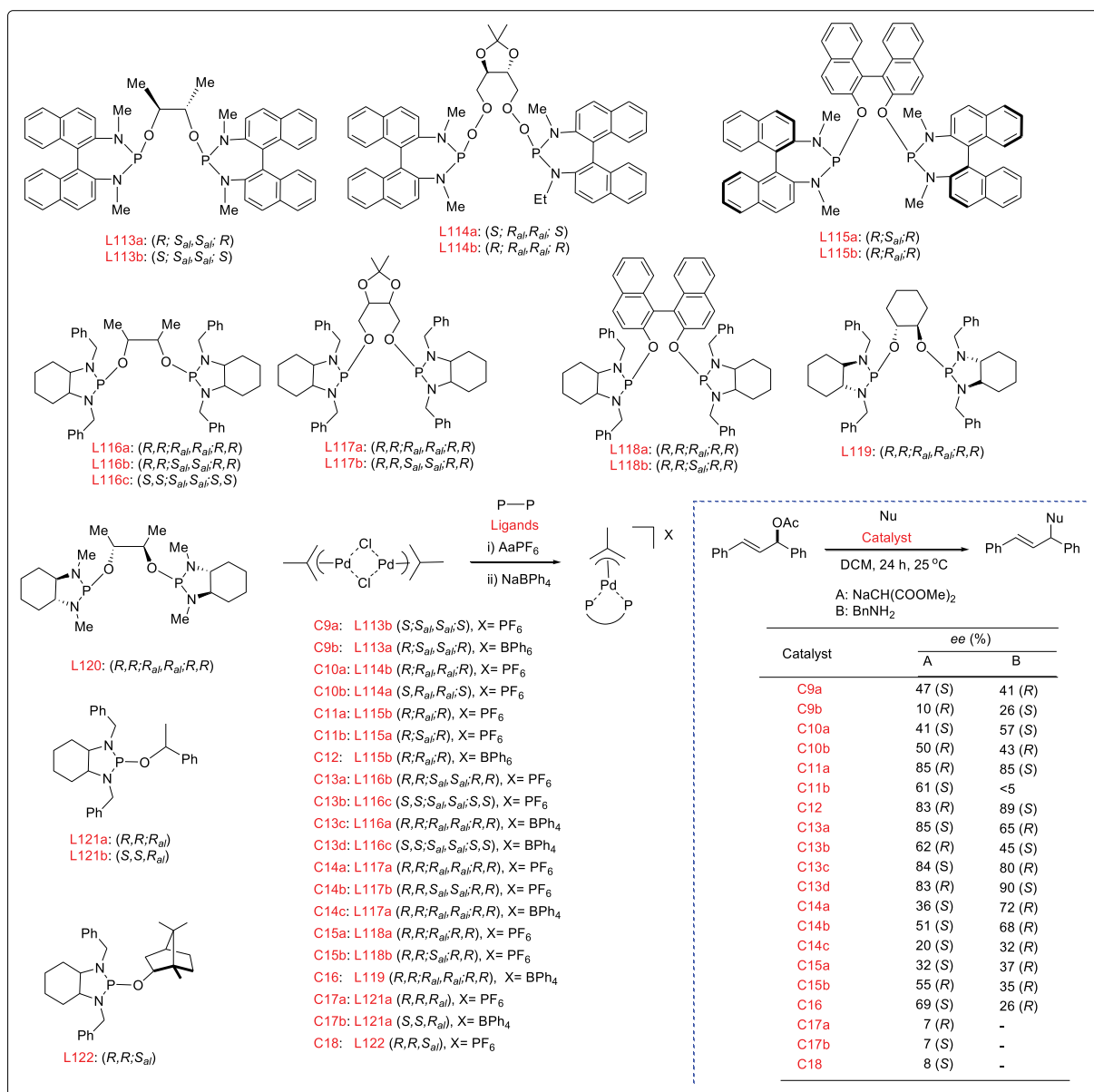
As an extension, they reported enantiopure bis(diamidophosphite) ligands (**L116-122**) bearing a cyclohexane-1,2-diamine skeleton in the allylic substitution reactions [127]. The cationic allyl Pd complexes **C13-18** obtained from the $[PdCl_2(COD)]$ and corresponding ligands were also tested for allylic alkylations. The catalyst **C13c** showed 100% conversion with good *ee* for alkylation (84%) and amination (80%) reactions.

Extensive work on a recyclable amphiphilic resin-supported P, N chelating Pd complexes was carried out by Uozumi and coworkers. They prepared novel P, N-chelate chiral ligands having pyrrolo[1,2-*c*]imidazolone skeleton as a basic chiral unit immobilized on the PS-PEG resin. All the ligands (**L123a-c** and **L124**) and catalysts (**C19a-c**) were applied for different coupling reactions under mild reaction conditions in an aqueous medium. The catalyst **C19a** achieved up to 90% *ee* in the asymmetric allylic substitutions of cyclic substrates in the presence of Li_2CO_3 in water at $40^\circ C$ [128]. The catalyst was recovered by simple filtration and reused without any loss of activity and stereoselectivity (Scheme 60).

The **C19a** also showed excellent yield and enantioselectivity (90–98% *ee*) for asymmetric allylic amination of cycloalkenyl carbonates and gave the corresponding cycloalkenylamines [129]. The recycling experiments showed that the catalyst **C19a** exhibited good catalytic activity and stereoselectivity for at least three runs in the amination reaction of cycloheptenyl ester with dibenzyl amine. Furthermore, the homogeneous catalytic system using **L124**, which lacks PS-PEG supports, gave only 6% of the desired product in DCM. The catalytic asymmetric etherification of cycloalkenyl esters with phenolic nucleophiles was also carried out using **C19a** (2 mol %). The reaction gave optically active aryl ethers (cycloalkenyl) with up to 94% *ee* in water [130].

Pd-catalyzed allylic substitutions of monosubstituted allylic compounds such as cinnamyl acetates and chlorides are very interesting reactions because these substrates are readily accessible. In addition, the obtained branched products can be transformed into different useful chiral intermediates. In this regard, various P–N ligands were used for the allylic substitutions of cinnamyl acetates and chlorides.

It was also known that strongly π -accepting phosphine ligands significantly reduce the electron density at the transition metal center, which makes it more reactive. It also imparts the Lewis acidic character to the catalyst, which can be advantageous for asymmetric catalysis. Due to the strong π -accepting properties of the perfluorinated alkyl phosphines [131], Shen et al. [132] reported a range of new (perfluoroalkyl) phosphinooxazoline (FOX) ligands (**L125a-h** and **L126a-h**) containing the CF_3 and C_2F_6 groups for asymmetric catalysis (Scheme 61). The CF_3 group on phosphorus makes the ligand a much stronger π -acceptor. It



Scheme 59. Bis(diamidophosphite) ligands for asymmetric allylic alkylations.

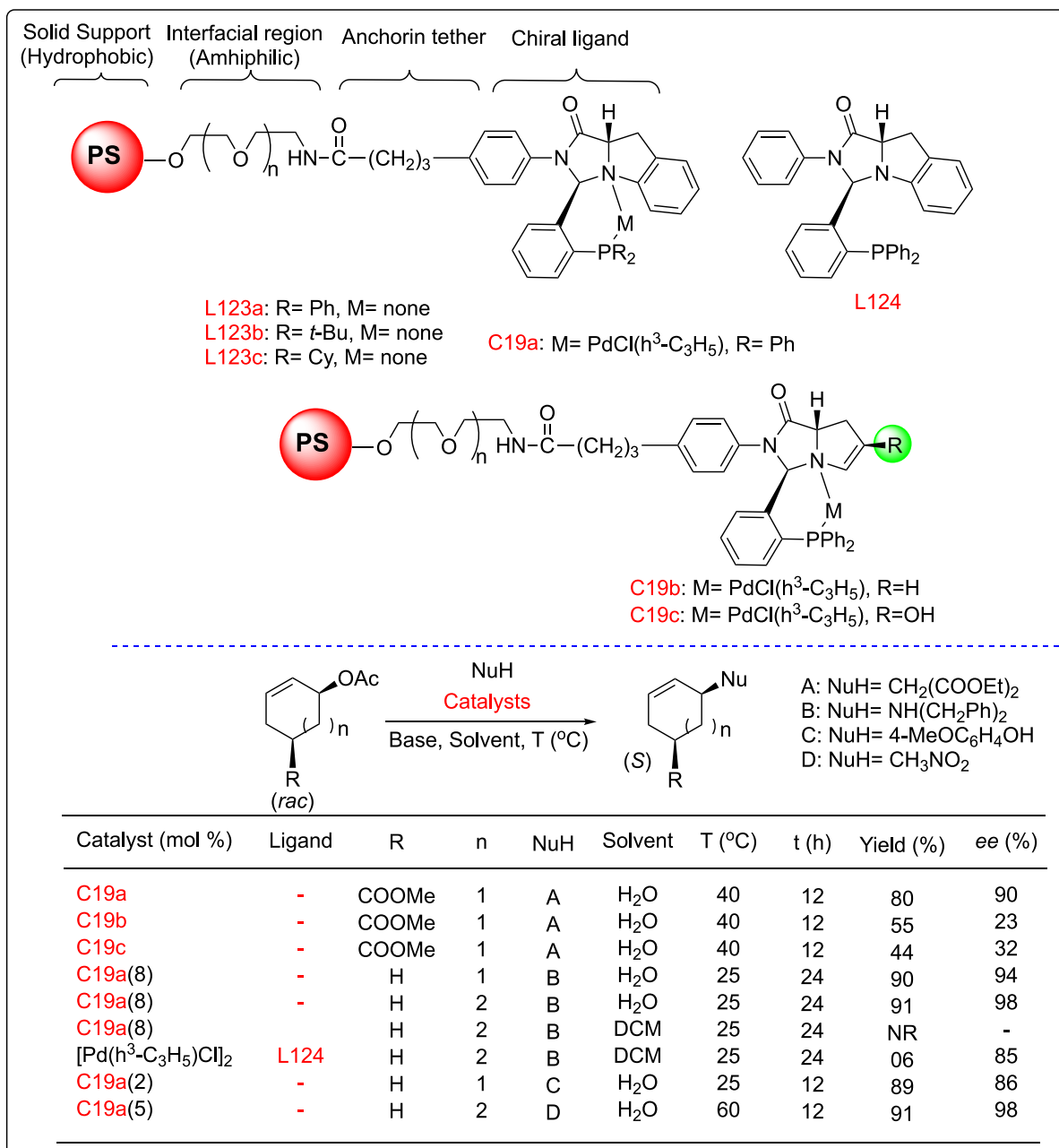
enables the introduction of asymmetry in monosubstituted allyl substrates with enhanced regio and enantioselectivity.

The ligand **L125d** (4 mol %) and Pd(dba)₂ (4 mol %) showed full conversion of (*E*)-cinnamyl acetate and dimethyl malonate with 11.5:1 regioselectivity and 94% *ee* using BSA, LiOAc in DCM at room temperature. When **L125h** was applied, the enantioselectivity improved to 97%, with slightly lowered regioselectivity (9:1). A series of mono-substituted cinnamyl acetates or carbonates were fully converted into the desired product in excellent regio, and enantioselectivity using **L125d** or **L125h** within 8 h. The high regio and enantioselectivity of the Pd-Fox system were supported by the solid structure of the [Pd(**L125d**)(η^3 -cinnamyl)]SbF₆ intermediate.

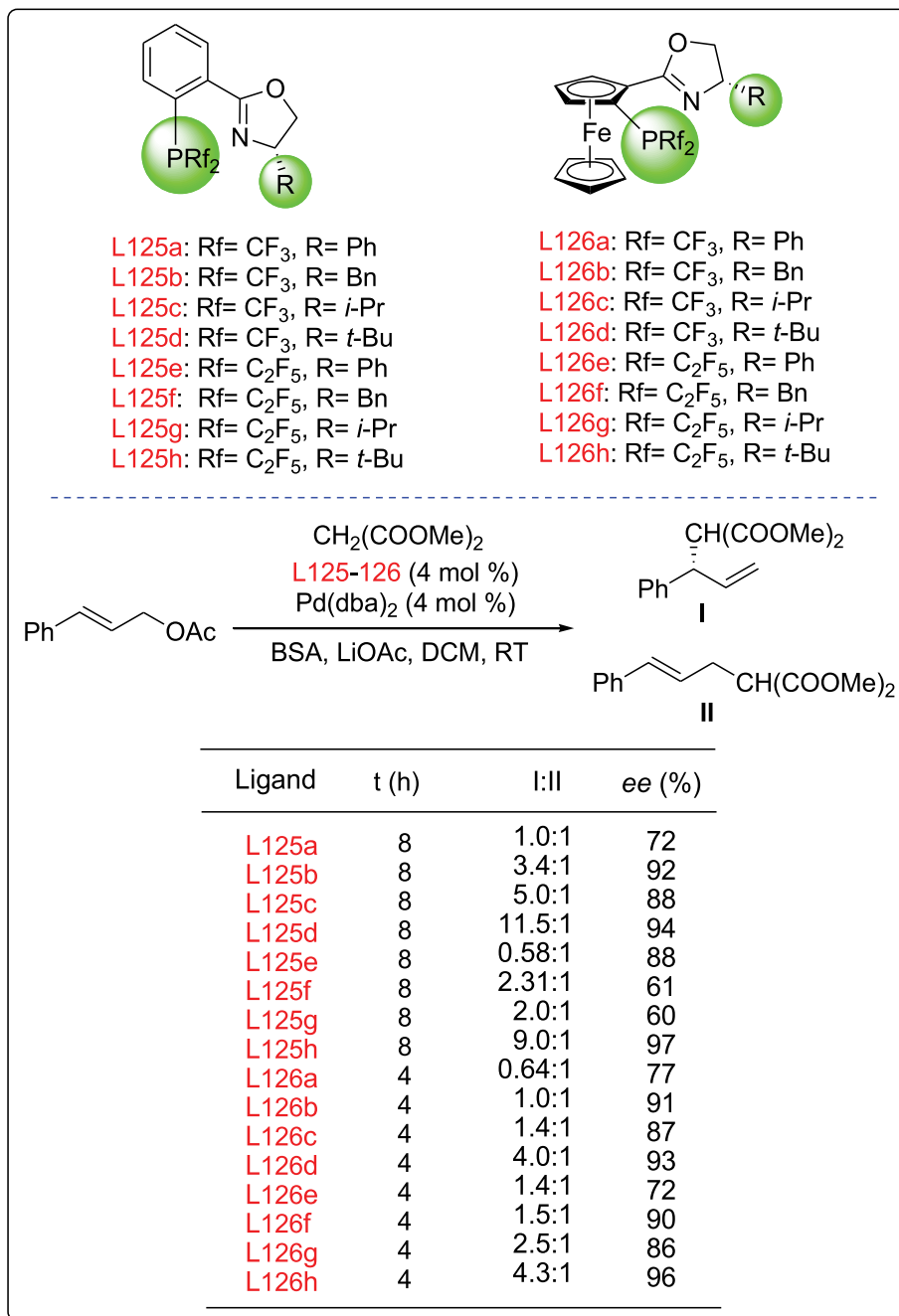
Pfaltz and co-workers showed that replacing phenyl substituent of PHOX (**L1e**) by electron-withdrawing pentafluorophenyl groups (C₆F₅) group (**L127a**) shifts the regioselectivity from 4:96 to 47:53 for enantioselective allylic alkylation of 3-phenyl-2-propenyl acetate by dimethyl malonate in the presence of [Pd(η^3 -C₃H₅)Cl]₂ (1 mol %), and BSA-KOAc in DCM at 23 °C (Scheme 62). Consequently, replacing the

phenyl substituent of PHOX (**6**) with the biphenylphosphite group (**L127b**: 2.5 mol %) further shifts the regioselectivity to 63:37 [133]. The introduction of a second stereogenic unit derived from binaphthol (**L128a**) further improved the regioselectivity to 76:24 and enantioselectivity to 90% *ee*. Furthermore, the introduction of two ortho-methyl groups in the binaphthol (**L128d**) improved the enantioselectivity to 91% with decreased regioselectivity (39:61). On the other hand, a comparison of results of **L128** with **L129** indicated that the enantioselectivity is determined largely by the chiral dihydro oxazole ring, while the binaphthol unit has a minor, but the still significant effect as **L129** showed 92% enantioselectivity with decreased regioselectivity (55:45). Of various substrates used, 1-naphthyl-substituted allylic acetates showed the best regioselectivity (95:5) with excellent enantioselectivity (94%) for ligand **L128a**.

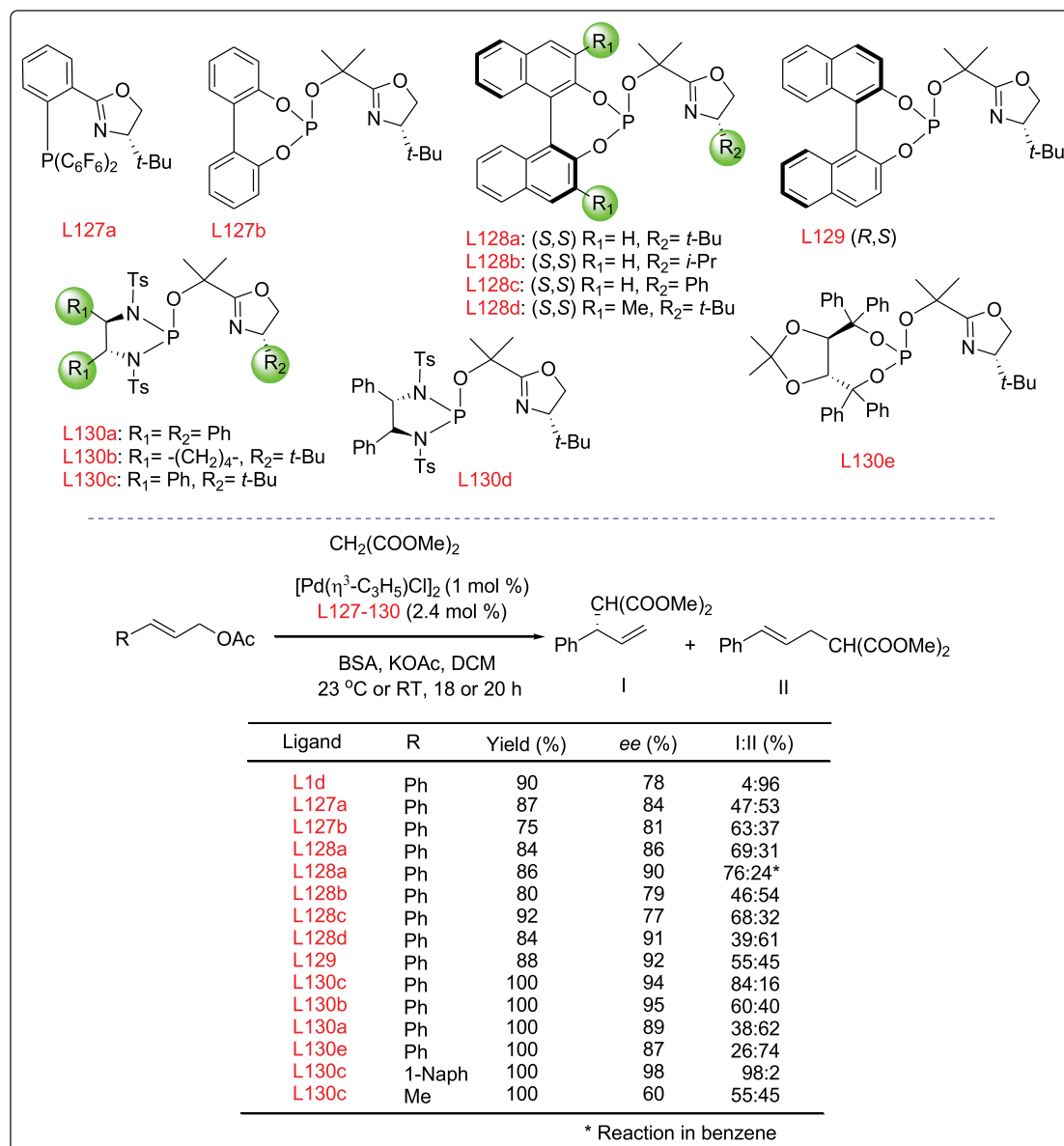
As promising results were shown by **L128** and **L129**, the chiral bis(N-tosylamino)phosphines (**L130a-d**) and TADDOL-phosphite oxazoline ligands (**L130e**) were acts as an efficient ligands for allylic alkylation of various substrates (Scheme 62) [134]. The [Pd(η^3 -C₃H₅)Cl]₂/**L130a**



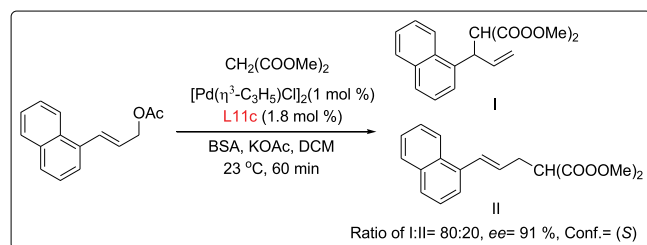
Scheme 60. Recyclable amphiphilic resin-supported P, N-chelating Pd complexes for coupling reactions.



Scheme 61. Bis(perfluoroalkyl) phosphine-oxazolines for asymmetric allylic alkylations.



Scheme 62. Biphénylphosphites and TADDOL-phosphite oxazolines for asymmetric allylic alkylations.



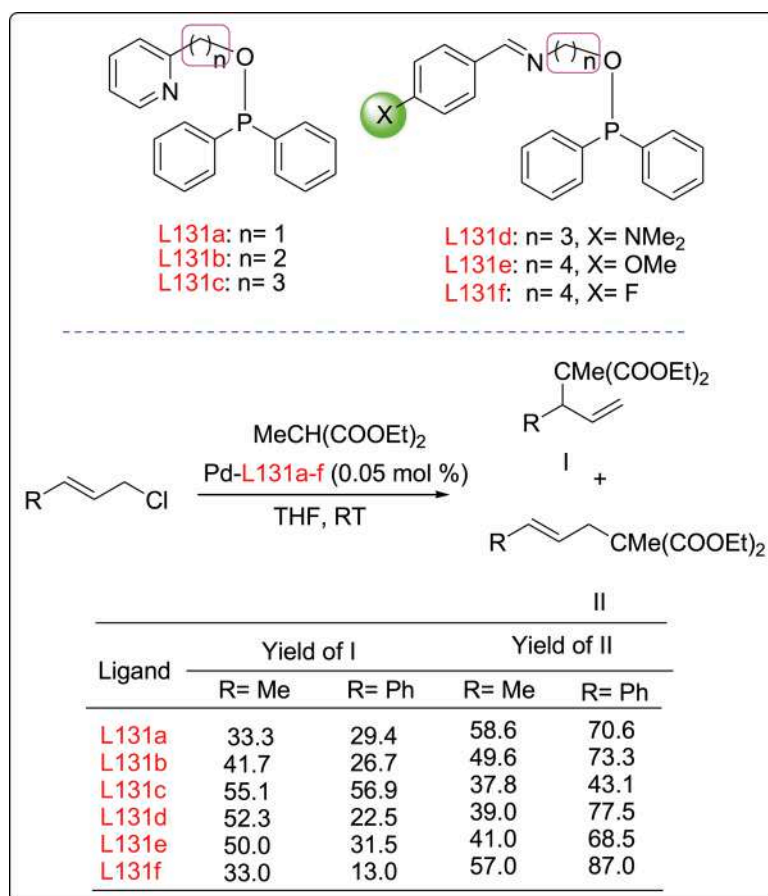
Scheme 63. Carbo-PHOX for asymmetric allylic alkylations.

(1:2.5 mol %) was very effective in reactions of 3-aryl-2-propenyl acetates as well as for 1-naphthyl-2-propenyl acetates in the presence of BSA, KOAc in DCM at 23 °C. Moreover, ligand **L130a** also proved to be superior for cyclohexenyl acetate under similar reaction conditions.

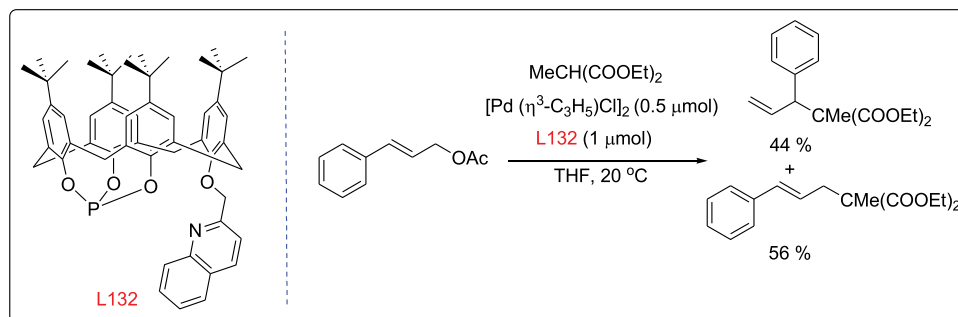
The ligands **L11c** (1.8 mol %) produced the desired branched isomer **I** as the major product with 88 % enantioselectivity and 80:20

regioselectivity in the presence of [Pd(η³-C₃H₅)Cl]₂ (1 mol %) and BSA-KOAc in DCM at 23 °C. [40b] (**Scheme 63**).

Leeuwen and coworkers [135] studied the influence of the relative donor-acceptor strength of N and P atoms, steric hindrance in the transition state, and the bite angle of the ligand on the regioselective formation of branched products (**Scheme 64**). For this study, they synthesized two sets of ligands **L131a-c** and **L131d-f** based on a mixture of P and N donor atoms. The first set consists of a Ph₂PO unit connected ortho to pyridine *via* different alkyl chain lengths, while the second set consists of Ph₂PO units of varying chain lengths linked to an imine moiety. The reaction of the appropriate ligand with the [(C₄H₇ or C₉H₅)PdCl]₂ dimers, followed by chloride abstraction with silver triflate, gives the cationic crotyl and cinnamyl palladium complexes. The stoichiometric and catalytic allylation reactions of crotyl chloride (*trans*-but-2-enyl chloride) and cinnamyl chlorides (*trans*-3-phenylprop-2-enyl chloride) with Na salt of diethyl-2-methyl malonate showed that ligands with larger bite angles lead to higher regioselectivity in THF at room temperature. Stoichiometric alkylation using [Pd(C₄H₇){**L131e**}] [O₃SCF₃] proceeds with 88% regioselectivity to the branched product.



Scheme 64. P–N ligands for asymmetric allylic alkylations of crotyl and cinnamyl chlorides.



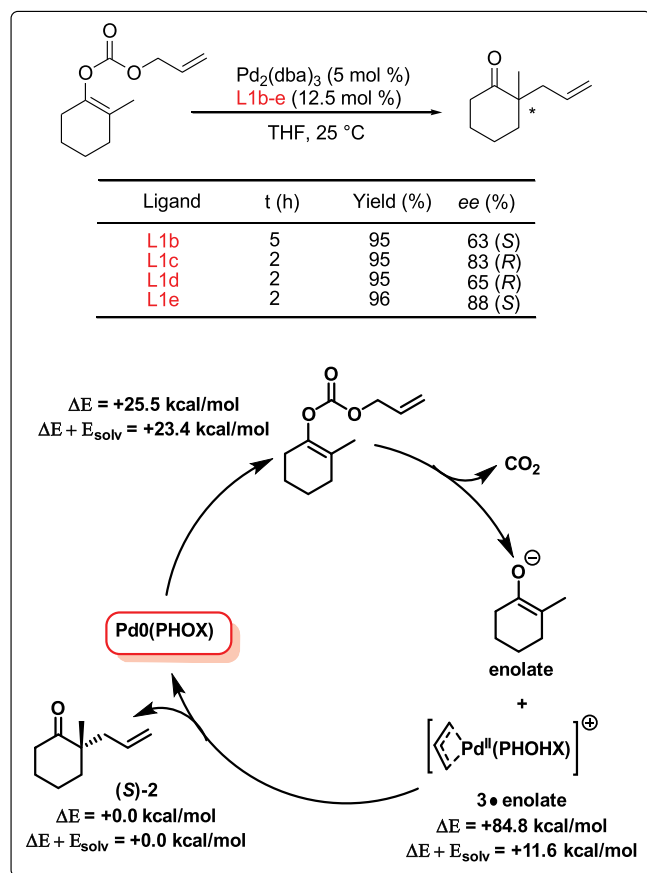
Scheme 65. Calix[4],quinoline-based P, N-ligand for asymmetric allylic alkylations of cinnamyl acetate.

Similarly, the calix[4],quinoline-based P, N-ligand **L132** was applied to Pd-catalyzed allylic alkylation of cinnamyl acetate with sodium diethyl 2-methylmalonate (**Scheme 65**) [136]. The ligand involves quinoline-containing moiety at the lower rim of the cavity. The calix [4]-cavity adopts a cone confirmation with an *exo* orientation of the phosphorus lone pair enabling chelation. The character of the P has been increased by using a μ^3 -bridging phosphite unit linked to three proximal phenolic oxygen atoms at the lower rim of a calix[4]arene cavity that displays the unique air-robustness and stability. The ligand **L132** was obtained from *t*-butylcalix[4]arene through a two-step synthesis. The catalysts were prepared *in situ* by mixing $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ with two equivalents of **L132** in THF solution. The catalyst (0.1 mol %) showed

quantitative conversion in 20 h providing only the 56% linear *trans* and the 44% branched products, while no linear *cis* product was observed at 20 °C.

Tsuji allylation reaction [137] is an ideal reaction for the preparation of asymmetric α -quaternary cycloalkanones. The reaction involves the selective formation of the enol carbonate moiety and provides position control to the enolate.

In this regard, Stoltz and coworkers [138] reported the use of PHOX ligands (**L1b–e**) for an enantioselective Tsuji allylation of allyl enol carbonates (**Scheme 66**). Using these reactions, they prepared various 2-alkyl-2-allyl cyclohexanone derivatives. The ligands **L1b–e** (12.5 mol %) were effective in terms of yield and enantioselectivity in the presence



Scheme 66. Enantioselective Tsuji allylation reactions using PHOX ligands.

of $\text{Pd}_2(\text{dba})_3$ (5 mol %) in THF at 25 °C. Though all the ligands showed a high degree of enantioselectivity and reactivity, the ligand **L1e** was more active in terms of yield (96%) and *ee* (88%). Various allyl enol carbonates of substituted cyclic ketones were converted easily into their corresponding products with 81–96% yields and 79–92% *ee* in the presence of $\text{Pd}_2(\text{dba})_3/\text{L1e}$ (2.5:6.25 mol %) system in 1–9 h. The different tetra-substituted trimethylsilyl enol ethers also showed smooth α -allylation by commercially available diallyl carbonates using the same catalytic system in the presence of a sub-stoichiometric amount of $\text{Bu}_4\text{NPh}_3\text{SiF}_2$ (TBAT). The yield of the reactions was 79–99% with 79–92% *ee* at 25 °C. Using quantum chemistry calculations, they suggested the formation of a likely intermediate in the reaction and there involvement in the inner-sphere process (Scheme 66) [139].

They also reported a practical method for the deracemization of quaternary carbon stereocenters using a $\text{Pd}_2(\text{dba})_3/\text{L1e}$ catalytic system (2.5:6.25 mol %) [140]. The method involves enantioconvergent decarboxylative allylation of β -substituted 2-carboxy allyl cyclohexanones and β -ketoesters. The catalyst was well involved in stereoablative (C–C bond breaking) and stereoselective (C–C bond-forming) reactions. The ligand also showed 80–99% yields with 81–91% *ee* in the decarboxylative allylations of various α -substituted 2-carboxy allyl cyclohexanones in THF or Et_2O at 25–30 °C (Scheme 67). Moreover, the various substituted cyclic quaternary β -ketoesters were converted into the corresponding products in high yields (77–97%) and enantioselectivities (85–92% *ee*) under similar reaction conditions. The ligand

also afforded cascade product by double allylation of a corresponding substrate. It involves the conversion of reactive enol carbonate and a β -ketoester moiety into a mixture of C_2 /Meso diastereomers (4:1) with a 76% yield. The major (–) diastereoisomer was obtained with 92% *ee*.

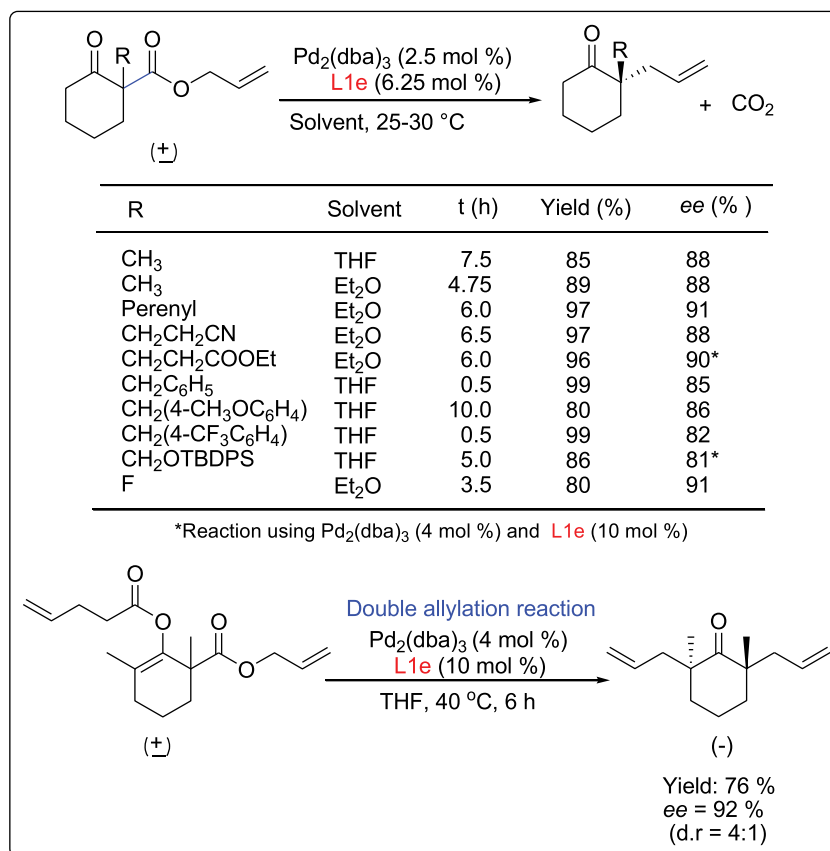
As an extension, they [141] synthesized cyclopentanoid, cyclohexanoid, and cycloheptanoid cores and used them in the total synthesis of natural products. All these compounds were easily prepared by enantioselective decarboxylative allylation of seven-membered methyl-substituted β -ketoesters using **L1e**, **L133a**, and **L133b** ligands. The catalytic system **L1e**/ $\text{Pd}_2(\text{pmdba})_3$ (2.5:6.25 mol %) gave vinylogous esters with 94% yield and 84% *ee* in THF at 30 °C (Scheme 68). As compared to other ligands, the electron-deficient **L133a** ligand (12.5 mol %) displayed improved enantioselectivity (90%) at the cost of higher $\text{Pd}_2(\text{pmdba})_3$ loading (5 mol % Pd) and lower yield (57%). Similarly, structurally modified **L133b** ligand proved to be less effective with reduced yield (77%) and *ee* (72%). The **L1e**/ $\text{Pd}_2(\text{pmdba})_3$ system (2.5:6.25 mol %) well alkylated several aromatic and heteroaromatic α -substituted β -ketoesters with 75–99% yield and 58–92% *ee* in toluene at 30 °C.

In 2012, Paquin et al. [142] applied various PHOX ligands (**L134a–j**) for three types of Pd-catalyzed enantioselective transformations. The first transformation is an enantioselective Tsuji–Trost allylation reaction of fluorinated silyl enol ether using $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (1.25 mol %), and **L134a–g** (3.1 mol %) in the presence of TBAB (35 mol %) in toluene at 40 °C (Scheme 69). The valine-based ligands (**L134a–d**) showed good performance (87–93% yields with 95:5 *er*) as compared to ligand **L1e** (91% yield with 96:4% *er*). The ligands derived from leucine (**L134e** and **L134f**) and isoleucine (**L134g**) gave 89–92% yields with low enantioselectivity (94.5:5.5 *er*). They also carried out the allylation reaction of fluorinated enol carbonates in the presence of $\text{Pd}_2(\text{dba})_3$ (2.5 mol %) and **L134a–g** (1.25 mol %) in toluene at 40 °C. Except for ligand **L134b** (79% yield with 93.5:6.5 *er*), the ligands derived from valine (**L134a**, **L134c**, and **L134d**) gave fluoroketone with 83–93% yields and the same enantioselectivity (95:5 *er*). In terms of product yields, the ligand derived from leucine (**L134e**) performed well (73%) but gave lower enantioselectivities (93:7 *er*). While isoleucine-based ligand **L134g** showed comparable results (80% yield with 95:5% *er*) to the valine-based ligands.

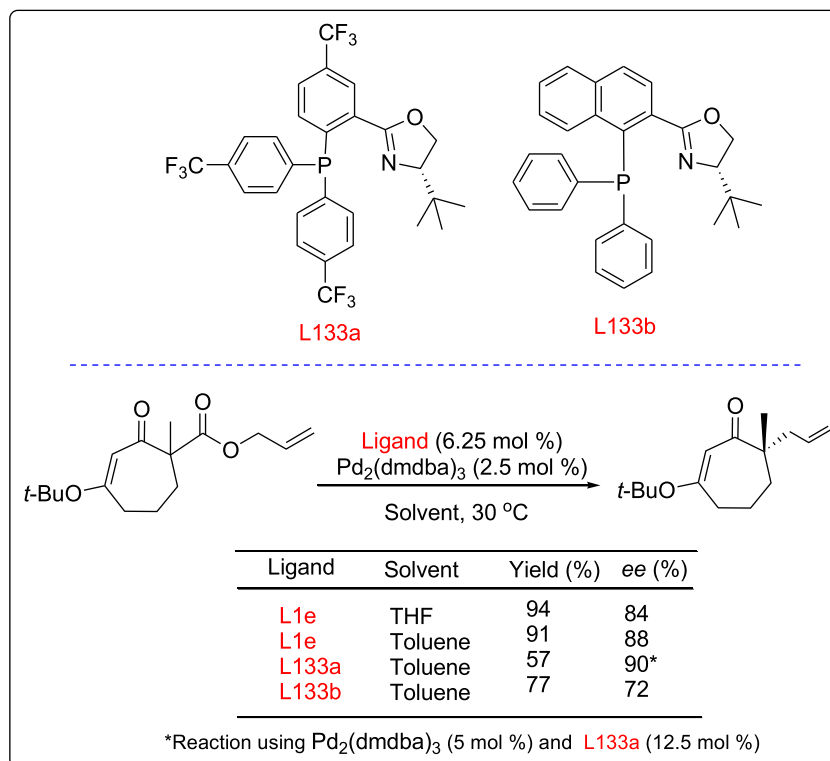
The ‘StePHOX’ [**L28a–h**] (8 mol %) also showed excellent activity and enantioselectivity for the asymmetric Tsuji allylation in THF in the presence of $\text{Pd}_2(\text{dba})_3$ (2.5 mol %) (Scheme 70).

In 2005, all the (*R,R*)-**L71a–d** ligands [143] were applied for the asymmetric allylation of simple ketones to generate both quaternary and tertiary centers from allyl enol carbonate using $\text{Pd}_2(\text{dba})_3\text{CHCl}_3$ (2.5 mol %) in toluene at 23 °C (Scheme 71). Among all, ligand (*R,R*)-**L71d** (5.5 mol %) gave the best *ee* (85%) and yield (88%), hence used in allylation of a range of five, six, and seven-membered ketones. It gave 78–99% yield with 78–99% *ee* of monoalkylated products. This protocol was also advanced for the preparation of many α -tertiary ketones that showed 98% *ee* through asymmetric α -allyl alkylation of acyclic ketones in dioxane at 23 °C [144]. Similarly, the catalytic system $\text{Pd}_2(\text{dba})_3/(\text{R,R})\text{-L71d}$ also showed an excellent yield and *ee* in the asymmetric allylic allylation of vinylogous thioesters under neutral conditions [145].

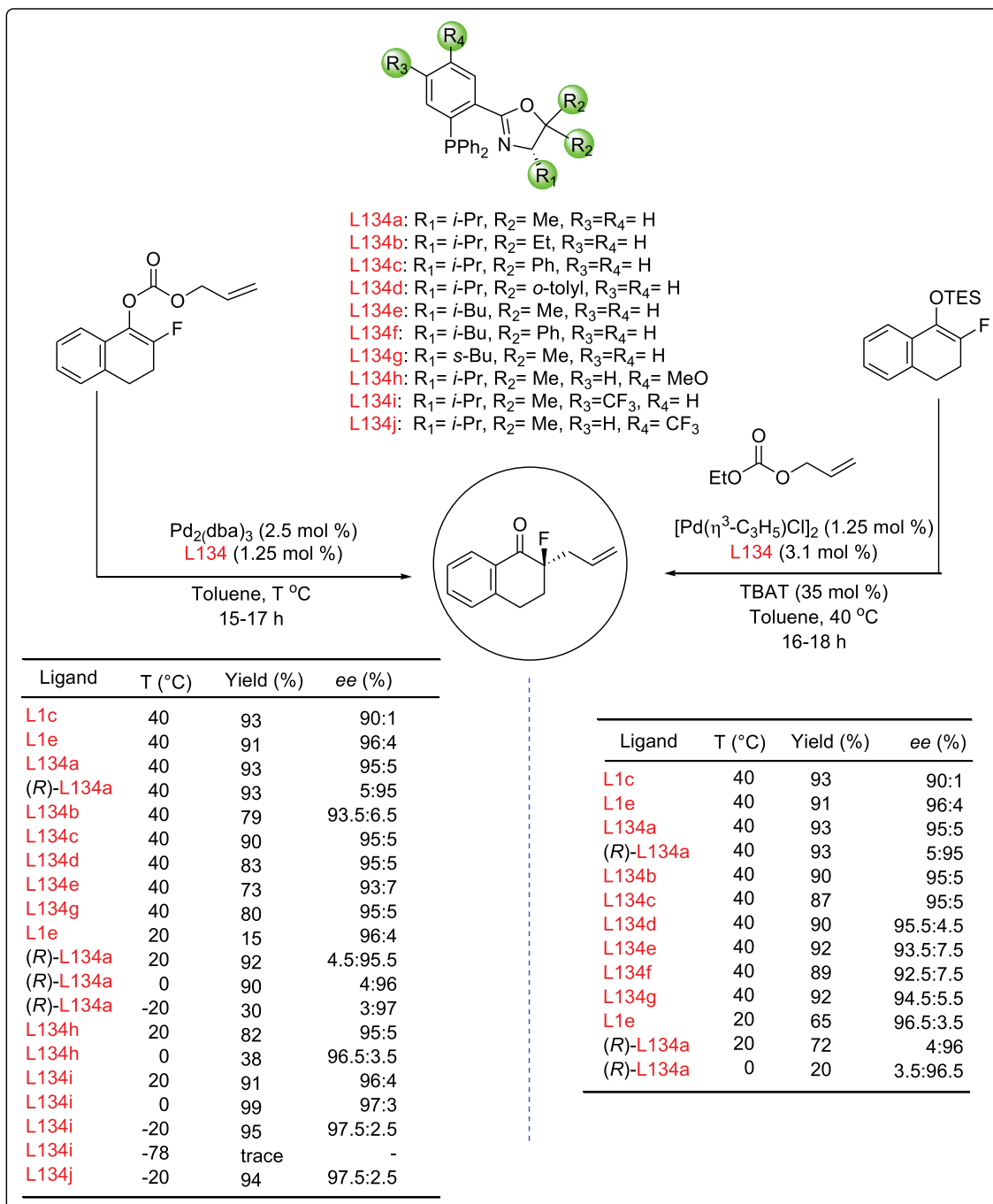
The (*S,S*)-**L71b** ligand was also successfully used in the asymmetric allylic alkylation of six-membered non-stabilized ketone enolates (Scheme 72) [146]. The ligand (5 mol %) with $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}]_2$ (2.5 mol %) gave an excellent yield (99%) and enantioselectivity (88% *ee*) in the presence of LDA and $(\text{CH}_3)_3\text{SnCl}$ in DME in 30 min. Under optimized conditions, a range of tetralone, benzylidene, and thioacetal derivatives are well alkylated with various alkylating agents. Linearly substituted allyl systems (either *E/Z* isomers) gave products of only *E* isomers with



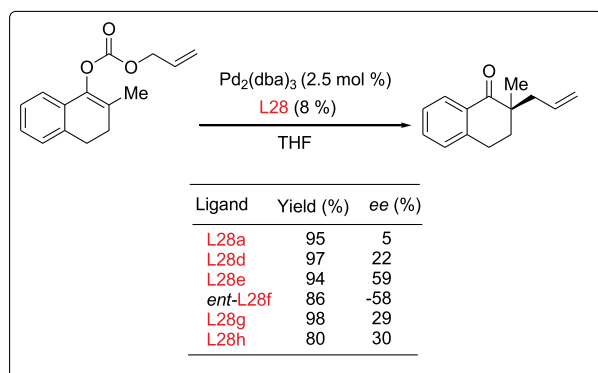
Scheme 67. Deracemization of quaternary carbon stereocenters and cascade enantioselective allylations using (*S*)-*t*-Bu-PHOX ligand.



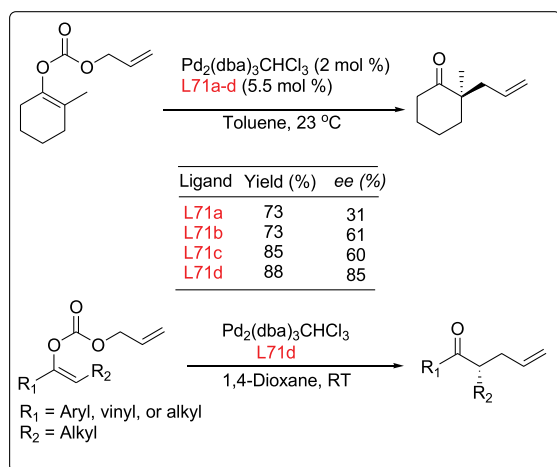
Scheme 68. Asymmetric alkylation of seven-membered methyl-substituted β -ketoesters using PHOX ligands.



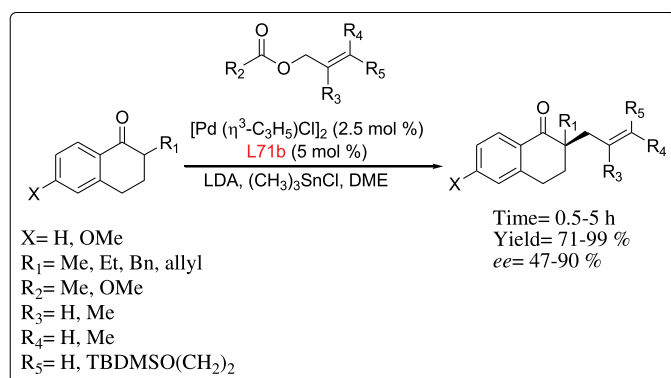
Scheme 69. Asymmetric allylic alkylations of fluorinated enol carbonates using PHOX ligands.



Scheme 70. StePHOX ligands for asymmetric allylic alkylations.



Scheme 71. Asymmetric allylic alkylations of simple ketones and vinyl-thioesters.



Scheme 72. Asymmetric allylic alkylations of non-stabilized ketone enolates of six-membered rings.

good *ee*, while a branched system gave lower *ee*. The activity, as well as enantioselectivity, was highly influenced by ring size and reaction temperature. Changing the ring size to either indanone or benzosuberone led to low *ee*. The benzylidene also gave the alkylated product with 98% yield and 82% *ee*. Moreover, in the case of the furanylidene derivative, the product yield was 89% with 79% *ee* at room temperature. The enantioselectivity increased to 92% at 0 °C. Similarly, at room temperature, the ketene thioacetal gave 70% *ee*, which was increased to 82% *ee* at -10 °C.

3.2. Heck coupling reactions

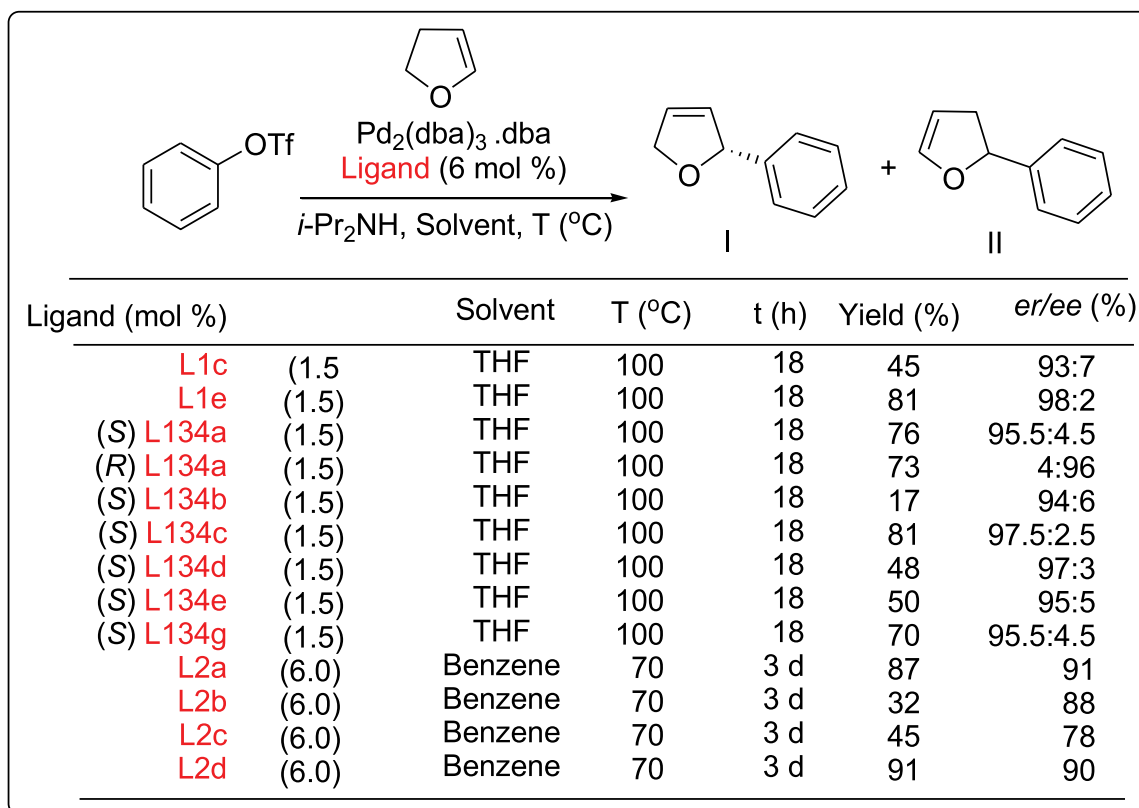
Due to their wide applications in organic transformations [147], the Heck coupling reaction [148] has been found to be most widely studied Pd-catalysed reaction as that other coupling reactions. Enantioselective intermolecular and intramolecular Heck coupling reactions [149] have been widely used extensively for the synthesis of many important compounds. This synthesis has been carried out using various P-N ligands.

The various PHOX ligands such as L1c, L1e, L2a-d, and L134a-g were applied for enantioselective Heck coupling reaction of 2,3-dihydrofuran and phenyl triflate (Scheme 73). The oxazoline ring containing a bulky R₂ substituent was beneficial for both reactivity and enantioselectivity. Thus, ligands L2a and L2d (6.0 mol %) provided desired products with 97% and 91% yield with 93% and 97% *ee* in the presence of Pd₂(dba)₃.dba (6 mol %) using *i*-Pr₂NEt as a base in benzene or THF at 78 °C [24]. Similarly, the ligands derived from leucine (L134e-f) and isoleucine (L134g) worked well as compared to the valine-based ligands (L134a-d) in the presence of Pd₂(dba)₃ (1.5 mol %), in THF at 100 °C under MW [142].

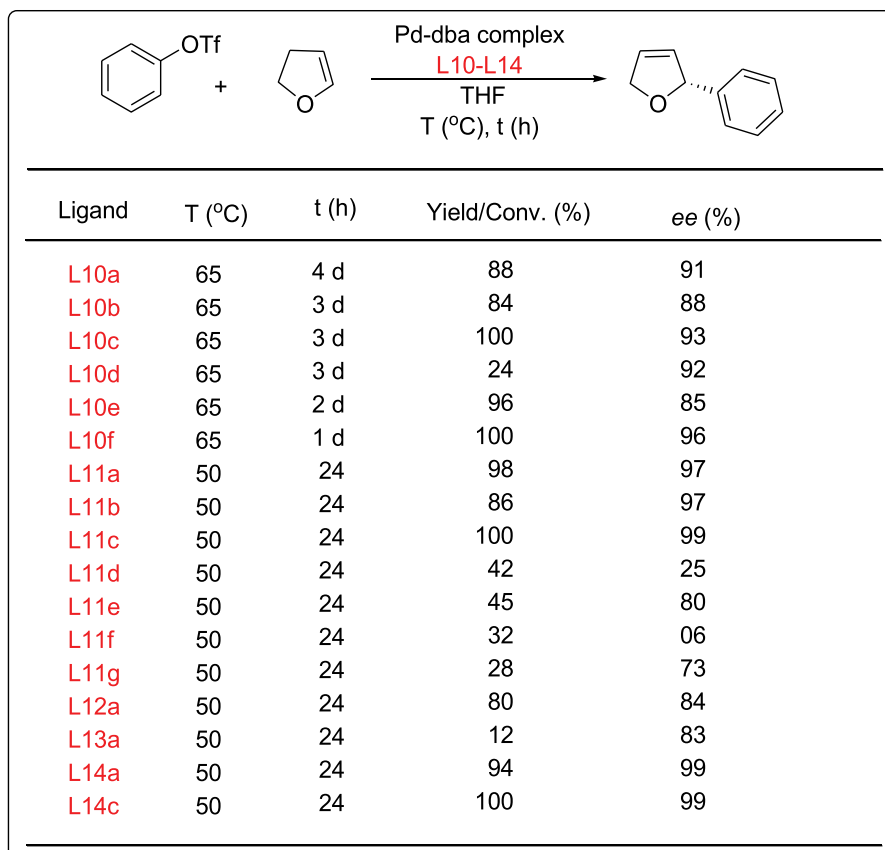
The carbohydrate-derived PHOX ligands (L10a-f) [150a] and L11-14 [150b,c] also gave high yields and enantioselectivity in the asymmetric Heck coupling reactions of 2,3-dihydrofuran and phenyl triflate in the presence of *i*-Pr₂NEt in THF (Scheme 74). The L10a-f (5.6 mol %) gave 2-aryl-2,5-dihydrofuran with up to 96% *ee* in presence of Pd(dba)₂ (5 mol %) at 65 °C. The L11-14 (5.6 mol %) gave 2-aryl-2,5-dihydrofuran with up to 99% *ee* in presence of [Pd₂(dba)₃].dba (5 mol %) at 55 °C. Ligands L12a and L13a showed lower activity as well as regio and enantioselectivities than ligands L11a. The activity as well as regio and enantioselectivity of the catalyst decreases (i.e., Ph > *i*-Pr > *t*-Bu) with an increase in the size of the group present on the oxazoline ring. The activity, regioselectivity, and enantiomeric excesses depend strongly on both the phosphite and oxazoline substituents, as ligand L11c and L14c containing SiMe₃ group at the *ortho* position of the biphenyl moiety showed excellent regio and enantioselectivity. In the presence of ligand L11a, the reaction of dihydrofuran and 1-cyclohexenyl triflate gave the coupling product with high enantioselectivity (98% *ee*) and regioselectivity (98%) with 100% conversion in 18 h. However, the reaction of cyclopentene and phenyl triflate gave the coupling product with 94% *ee* and 100% conversion, but there is the formation of a small amount of achiral 4-phenylcyclopentene at 70 °C. In addition, the reaction of phenyl triflate and 4,7-dihydro-1,3-dioxepin also proceeded with 92% *ee*. These ligands were also screened successfully in the Heck coupling reactions of several substrates under thermal as well as microwave conditions and showed high regio (up to 99%) and enantioselectivities (*ee* up to 99%).

The Heck coupling reactions of 2,3-dihydrofurans and 2,2-dialkylidihydro furans were carried out successfully by using the thiophene-based HetPHOX ligands (L18a-b, and L20a-c). The ligand showed excellent enantioselectivity (up to 95% *ee*) and yields (up to 97%). The ligand L18b containing the *t*-butyl group was the most active and enantioselective ligand [151].

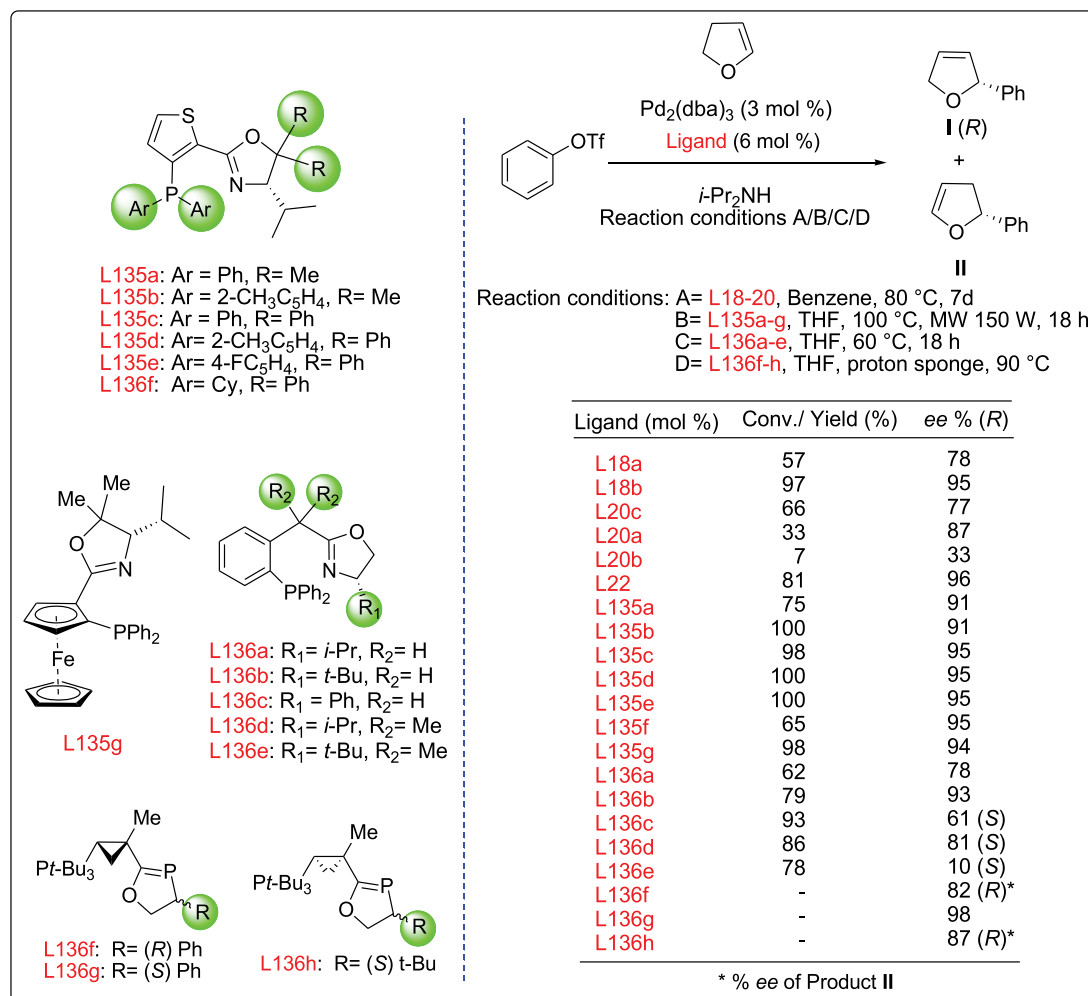
The Indanol PHOX ligand L22 (15 mol %) with [Pd₂(dba)₃.dba] also worked efficiently for the Heck coupling reaction of 2,3-dihydrofuran and phenyl triflate using *i*-Pr₂EtN as a base with high enantioselectivities (up to 96%) and yields (81%) in benzene at 70 °C [44]. Various gem-disubstituted and electronically varied thiophene-oxazoline (HetPHOX: L135a-f) and ferrocene-oxazoline (FcPHOX: L135g) ligands were used in asymmetric intermolecular Heck arylation of 2,3-dihydrofuran under microwave irradiation (100 °C, 150 W) (Scheme 75) [152a]. The ligand L135c (6 mol %) gave 95% *ee* using Pd₂(dba)₃ (6 mol %) in the presence of *i*-Pr₂NH as a base under MW in 18 h. The enantioselectivity increased to 97% in THF under conventional heating at 70 °C for 40 h. A dramatic switch of enantioselectivity was observed by a PHOX ligands (6 mol %) with and without substituents at the benzylic position in asymmetric Heck coupling of dihydrofuran and



Scheme 73. Enantioselective Heck coupling reactions using PHOX ligands.



Scheme 74. Asymmetric Heck coupling reactions using Carbo-PHOX ligands.



Scheme 75. PHOX and HetPHOX-based ligands for asymmetric Heck coupling reactions.

phenyl triflates [152b]. The ligands **L136a-c** offer products with (*R*)-configuration while ligands **L136d-e** resulted in products with (*S*)-configuration.

Similarly, dramatic stereo- and enantioselectivity were observed in the asymmetric Heck reaction using cyclopropane-based PHOX ligands (**L136f-h**) (6 mol %) using Pd(OAc)₂ (6 mol %) in the presence of proton sponge in THF at 60 °C [152c].

Due to their ease of preparation from readily available alcohols, phosphite ligands are extremely attractive and widely used for transition metal catalysis [153]. They are less sensitive to air and other oxidizing agents than phosphines and are applied for parallel synthesis.

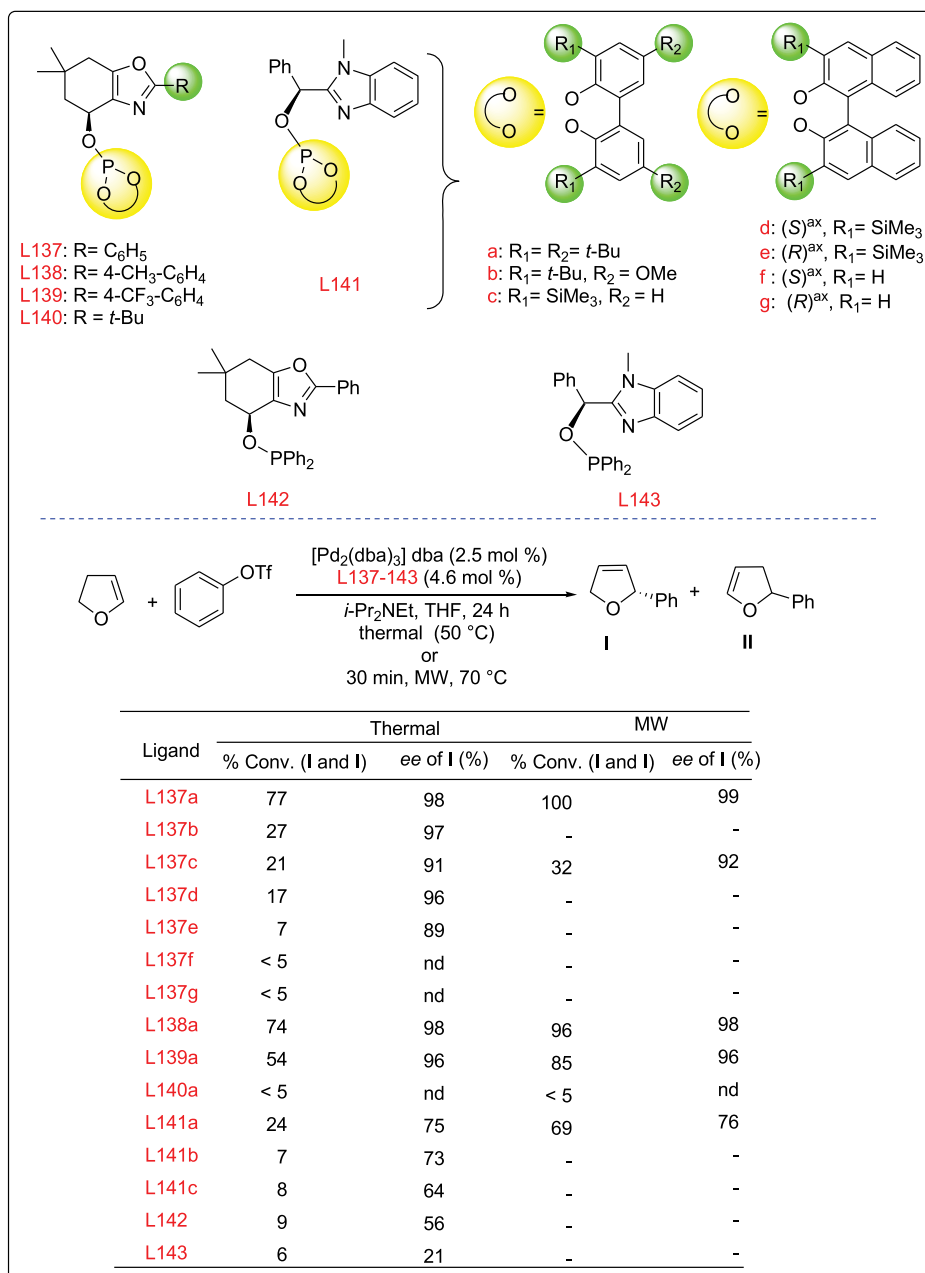
In this connection, Dieguez *et al.* [154] reported an air-stable phosphite oxazole/imidazoline **L137-L143** ligands for intermolecular asymmetric Heck coupling reactions under thermal and MW conditions (Scheme 76). The activity, regioselectivity, and enantioselectivity of ligands depend on the type of nitrogen donor group (oxazole or imidazole), the nature of oxazole, and biaryl-phosphite substituents, as well as the chirality of the biaryl moiety of the ligand. It was observed that MW is beneficial in terms of yield and time as an excellent activity (up to 100% conversion in 30 min), regioselectivity (up to > 99%), and enantioselectivity (up to 99% *ee*) were obtained using several triflates.

Busacca and coworkers developed a family of novel and flexible phosphinoimidazoline ligands similar to PHOX, having additional nitrogen-containing different R₃ substituents called 'BIPI' (Boehringer-Ingelheim Phosphino Imidazoline) ligands (**L144-149**) for intramolecular asymmetric Heck coupling reactions [155]. All these ligands

were applied for the creation of a chiral quaternary center. Its enantioselectivity depends linearly on phosphine electron density. The BIPI ligands were designed such that gross electronic tuning could be achieved by varying the substituents at R₂ or R₃. These groups are mostly alkyl and acyl groups. The alkyl groups R₂ and R₃ lead to strongly basic systems, while acyl and sulfonyl R₃ groups lead to more neutral ligands. The ligands were constructed in a modular fashion, in which *o*-haloimidate reacts with a chiral diamine and furnishes the haloimidazolines (Scheme 77). The S_NAr reaction of phosphite nucleophiles with haloimidazolines gave phosphinoimidazolines which could be converted into desired ligands in two different methods [156].

All the ligands (**L144-149**) were screened for intramolecular asymmetric Heck coupling reactions of triflates (**I-III**) in the presence of Pd₂(dba)₃, PMP (1,2,2,6,6-pentamethyl-piperidine), Ph₂O at 95 °C (Scheme 78). Initial screening was carried out for triflate **I** by varying the phosphine substituent R₁ holding R₂/R₃ as phenyl and R₄ as a 2-naphthoyl (**L144a-h**). The phosphine substituted with 3,5-difluorophenyl (**L144h**) gave oxindole with 78% *ee*, which was significantly higher than both (*R*)-BINAP and **L144e** under similar reaction conditions.

The screening of C₂-symmetric ligands **L145a-g** (R₂=R₃) and non-C₂-symmetric ligands **L145h-m** (R₂≠R₃) was carried out by varying R₂ and R₃ substituents by keeping the phosphine substituent R₁ as phenyl and R₄ as a 2-naphthoyl. The results showed that compared to neutral ligands electron-rich and electron-poor ligands gave increased enantioselectivity. In addition, all the ligands having different R₂ and R₃ substituents gave lower enantioselectivity. Thus, in terms of



Scheme 76. Phosphite oxazole-imidazoline ligands for asymmetric Heck coupling reactions.

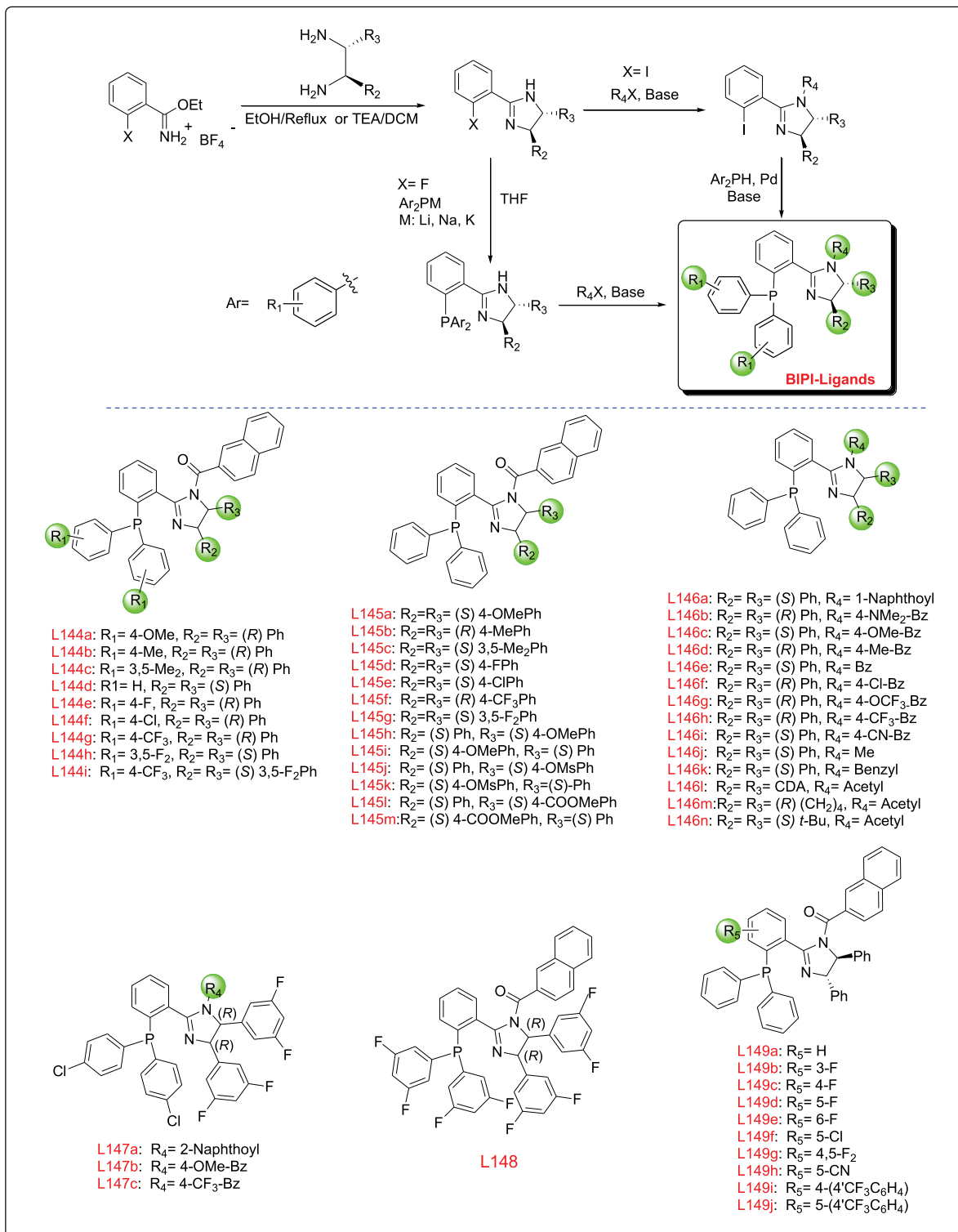
enantioselectivity (62.7% *ee*) and yield (93%), the best results were shown by the **L145g** containing a 3,5-difluoro system.

An examination of R₄ substituted ligands (**L146a-n**) doesn't have so much effect on enantioselectivity. Compared to ligands derived from aryl diamines or ligands with electron-withdrawing substituents on N, the ligands **L146a-n** gave the opposite enantiomer. Moreover, out of **L147a-c**, ligand **L147a** gave excellent enantioselectivity (80.6% *ee*). Thus, the ligands **L147a** having chlorophosphine with 2-naphthoyl as the optimum R₄ group showed 75.9–80.6% *ee* using different bases. However, the ligand **L148** containing R₂ and R₃ as difluorophenyl with 2-naphthoyl R₄ gave the highest enantioselectivity of 87.6%. They also briefly examined the effect of substituents R₅ on the central benzene ring (**L149a-j**) by keeping the standard phenyl substituent at R₁, R₂, and R₃ and holding R₄ as 2-naphthoyl. It was observed that **L149f** gave the best results.

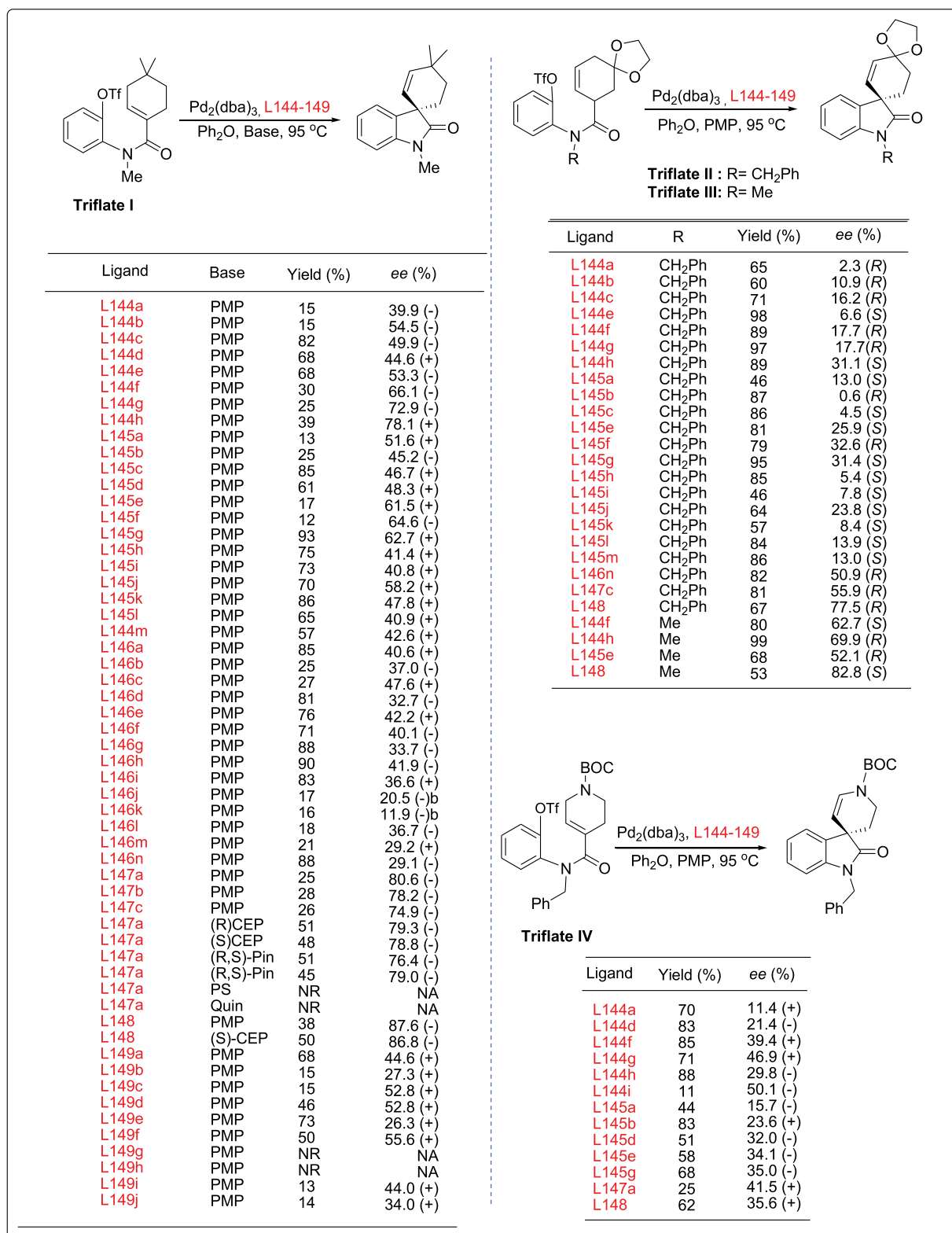
The screening of ligands for triflate (**II**) showed that the highest

enantioselectivity (77.5% *ee*) was obtained for ligand **L148**. For triflate (**III**), ligand **L148** was excellent in terms of *ee* (82.8%), while ligand **L144h** was excellent in terms of yield (99%) as well as *ee* (69.9%). The *p*-trifluoromethyl ligand (**L144g**) was practically good in terms of yield and *ee* for an asymmetric Heck coupling reaction of triflate **IV** bearing the protected tetrahydropyridine functionality.

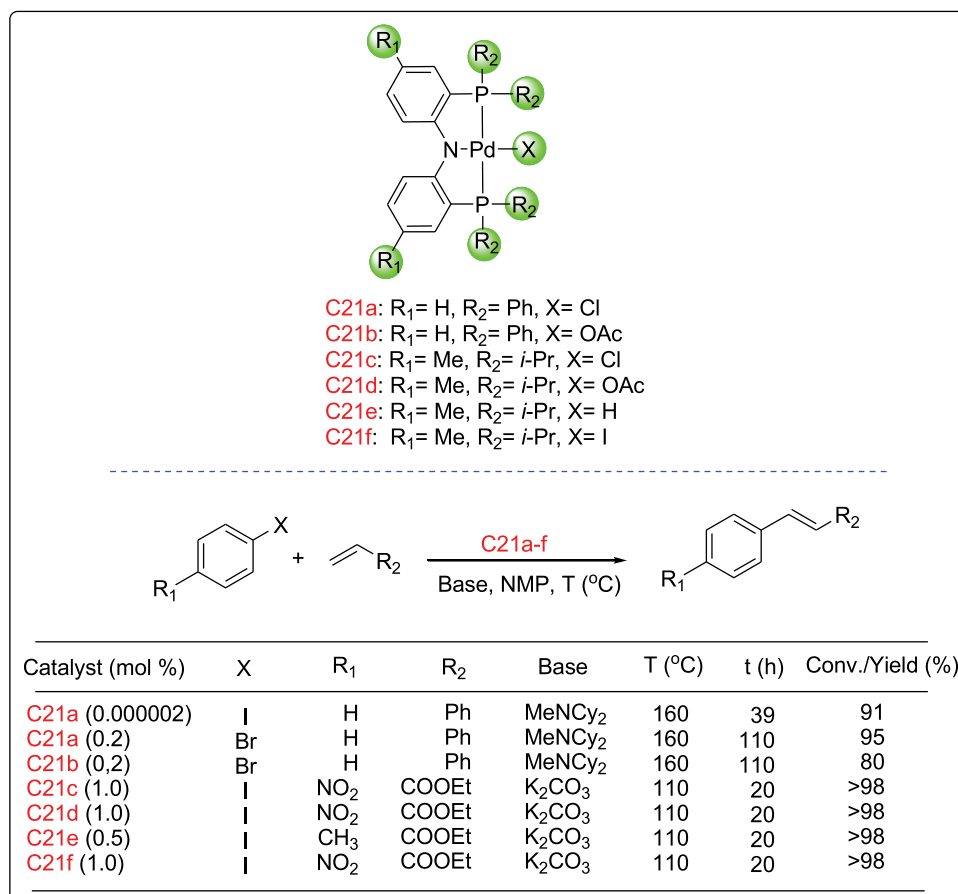
Guary and coworkers [157] synthesized indanol-derived ligand **L150** by direct metalation followed by a reaction with chlorodiphenylphosphine of oxazoline obtained from thiophene-2-carbonitrile (Scheme 79). The ligand was applied for intramolecular Heck coupling reactions of aryl iodides and aryl triflates in toluene at 80 °C in the presence of PMP as a base. The ligand (10 mol %) proved to be less successful both in terms of conversion (13%) and enantioselectivity (22% *ee*). The *ee* was increased up to 76% with the *tert*-butyl-substituted HetPHOX ligand **L18b** [43] in the presence of a proton sponge [1,8-bis(dimethylamino)naphthalene].



Scheme 77. Synthesis of various BIPI ligands.



Scheme 78. BIPI-PHOX for the synthesis of oxindole by intramolecular asymmetric Heck coupling reactions.



Scheme 81. P-N-P pincer-type complexes for the Heck coupling reactions.

(**C21a-b**) containing *o*-phenylene-derived amido diphosphine ligands. The complexes were obtained as brick red solids by the metathetical reaction of [PNP]Li(THF)₂ with PdCl₂(PhCN)₂ or Pd(OAc)₂ in THF at -35 °C. The utility of the complexes was investigated for the Heck coupling of a variety of aryl halides and styrene using Cy₂NMe as a base in NMP at 160 °C. The catalysts showed TON up to 4.5 × 10⁷ and TOF up to 1.1 × 10⁶ h⁻¹ (Scheme 81). Similarly, Ozerov et al. [164] reported the synthesis of PNP-Pd pincer complexes (**C21c-f**) wherein *o*-arylene units link the amido and phosphine sites that offer increased rigidity, devoid of β-hydrogens and moisture-sensitive functionalities. All the complexes (**C21c-f**) were used as pre-catalysts for the Heck coupling of aryl halides with ethyl acrylate in NMP and K₂CO₃ at 110 °C.

A Pd coordinated with Schiff bases ligands (iminophosphine) was found to be a remarkably active catalyst for many coupling reactions [165]. In this regard, Pfaltz et al. [166] applied an air and moisture-stable dimeric, acetate-bridged phosphine-free Pd (II) Schiff base complexes (**C22a-b**), as well as monomeric, air and moisture-stable phosphine-containing Schiff base complexes (**C22c-d**) for the Heck coupling reactions of bromobenzene with styrene. In the presence of NaOAc in DMA at 140 °C, essentially high yields of *trans*-stilbene were obtained with 0.01 mol % of the **C22a-b**, while complexes **C22b-d** gave slightly lower yields (Scheme 82). Moreover, catalysts **C22a-b** are more sensitive as compared to complexes **C22c-d**.

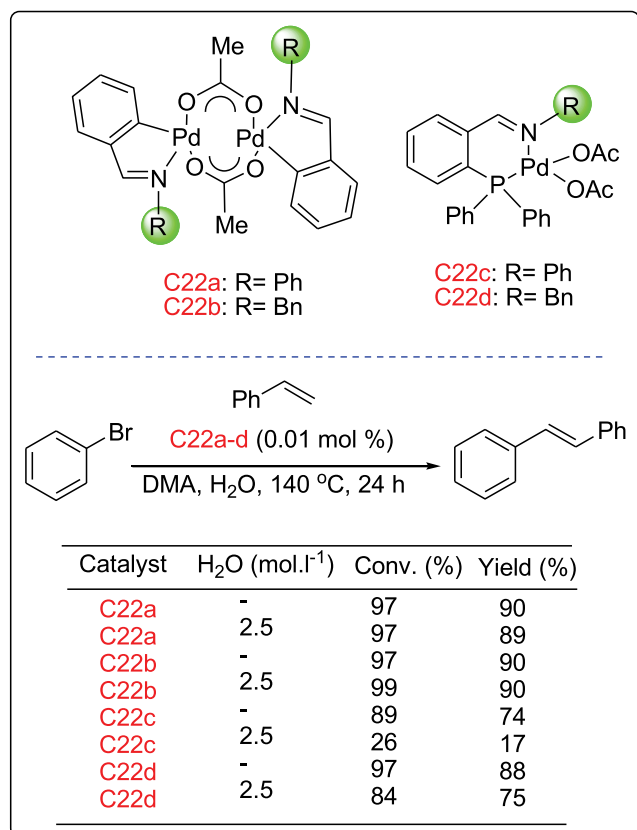
Singh et al. [167] used Pd(0) complexes of P-N ligands anchored on silica gel (**C23a-c**) for the Heck coupling reactions between aryl bromides and iodides with acrylic acid or styrene in *p*-xylene and TEA at 110–140 °C (Scheme 83). There was a selective formation of *trans* products in good yields (92%). The catalysts showed 15–16 times reusability without significant loss of activity. Recently, magnetically

separable iminophosphine Pd-Fe₃O₄-PEG catalyst was also applied for the Suzuki coupling reactions in pure water [168].

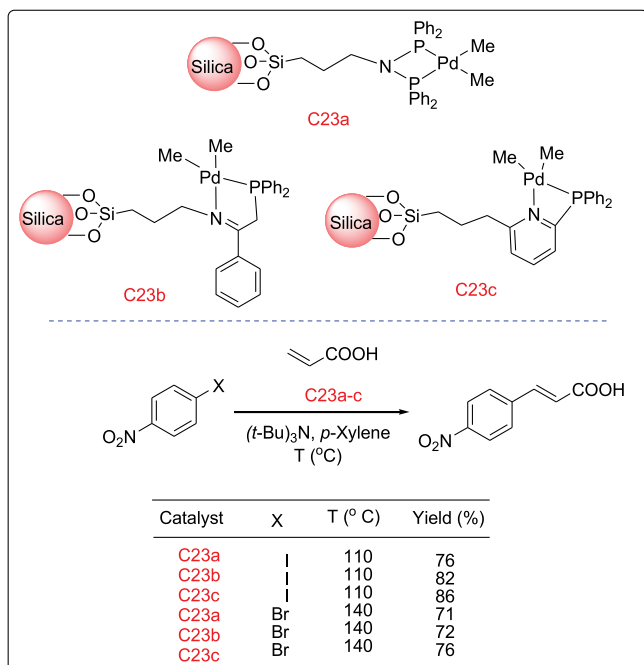
Yilmaz et al. [169] used new hemilabile ligand systems for the green synthesis of substituted olefins via Heck coupling reactions in supercritical CO₂ (Scheme 84). They designed Pd(II) complexes (**C24a-f**) bearing hetero donor phosphine-imine ligands for the reactions of aryl halides in DIPEA at 120 °C, in CO₂ (15 MPa) in the presence of Bu₄NBr. The catalytic system (0.5 mol %) leads to slower reaction times, good conversions, and selectivity, than the previously reported mono or bisphosphine Pd precatalysts.

Verkade et al. [170] reported [Pd₂(dba)₃/L151 (2:4 mol %) catalytic system for the one-pot synthesis of a variety of symmetrically and unsymmetrically substituted *trans*-4-*N,N*-diaryl aminostilbenes and *N,N*-diarylaminostyrenes (Scheme 85). This catalytic system was also used for both amination and intermolecular Heck coupling reactions. The 5, 6-dihydro-7H-dibenz[*b,d*]azepine was also synthesized in moderate yield in a one-pot amination/Suzuki coupling reaction using a Pd(OAc)₂/L151 (4:8 mol %) catalytic system.

Iranpoor et al. [171] used 1,3,2,4-diazadiphosphetidine-based phosphazanes as a new generation of heterogeneous bidentate ligands (L152a-d) for the Heck coupling reactions (Scheme 86). The reactions of aryl halides with different olefins were carried out under base-free conditions in the presence of PdCl₂ (2.5 mol %) in water at reflux. The ligand L152a was found to be highly efficient and can be reused at least 10 successive runs for the reaction of bromobenzene and styrene. The use of this ligand provides the possibility of performing the reaction in water under heterogeneous reaction conditions.



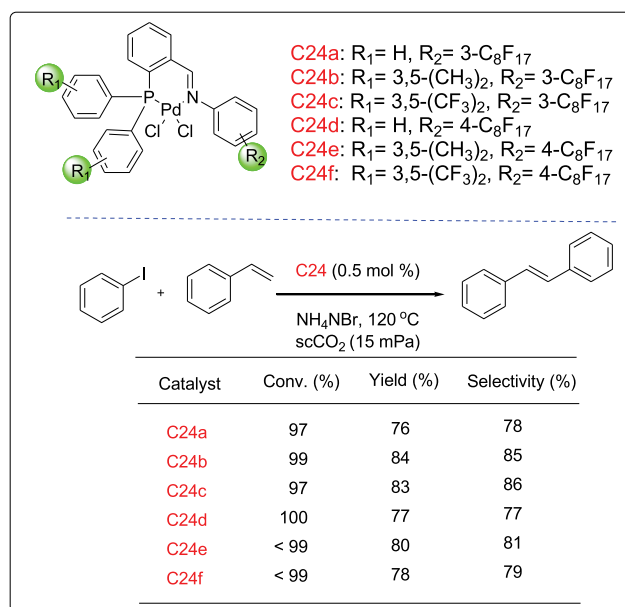
Scheme 82. Pd(II)-Schiff base complexes for the Heck coupling reactions.



Scheme 83. Pd(0) complexes of P-N ligands anchored on silica gel for the Heck coupling reactions.

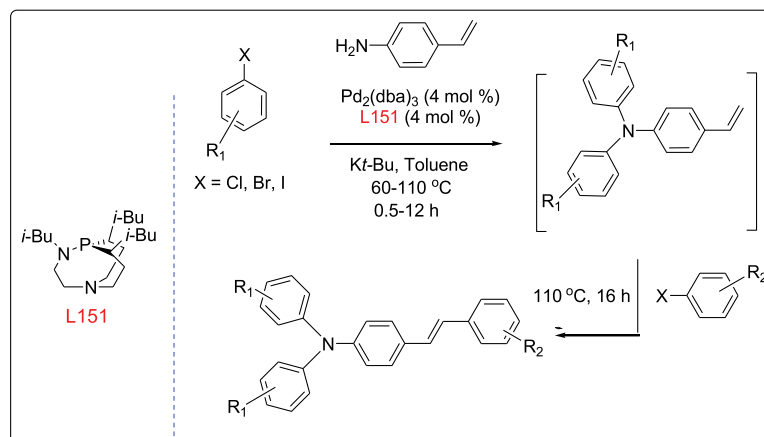
3.3. Suzuki coupling reactions

It has been known that asymmetric Suzuki coupling reactions have been widely used for the synthesis of many important compounds [172].

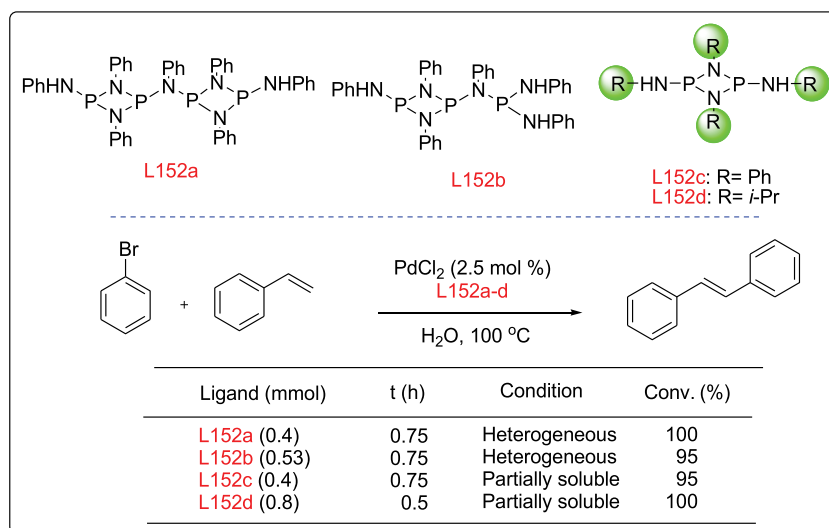
Scheme 84. Heck coupling reactions in supercritical CO₂ using phosphine-imine ligands.

In this connection, unsymmetrical chiral PCN pincer Pd(II) complexes **C25a-h** based on (imidazolyl)aryl phosphinite were synthesized by one-pot phosphorylation/palladation reaction via C-H bond activation [173]. All the complexes were characterized by XRD and adopted a typical distorted-square-planar geometry. The complexes showed moderate enantioselectivities in the asymmetric Suzuki coupling reactions (Scheme 87). It was found that except for **C25b**, with an *i*-Pr₂PO group, the other Pd complexes were effective catalysts in the coupling reaction of 1-iodo-2-methoxynaphthalene with 1-naphthylboronic acid. Thus, **C25a** was the most active catalyst, giving a 92% yield of the product but with less enantioselectivity (up to 26% *ee*). They proposed the cleavage of the Pd-C bond in the complexes to form the Pd(0) species that take part in the reaction.

Extensive work on a chiral phosphine hydrazone ligand was reported by Lassaletta and coworkers [174]. The bidentate **L153a-c** ligands were synthesized from commercially available 2-(diphenylphosphino)benzaldehyde and C₂-symmetric hydrazines. The ligands **L153d-h** with different PAR₂ groups were synthesized by reaction of the corresponding diarylphosphinobenzaldehydes with (2*S*, 5*S*)-1-amino-2,5-diphenylpyrrolidine. The ligand **L153i** was obtained by condensation of corresponding phosphine oxide with diphenyl-substituted pyrrolidine followed by reductive deoxygenation using HSiCl₃ in Et₃N and toluene at reflux (Scheme 88). The phosphine hydrazones **L153a-c** react with [PdCl₂(CH₃CN)₂] and afforded **C27a-c** a neutral Pd(II) complexes. The PdCl₂(CH₃CN)₂/**L153a-i** (5:5 mol %) were tested in the asymmetric Suzuki coupling reaction of 1-bromo-2-methoxynaphthalene and 1-naphthyl boronic acid for the synthesis of axially chiral biaryls in the presence of Cs₂CO₃ in toluene at room temperature. The pyrrolidine-based phosphine hydrazones (**L153a** and **L153c**) provided a better chiral environment than the piperidine-based analog (**L153b**). This result was attributed to the higher conformational rigidity associated with the stronger n-π conjugation. Though the enantioselectivity (*er* = 95:5) was observed previously for bis-hydrazone ligands, the ligands **L153a** and **L153c** were promising alternatives in terms of catalytic activity. The ligands **L153i** showed poor catalytic activity and afforded the racemic mixture of the desired product. The reactions carried out using preformed complexes **C27a** and **C27c** afforded improved results. The scope of the protocol was explored in the



Scheme 85. $[\text{Pd}_2(\text{dba})_3/\text{P}(\text{i-BuNCH}_2\text{CH}_2)_3\text{N}]$ for the one-pot synthesis of a diaryl amino stilbenes and N, N-diaryl amino styrenes.



Scheme 86. Diazadiphosphetidine-based phosphazanes as a heterogeneous bidentate ligand for the Heck coupling reactions.

asymmetric Suzuki coupling of a variety of aryl bromides and triflates with aryl boronic acids. The functionalized biaryls afforded good yields and enantioselectivities.

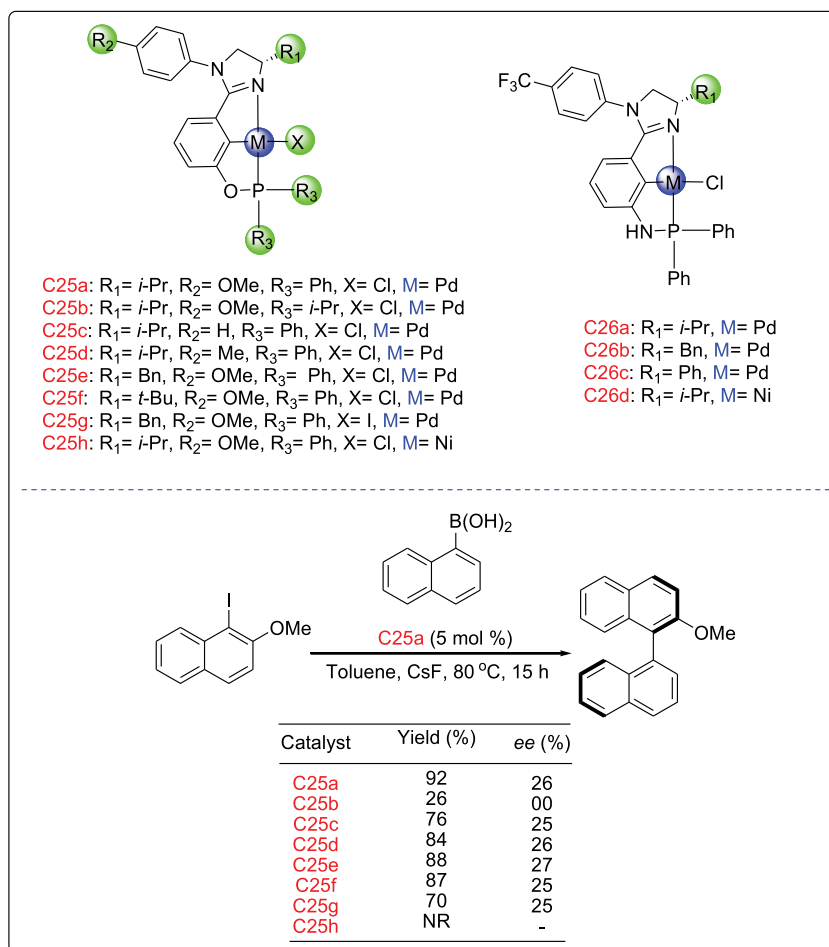
Recently, the dynamic kinetic asymmetric transformation (DYKAT) technique was used for the synthesis of 2'-substituted 2-aryl pyridines/isoquinolines and related heterobiaryls [175]. Such compounds were obtained by the Pd-catalyzed coupling of racemic 2-triflates with aryl boroxines using a monodentate TADDOL-derived phosphoramidite. The ligand **L154** gives good to excellent enantioselectivities. The catalytic system $[\text{Pd}_2(\text{dba})_3]/\text{L154}$ (10:10 mol %) showed complete conversion of *p*-anisyl boroxine with 1-(isoquinolin-1-yl)naphthalene-2-yl triflate into expected coupling product with 91% yield and 34% *ee* in the presence of Cs_2CO_3 as a base in THF at 55 °C (Scheme 89). Alternatively, $\text{Pd}(\text{dba})_2/\text{L154}$ (5:5 mol %) led to 76% yield and 84% *ee* in dioxane. Moreover, several combinations of heterobiaryl triflates with arylboroxines afforded desired products in good to excellent yields and high *ee* in the presence of Cs_2CO_3 as a base in dioxane at 25–55 °C.

A chiral imidazoindole phosphines supported on amphiphilic polystyrene-poly(ethylene glycol) resin (**L123a-c**) (10 mol %) was successfully carried out asymmetric Suzuki coupling of 2-substituted 1-iodonaphthalenes and 2-substituted naphthalene-1-ylboronic acid into

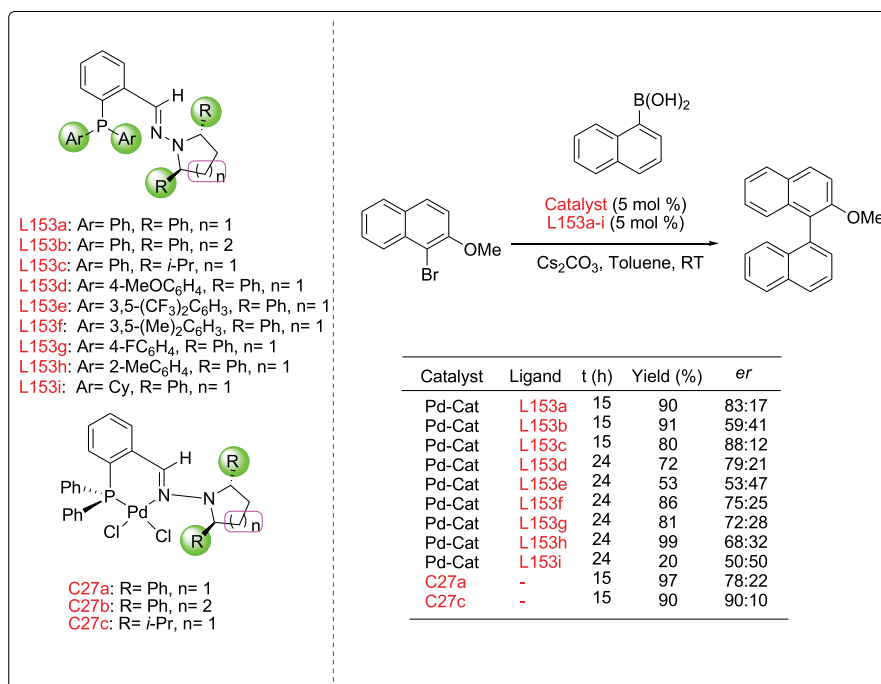
(*S*)-2,2'-disubstituted 1,1'-binaphthyl in the presence of $\text{Pd}(\text{OAc})_2$ (10 mol %) with 94% *ee* in water (Scheme 90) [176]. XRD analysis of the complex **L123a-PdCl₂** showed that the ligand coordinates to Pd(II) in a P, N-chelating fashion. The catalytic system could be recycled at least 4 times with consistent activity and selectivity.

Dupont and co-workers [177] reported the air and water stable, non-symmetrical NCP type pincer complex **C20c** for the Suzuki coupling of arylboronic acids with electron-rich and electron-poor aryl chlorides. The catalyst furnished the corresponding coupling products with excellent yields (70–98%) in the presence of CsF as a base in dioxane at 130 °C. The complex **C20c** (0.5 mol %) was also active in coupling sterically hindered bromomesitylene with a 2-methyl phenylboronic acid in CsF, TBAB, and dioxane at 130 °C. A series of aryl bromides and iodides undergo easy coupling with various aryl boronic acids in the presence of complex **C20c** (0.2 mol %) in K_3PO_4 and dioxane at 100 °C and 130 °C (Scheme 91).

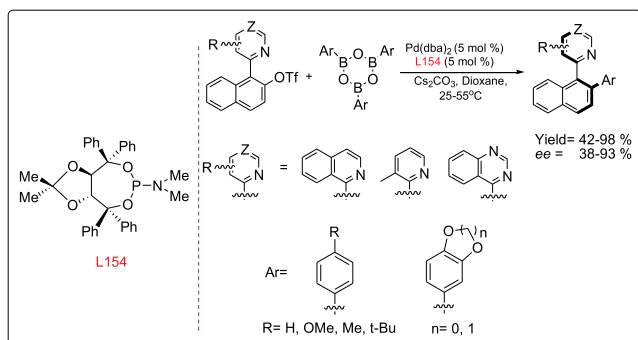
The catalysts **C21a** and **C21c** were applied for the Suzuki coupling reactions [178]. Several electronically activated, unactivated, and deactivated (hetero)aryl bromides and iodides were coupled with aryl boronic acids using $\text{K}_3\text{PO}_4 \cdot \text{H}_2\text{O}/\text{K}_3\text{PO}_4$ as a base in 1,4-dioxane/toluene at 110–80 °C under aerobic conditions.



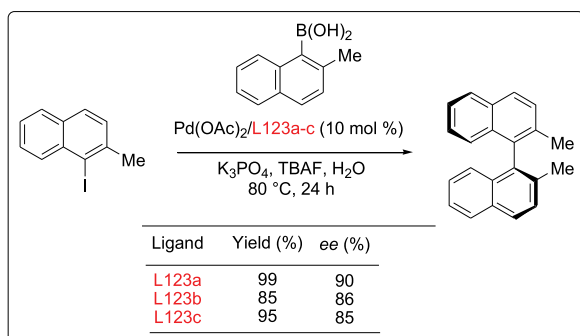
Scheme 87. Chiral Pd(II)-PCN pincer complexes for asymmetric Suzuki coupling reactions.



Scheme 88. Phosphino hydrazones for asymmetric Suzuki coupling reactions.



Scheme 89. TADDOL-derived phosphoramidite for coupling of racemic 2-triflates with aryl boroxines.



Scheme 90. Recyclable amphiphilic resin-supported P-N chelating Pd complexes for asymmetric Suzuki coupling reactions.

Arriortua et al. [179] prepared PCN Pd pincer complex (**C28**) containing a phosphinoamino group. The complex was readily obtained from available 1-(3-nitrophenyl)pyrazole and Pd(COD)Cl₂ in toluene. The structure of the complex was confirmed by ¹H NMR, DEPT, HSQC,

and X-ray diffractometry analysis. The complex was tested in the Suzuki coupling reactions in aqueous media. Various biaryls were obtained in excellent yields in the presence of **C28** (0.01 mol %), and K₂CO₃ in H₂O at 100 °C for 2 h (**Scheme 92**).

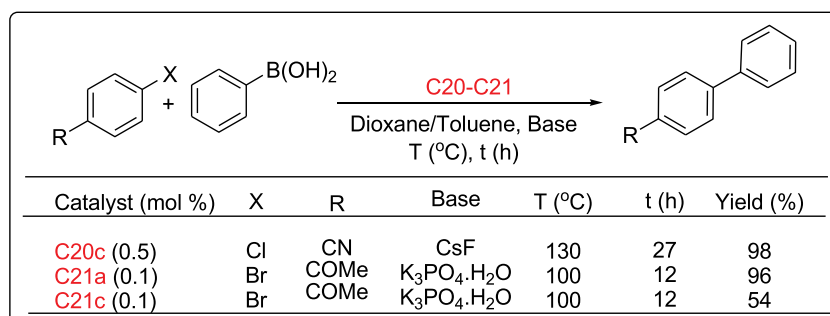
Song et al. [180] reported the facile synthesis of achiral and chiral PCN pincer Pd(II) complexes (**C29a–e**) in two steps from commercially available 3-hydroxybenzaldehyde and amines (**Scheme 93**). The structure of **C29a** and **C29b** were determined by single-crystal X-ray analysis. After optimization of all the catalysts, complex **C29a** (0.1–1.0 mol %) was found to be an effective catalyst for the Suzuki coupling reactions of various aryl bromides with phenylboronic acids in K₃PO₄·7H₂O, EtOH at 50 °C, under air.

Similarly, Xia et al. [181] synthesized iminophosphinite pincer Pd complexes **C29f–j** and evaluated catalytic potential (0.010 mmol) in the Suzuki coupling reactions of various aryl bromides and chlorides with phenylboronic acid in the presence of Cs₂CO₃ in dioxane at 100 °C. The catalyst preparation involves the complexation of imines with Pd₂(dba)₃ to give the Pd iminophosphinite pincer complexes **C29f–i**. The reaction of **C29g** with AgTFA in THF at room temperature afforded the Pd iminophosphinite trifluoroacetate complex (**C29j**).

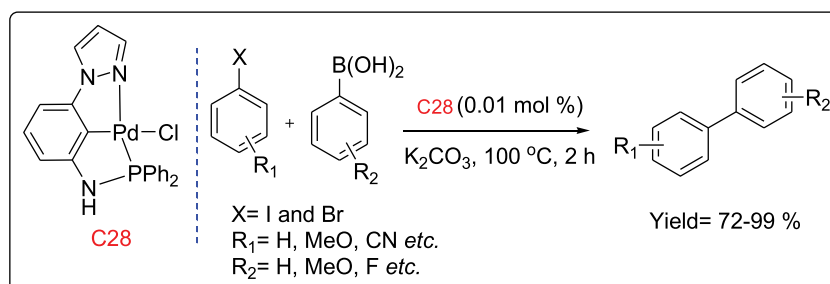
Cyclopalladated azido complexes (**C30a–h**) containing ligands with C and N-donor atoms were studied for the Suzuki coupling reactions (**Scheme 94**) [182]. The activity of complexes (1 mol %) was evaluated in the coupling of 4-acetylbromobenzene with phenylboronic acid in the presence of K₂CO₃ in toluene at 80 °C. All the catalysts exhibited high activity. However, the reactions involving aryl chloride showed poor reactivity. The catalysts **C30h** also exhibited excellent activity in ethanol.

Recently, Kapdi et al. [183] used triple Suzuki coupling reactions for the synthesis of anthracene-based OLED emitters using K₂CO₃ as a base in aqueous THF (**Scheme 95**). The synthesis involves the use of palladacyclic complexes (**C31a–b**) containing imidate ligands. All the synthesized molecules were studied by UV–vis spectroscopy and TG-DSC analysis. The DFT studies of the different properties showed that all the derivatives possess a more significant hole mobility character than electron transfer capability.

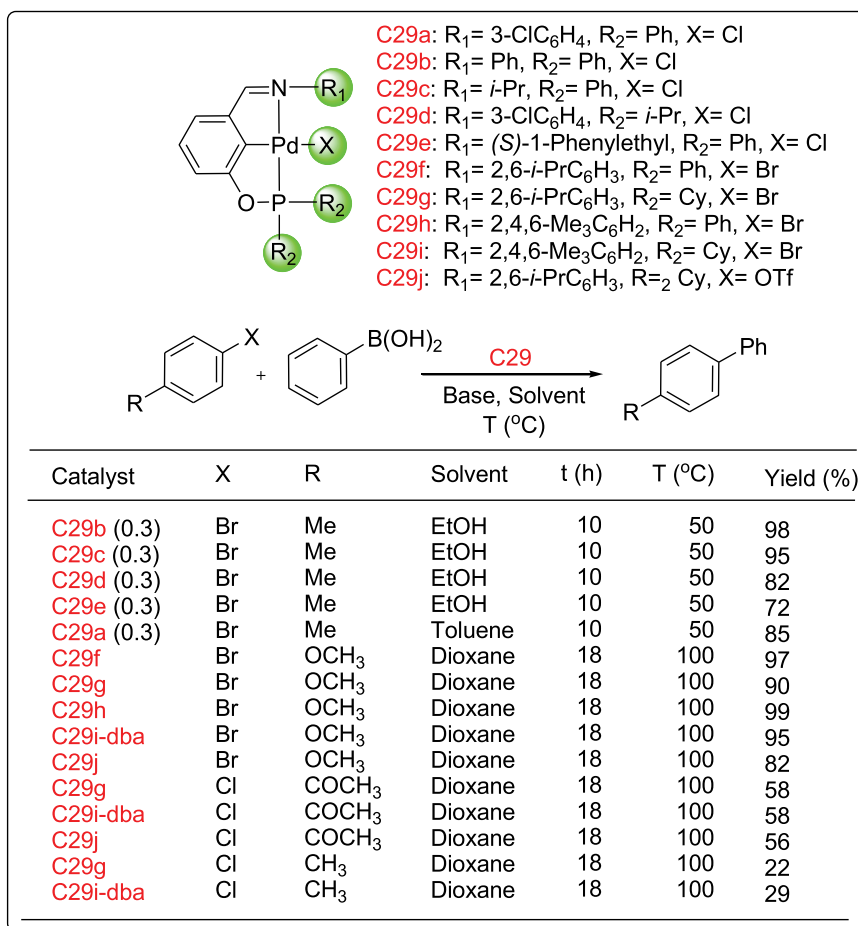
Scrivanti et al. studied the role of olefin in catalyst stabilization by using η²-(olefin)Pd(0) complexes containing iminophosphine ligands



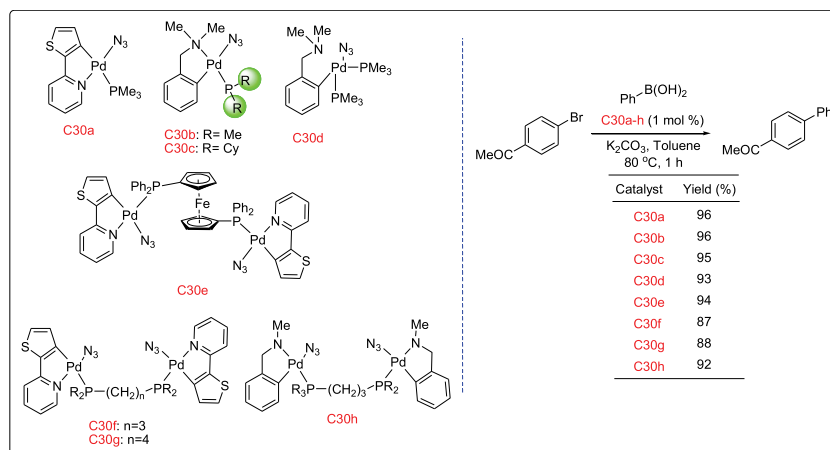
Scheme 91. Pd(II)-NCP pincers for the Suzuki coupling reactions.



Scheme 92. Pd(II)-NCP pincers for the Suzuki coupling reactions.



Scheme 93. Iminophosphinite Pd pincers for the Suzuki coupling reactions.



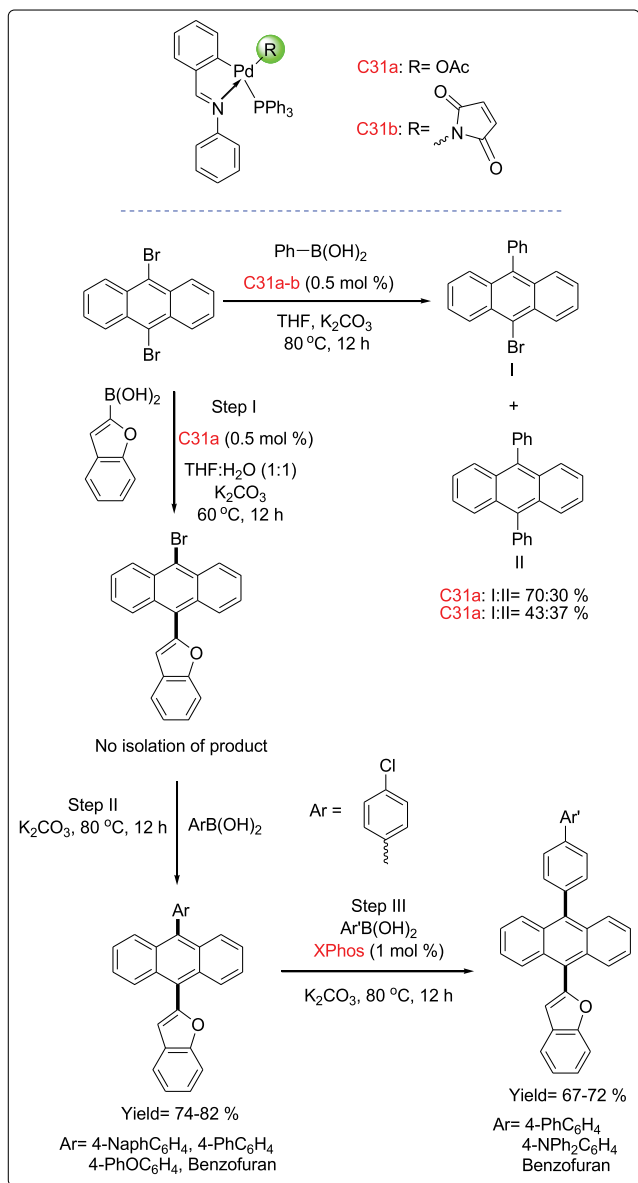
Scheme 94. Cycloplalladated azido complexes for the Suzuki coupling reactions.

(C32a-e and C32h) in the Suzuki coupling reactions [184]. The coupling of 4-bromoacetophenone with phenylboronic acid showed that C32c was the best catalyst in toluene in the presence of K₂CO₃ at 110 °C. The catalyst obtained high efficiency with TON up to 10⁵ h⁻¹. The system is also applied for the synthesis of undecaaryl substituted corroles (Scheme 96).

Buono and coworkers [185] synthesized a tunable P-stereogenic P, N-phosphine ligands (L155), and their Pd complex (C33). Both were fully characterized by spectroscopic and crystallographic studies. The

new ligand and catalyst were tested in the Suzuki coupling reaction with 92% yields, and the intramolecular α -arylation of ketones (Scheme 97).

Kwong and coworkers explored and developed a new class of indolylphosphine ligands for the Suzuki coupling reactions. They applied Pd(OAc)₂/(L156a-h) system for aryl chlorides [186]. The ligands L157a-c and L158a-e also used in the Suzuki coupling reactions of aryl tosylates [187] under different reaction conditions (Scheme 98). Dramatic reactivity differences were observed when the position of the phosphino group interchanged (L157b and L158d). It may be the phosphino group



Scheme 95. Triple Suzuki coupling reactions for the synthesis of OLED emitters.

on the aryl ring instead of on the heterocyclic ring provides better geometry of chelation to the Pd and facilitates the oxidative addition of the Ar-OTs. Similarly, a new P, N-type “PhMezole-phos” ligands (**L159a-c**) acts as an efficient ligand for the Suzuki coupling reactions of sterically hindered aryl chlorides in presence of K₃PO₄·H₂O in dioxane:mesitylene

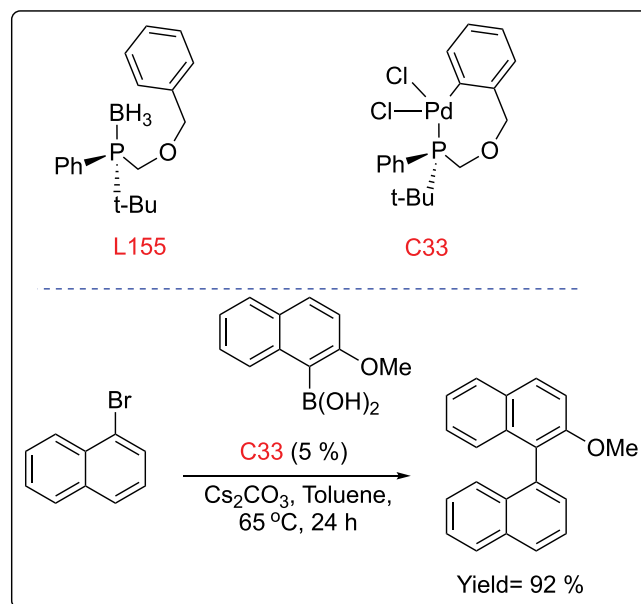
(1:2) mixture at 135 °C [188]. The catalyst loading could be reduced to 0.001 mol % Pd with 80000 TON.

Sarkar *et al.* [189] synthesized pyrazole-derived (**L160a-f**) and pyrrole (**L160g-j**) bidentate ligands for the Suzuki coupling reactions of the aryl bromides and chlorides. The ligand **L160b** gave an excellent yield of the coupling products in presence of Pd₂(dba)₃ and CsF in toluene at 65–85 °C (Scheme 99). An enhanced catalytic activity due to steric crowding in the Pd complex suggested the participation of a chelated structure in the intermediate catalytic steps.

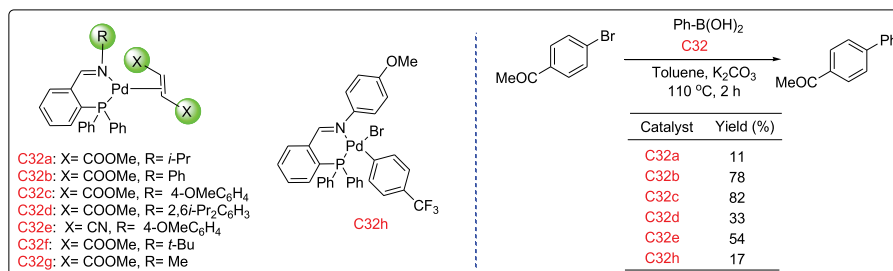
It was known that strongly basic and sterically demanding ligands activate the less reactive aryl chlorides. In this regard, Scrivanti *et al.* [190] reported [PdCl₂(P–N)] complexes (**C34a-b**) containing the basic and sterically demanding 8-(di-*t*-butylphosphinoxy)-quinoline (P–N) ligand. The complex **C34a** showed good to excellent activity for the Suzuki coupling reactions of the aryl bromides and chlorides with phenylboronic acid using K₂CO₃ as a base in toluene at 110 °C (Scheme 100). The catalytic activity of **C34b** was almost identical to that of **C34a**.

Frost *et al.* [191] synthesized water-soluble Pd complexes of trisubstituted PTA derivatives called PTA_{R3} (**L161**) and their precatalysts (**C35**). The **L161c** with Pd(OAc)₂ or PdCl₂ was found to be the most active catalytic system for the Suzuki coupling of aryl bromides and phenylboronic acid in water: acetonitrile (1:1) mixture at 80 °C. The Pd(OAc)₂/**L161c** system showed good yields with comparable activity to TPPMS and TPPTS. While less active than water-soluble diamines and phosphines like TXPTS (Scheme 101).

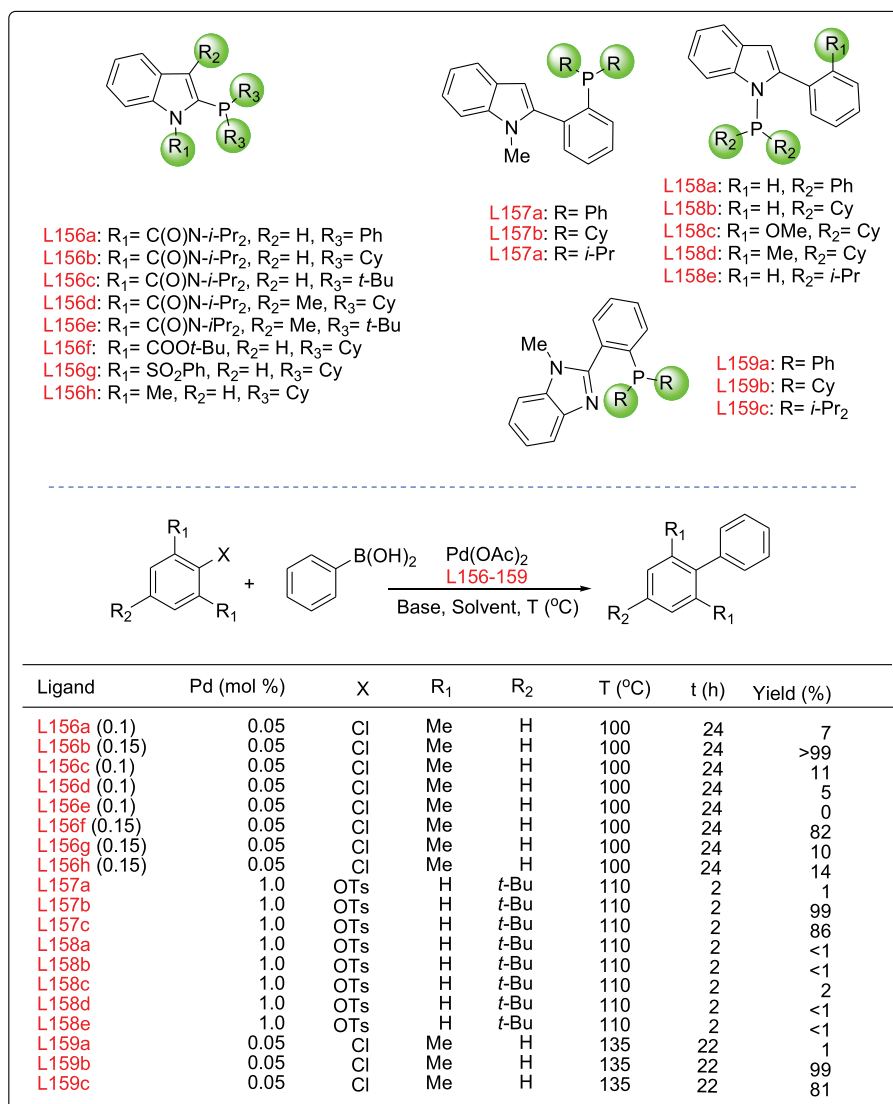
Floch *et al.* [192] synthesized Ni (**C36a**) and Pd (**C36b**) complexes



Scheme 97. Tunable P-stereogenic P–N phosphine ligands for the Suzuki coupling reactions.



Scheme 96. Pd(0)-Schiff base complexes based on P–N ligands for the Suzuki coupling reactions.



Scheme 98. Indol-derived aminophosphines for the Suzuki coupling reactions.

containing PNNP tetradentate ligand. The ligands were obtained from commercially available bis(diphenylphosphino)methane and ethylenediamine (Scheme 102). The activity of complexes was examined in the Suzuki coupling reactions of bromobenzene with phenylboronic acid. The reaction was carried out in toluene, water, and toluene: water mixture (1:1) in the presence of a TBABF₄ at 60 °C. The Pd complex **C36b** was more efficient than their corresponding nickel complex (**C36a**). Though the complex **C36b** (1 mol %) was coupled activated aryl chlorides successfully, the yields were lower (34–44%). While chlorobenzene was almost unreactive. The aqueous phase containing catalyst **C36b** can be separable easily from the organic phase and reused at least four times without any noticeable loss in activity.

Various P-imidate Pd complexes **C37a-d** were used for the Suzuki coupling reactions [193]. A variety of aryl halides were coupled with aryl boronic acids. The corresponding cross-coupled products were obtained in good to excellent yields using **C37a** and **C37b** (1 mol %) in a THF at 60 °C (Scheme 103).

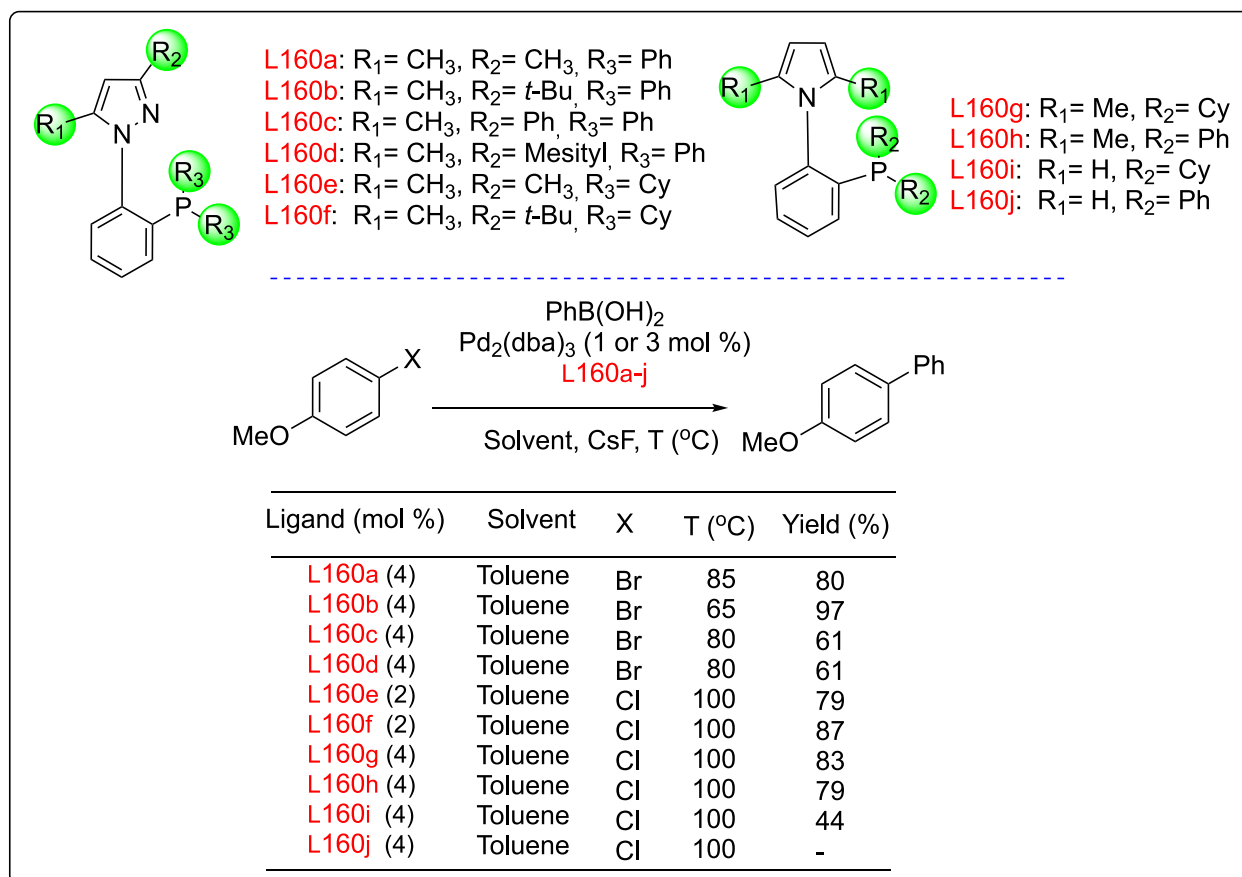
The water-soluble Pd-imidate complexes (**C38a-d**) were developed as highly efficient catalysts for the synthesis of C5-arylated pyrimidine nucleosides (Scheme 104) [194]. The complexes were prepared by the direct reactions of *trans*-[Pd(imidate)₂(SMe₂)₂] and 1,3,5-tri-aza-7-phospha adamantane (PTA). Moreover, the catalyst **C38d** (1 mol

%) was found to be a more efficient catalyst for the synthesis of 5-arylated deoxyuridine analogs using 5-iodo-2'-deoxyuridine and various phenylboronic acids in the presence of TEA in water at room temperature or 80 °C. After completion of the reaction, the product was isolated by extracting the aqueous layer with ethyl acetate and then recharging the aqueous phase with fresh substrates and bases. It was found that the catalyst was active for five consecutive runs without significant loss of activity.

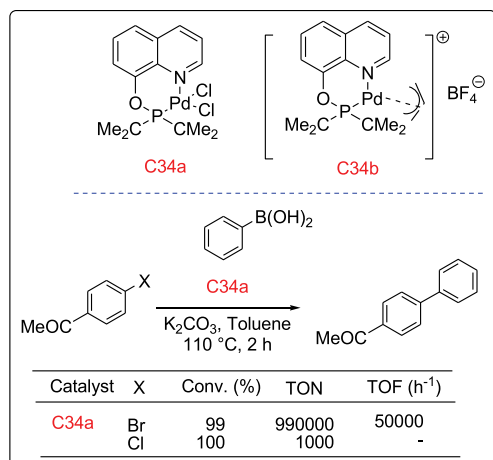
3.4. Sonogashira coupling reactions

The Pd-catalyzed Sonogashira cross-coupling reactions between aryl halides and acetylenes are an important reaction for the synthesis of acetylene derivatives [195].

The firstly prepared Pd-PNF pincer-type complex (**C39**) bearing a halogen (fluorine) arm facilitated the Sonogashira coupling reactions [196] (Scheme 105). This 8-fluoroquinoline-based chelated Pd(II) complex was prepared *via* the base-assisted dearomatization of a phosphine-quinoline (P-N) ligand. This dearomatization is reversible and facilitates catalytic coupling. The catalyst (1 mol %) showed moderate to excellent catalytic activity for the Sonogashira coupling of various aryl iodides and bromides in NaOt-Bu and benzene at 55 °C.



Scheme 99. Pyrazole-tethered aryl phosphine ligands for the Suzuki coupling reactions.



Scheme 100. Quinoline-based P–N ligands for the Suzuki coupling reactions.

The PCN-Pd pincer complex **C28** was also tested for the copper-free Sonogashira coupling reactions in aqueous media [179]. The complex was also applied to the arylation of phenylacetylene with a series of aryl iodides, providing moderate to good yields under three reaction conditions (Method A: Et₃N, CH₃CN, room temperature, 12 h; Method B: pyrrolidine, H₂O, 50 °C, 24 h; Method C: 100 °C, 6 h). The catalyst was highly active for various aryl iodides with phenylacetylene (Scheme 106). In addition, the complex **C29a** (1 mol %) was also applied to

copper-free Sonogashira coupling reactions of halobenzene, and acetylenes in Cs₂CO₃ and MeOH under air at room temperature [180].

Wendt and coworkers [197] reported aromatic PCN-Pd pincer complexes (**C40a-c**) for the decarboxylative cross-coupling reactions of phenyl propionic acid and iodobenzene (Scheme 107). The best result (54% yield) was shown by the **C40a** using a catalytic amount of CuI in the presence of K₂CO₃ as a base in MeCN at 135 °C for 48 h.

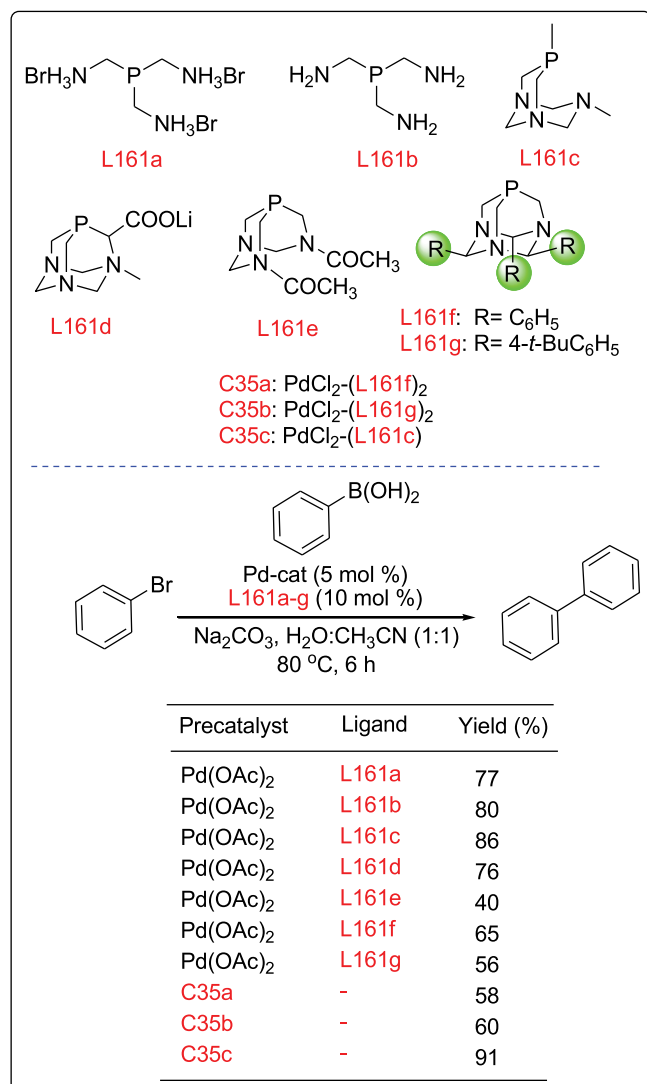
3.5. Stille coupling reactions

The Stille coupling reactions are powerful organic transformations, used for the synthesis of important biological compounds [198].

In this regard, Koprowski *et al.* [199] applied Pd complexes (**C41a-d**) of a variety of β and γ -iminophosphines as well as Schiff base ligands (**L162a-d**) to Stille coupling reactions (Scheme 108). The Pd complexes of the β -iminophosphine **L162d** and its corresponding metallic dendrimer **C41d** are better catalysts than the other complexes. In addition, catalyst **C41d** can be easily recycled at least three times with a slight decrease in yields.

An air and moisture-stable monodentate **C37c-d** and **C42a-c** and bidentate **C42d-f** Pd(II) phosphine complexes were applied for the Stille coupling of benzyl bromide and *Z*-organostannane in toluene at 60 °C (Scheme 109) [200]. The complex **C37c** exhibited the best result with 99% formation of the desired product in 1.5 h. Thus, **C37c** was shown to be an efficient catalyst for the Stille coupling reactions of a range of allylic and benzylic halides with different vinylstannanes.

Scrivanti *et al.* [201] studied the role of olefin in catalyst stabilization by using [η^2 -(olefin)Pd(0)] complexes containing iminophosphine ligands (**C32b**, and **C32e-g**) in the Stille coupling reactions. The catalytic efficiency depends on the nature of the complex ([Pd(η^2 -dmfu)



Scheme 101. Pd complexes of trisubstituted PTA derivatives for the Suzuki coupling reactions.

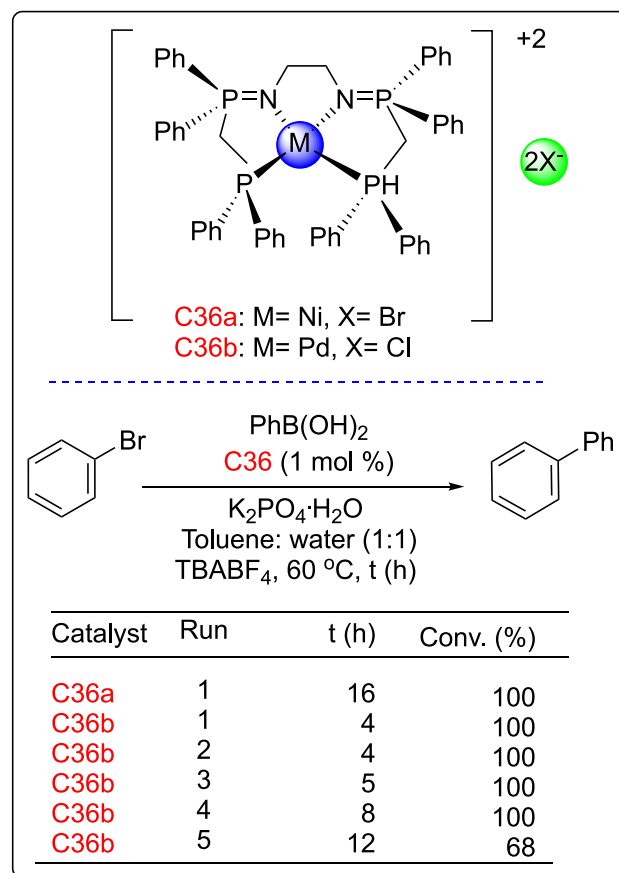
(P-N)₂] > [Pd(η²-dmfu)(P-N)], and substituent. The genuine active catalyst was Pd(0) complex (Scheme 110). The screening of catalysts and their corresponding ligands was carried out by coupling iodobenzene with tributylphenylethynyl stannane in THF at 50 °C. The highest reaction rates were achieved by using aryl-substituted iminophosphines C32b and C32e.

3.6. C-H arylation

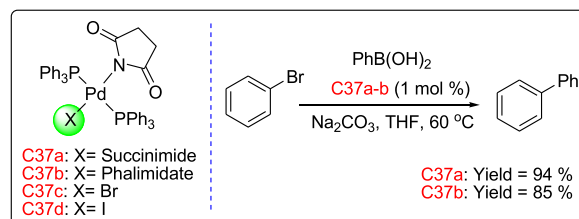
Asymmetric C-H functionalization is one of the most significant and challenging bond-forming reactions. It is used for the synthesis of a variety of functionalized molecules [202].

In this regard, the Pd-phosphine imine derived from various amino acids (L163a-l) showed excellent activity (up to 99% yield) and enantioselectivity (up to 98% *ee*) in the asymmetric arylation of cyclic N-sulfonyl imines. The reaction forms cyclic N-sulfonyl aldimines and ketimines with five and six-membered ring structures (Scheme 111) [203]. The ligand L163l (5.5 mol %) was found to be highly effective both in terms of yield and *ee* in the presence of PdCl₂(PhCN)₂ (5 mol %), AgBF₄ (5 mol %) in dichloroethane at 65 °C.

Similarly, 'SMIPHOX' ligands L27a-d (7.5 mol %) were applied for Pd-catalyzed asymmetric arylation of cyclic N-sulfonyl imines. All the



Scheme 102. Biphasic Suzuki coupling reactions using iminophosphorane ligands.

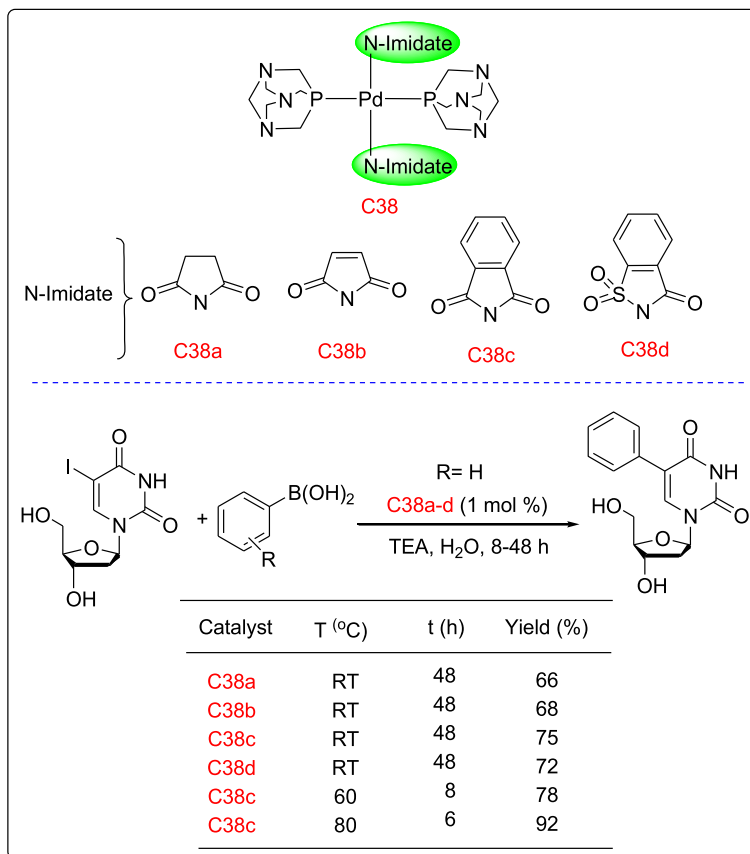


Scheme 103. Pd-imidate complexes for the Suzuki coupling reactions.

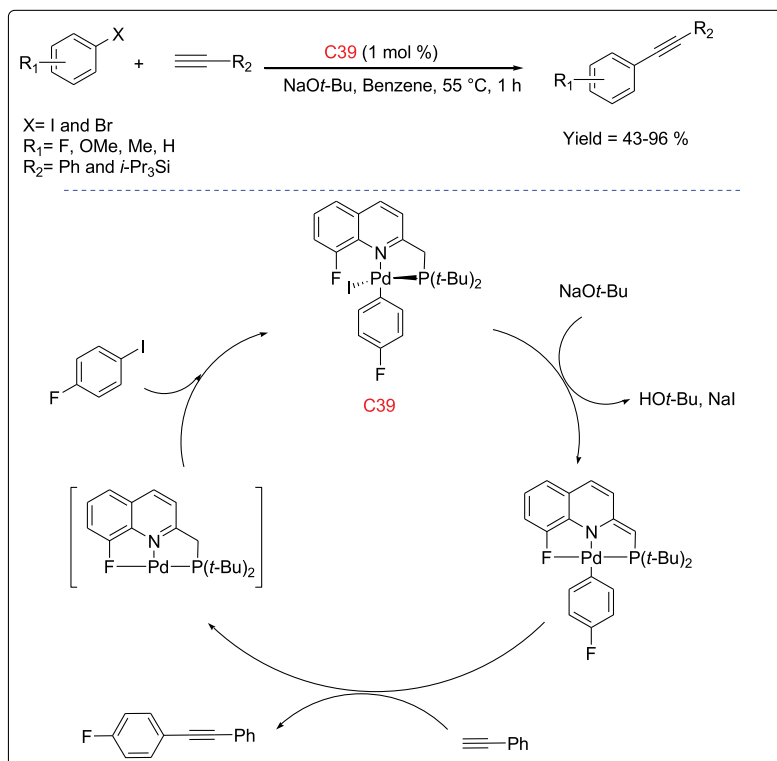
ligands were highly efficient in terms of yields (up to 99%) and enantioselectivities (up to 99% *ee*) for the reaction of cyclic N-sulfonyl aldimines and phenylboronic acid in the presence of Pd(TFA)₂ (5 mol %) in TFE at 40 °C [204].

Punji and coworkers reported an extensive mechanistic investigation of PCN pincers like C43 and C44 for C-H arylation [205]. This investigation was based on a study of the activity of pincer-based-[PdCl(*i*-Pr₂POCNEt₂)] complex (C43a) along with CuI catalyzed direct arylation of azoles involving a Pd(II)-Pd(IV)-Pd(II) redox catalytic pathway (Scheme 111). They also reported the synthesis, characterization, and applications of phosphinito aryl benzimidazole PCN pincer Pd(II) complexes (C44a-c) in C-H arylation of azoles with aryl iodides [206].

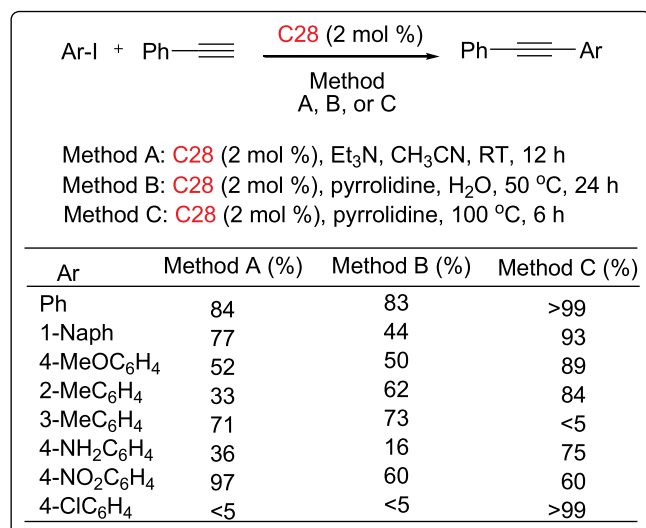
Unsymmetrical pincer Pd(II)-NHC-N-P chelated complexes C45a and C45b were synthesized and applied for C(sp²)-H arylation of benzoxazoles (Scheme 112) [207]. The catalytic system offers the direct arylation of aryl bromides with a catalyst loading as low as 0.25 mol %. The



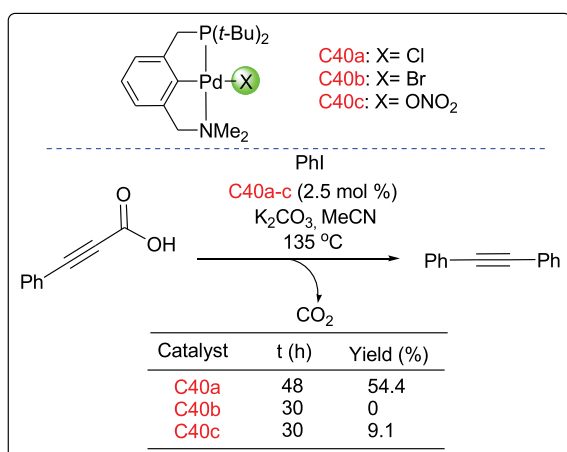
Scheme 104. Pd-imidate complexes for the synthesis of C5-arylated pyrimidine nucleosides using Suzuki coupling reactions.



Scheme 105. Pd-PNF pincer-type complex for the Sonogashira-type coupling reactions.



Scheme 106. PCN-Pd pincer for the copper-free Sonogashira coupling reactions.



Scheme 107. Pd-PCN pincers for the decarboxylative coupling reactions.

complex C45a (0.5 mol %) give up to 97% arylation products for aryl bromides.

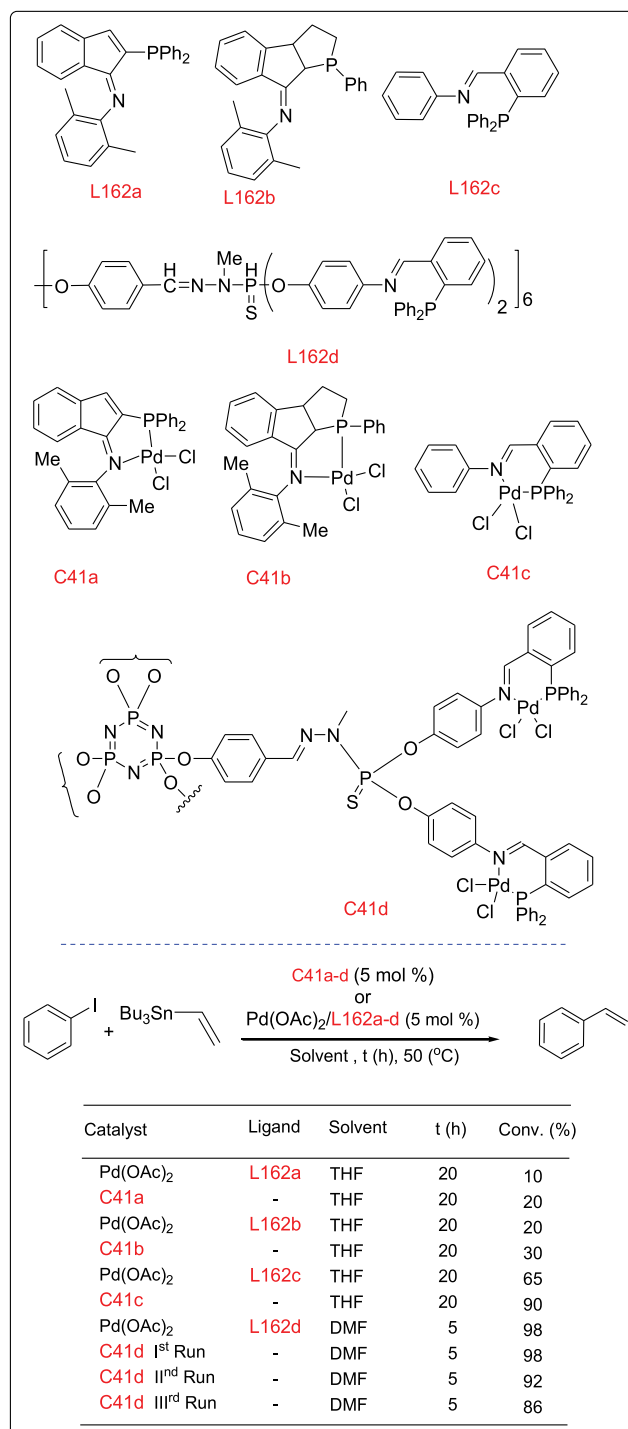
3.7. Kumada coupling reactions

Kumada coupling reaction is very important in organic synthesis [208]. In this regard, Aoyama et al. [209] studied the activity of ligands L57b-c, L58b, L58d-j, and L164a-c in the asymmetric Kumada-Corriu cross-coupling reaction (Scheme 113). Optimization of various reaction conditions for the reaction of 1-phenyl ethyl magnesium chloride with β-bromostyrene showed that the piperidine-based ligand L164c possessing the pendant piperidinyl group gave better results (58% yield, 66% ee) in the presence of PdCl₂(MeCN)₂ (5 mol %) in CF₃C₆H₅ at 0 °C.

3.8. Hiyama coupling reactions

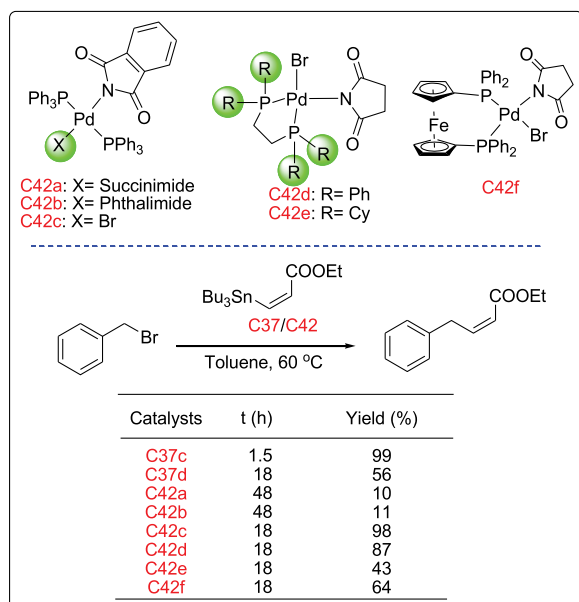
Since the last decade, Hiyama cross-coupling reactions have attracted much more attention in metal catalysis [210].

In this regard, the PCN-Pd pincer complex (C28) was tested for the Hiyama coupling reactions in aqueous media [179]. The complex was

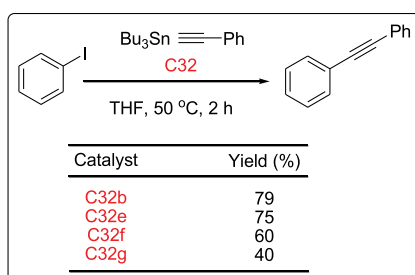


Scheme 108. Pd complexes of β and γ-iminophosphines for the Stille coupling reactions.

less active in the Hiyama coupling reaction of (trimethoxysilyl)benzene and different aryl bromides under different conditions (Method A: C28 (2 mol %) NaOH, H₂O, 140 °C, 3 h; Method B: C28 (4 mol %) TBAF, o-xylene, 80 °C, 4 h). Similarly, the Pd(OAc)₂/L157 (2:8 mol %) was also efficient for Hiyama coupling aryl and heteroaryl mesylates in the presence of TBAF in t-BuOH at 90 °C [211]. Sarkar et al. [189] applied pyrazole-derived bidentate ligands to the Hiyama coupling reactions of the aryl bromides and chlorides. The pyrrole (L160g-j) derived ligands containing arylcyclohexylphosphino or arylidiphenylphosphino donor



Scheme 109. Pd-imidate complexes for the Stille coupling reactions.



Scheme 110. The iminophosphine ligands for the Stille coupling reactions.

group showed moderate activity in Hiyama coupling reactions (Scheme 114).

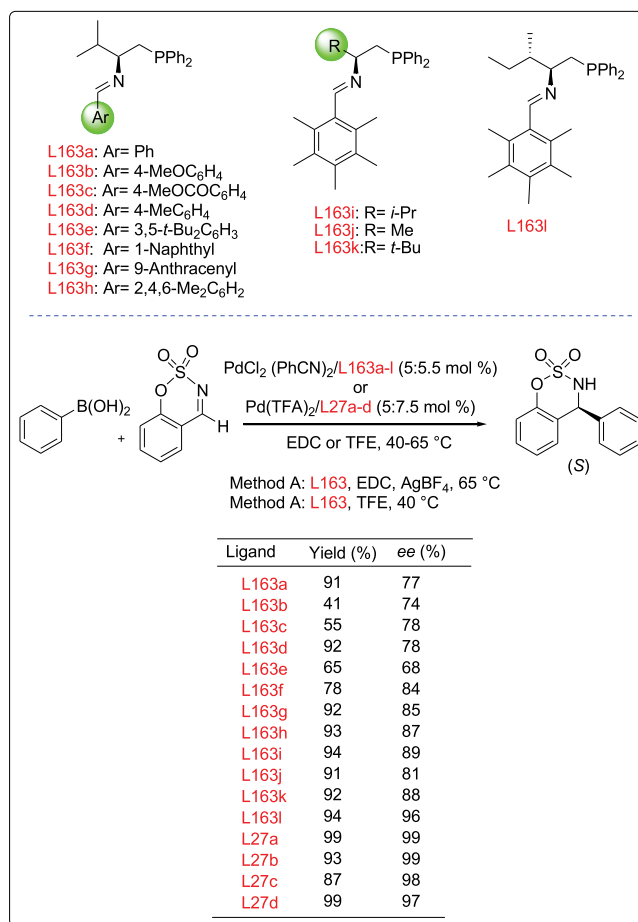
The ligand (**L158a-b**) (0.1 mol %) was first time applied for Pd-catalyzed titanium-mediated cross-coupling of aryl halides (Br and Cl) in the presence of Pd(OAc)₂ (0.05 mol %) in toluene at 90–110 °C [212].

3.9. Buchwald–Hartwig amination reactions

The Pd-catalyzed amination reaction of aryl halides by using various ligands has attracted much attention for the synthesis of substituted amines [213].

In this regard, Lassaletta et al. [214] applied various Quinap ligands (**L42a-g**) in the Pd-catalyzed dynamic kinetic asymmetric Buchwald–Hartwig amination reaction (Scheme 115). The catalytic system Pd₂(dba)₃/**L42a** (10:12 mol %) showed a high level of enantioselectivity (87–92% *ee*) with a 99% yield for different substrates in toluene at 50 or 60 °C. Other ligands **L42b-g** containing modified diaryl and dialkyl phosphine groups led to no improved enantioselectivities. In addition, they [215] applied a similar type of catalytic system for dynamic kinetic Pd-catalyzed alkynylation of racemic heterobiaryl nonaflates with excellent enantioselectivity. The Pd(OAc)₂/**L42a** and **L42d** (5:6 mol %) precatalysts provided products in excellent yields (> 99%) and enantioselectivity (93–97%) using TEA as a base in DMSO at 40 °C.

Kocovsky and coworkers reported Pd(II) complexes of 2-dimethylamino-2'-diphenylphosphino-1,1'-binaphthyl (MAP) (**R**) **L165** in allylic substitution, Buchwald amination, and Suzuki coupling reactions



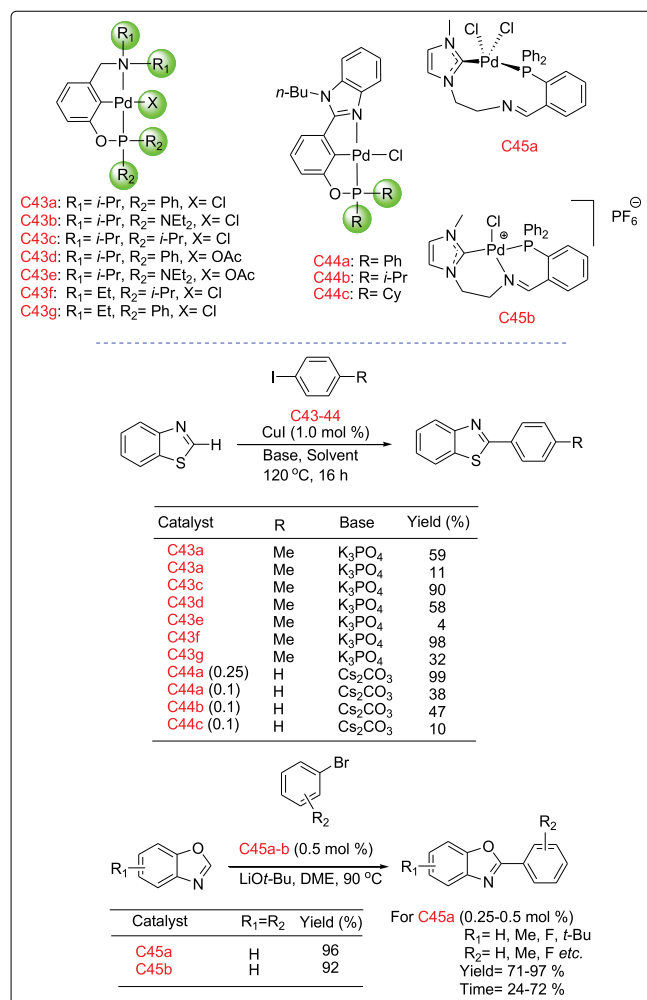
Scheme 111. P–N ligands used for asymmetric arylation of cyclic N-sulfonyl imines.

(Scheme 116). The ligand exhibited a dramatic accelerating effect in the amination reaction of amino alcohol and diamine with bromobenzene in toluene at 60 °C [216]. These ligands were assumed to coordinate Pd via P, N-chelation however, it was evidenced that (*S*) **L165** acts as a P, C ligand with an unusual Cσ–Pd bonding mode [217]. Moreover, a substantial rate acceleration was observed for the *m* (PhCN)₂PdCl₂/*(S)* **L165** (3 mol %) in the coupling reaction of 4-*t*-BuC₆H₄Br with *n*-Bu₂NH at 50 °C. The phenylation of 4-Cl-C₆H₄CHO with phenyl boronic acid also occurred in < 20 h at room temperature in the presence of Pd(OAc)₂/*(S)* **L165** (3:4.5 mol %) in CsF or Cs₂CO₃.

P–N phosphinobenzimidazole ligands (**L166a-d**) were used in Pd-catalyzed C–N coupling reactions (Scheme 117) [218]. The ligand contains a sterically large and exceptionally powerful *P*(*t*-Bu)₂ as a donor group attached at the 2-position of a benzimidazole backbone, which interacts dynamically with the Pd center under conditions of catalysis. A precatalytic system was designed from [Pd(cinnamyl)Cl]₂ and **L166a** (1:2 mol %) and proved to be effective for the reaction of aryl bromides with aryl amines using KO*t*-Bu as a base in toluene at 80 °C.

Hong et al. [219] carried out C–N and C–C coupling reactions using the alkyne bridged dicobalt phosphine ligands (**L167**) and their Pd complexes (**C46**) (Scheme 118). The complex **C46c** was characterized by spectroscopic as well as single-crystal X-ray diffraction techniques. The catalyst **C46b** lead to excellent results for the reaction of bromobenzene with morpholine in the presence of NaO*t*-B in toluene at 80 °C.

Hii et al. [220] reported that the incorporation of a hemilabile amino group containing a bulky, and electron-rich di-*t*-butyl phosphine aminophosphine ligand (**L168**) reversed the order of amination reactivity of



Scheme 112. Pd(II)-NHC-N-P chelated complexes for C(sp²)-H arylation of benzoxazoles.

aryl bromides (Scheme 119). Compared with L168, the dpfp ligand was highly ineffective under the optimized reaction conditions.

3.10. Asymmetric addition reactions

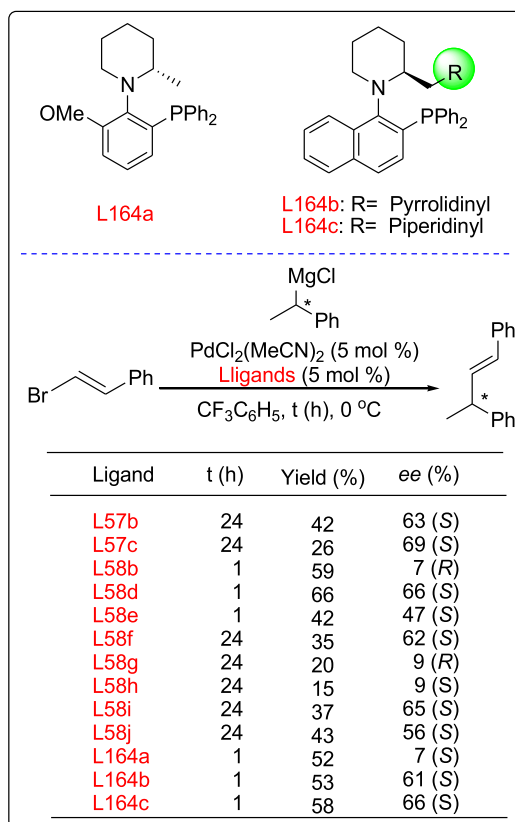
The unsymmetrical chiral PCN pincer Pd (II) complexes (C26a-e) with aryl-based aminophosphineImidazoline ligands were applied for asymmetric addition of diaryl phosphines to β -aryl enones, (Scheme 120) [221]. The products were obtained in high yields (up to 99%) and enantioselectivities (94% ee).

Shi et al. [222] reported diethylzinc-mediated asymmetric umpolung allylation of aldehydes to homoallylic alcohols using chiral phosphine-Schiff base type ligands (L169a-f) (0.02 mmol) in the presence of Pd(OAc)₂ (0.01 mmol), and Et₂Zn in THF at room temperature (Scheme 121). Moderate enantioselectivities and high syn diastereoselectivities were achieved using ligand L169c under mild conditions.

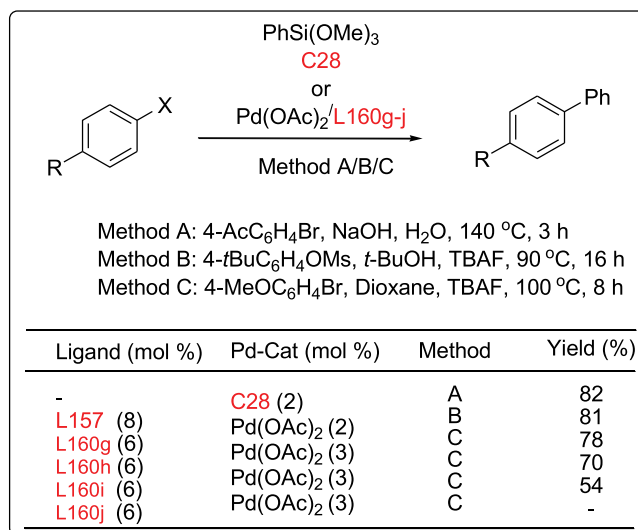
The enantiopure tetra-*ortho*-substituted biphenyl phosphinoimine ligands (*Ra,S*)-3 and (*Sa,S*)-3 (L170a-d) ligands (10 mol %) provided 3-aryl-3-hydroxyoxindoles with moderate yields and enantioselectivity via Pd-catalyzed asymmetric addition of aryl boronic acids to *N*-benzylisatin in the presence of Pd(OAc)₂ (5 mol %) and BF₃·Et₂O in THF at room temperature (Scheme 122) [223].

4. Conclusion

The present part of the review offers a critical overview of the utility

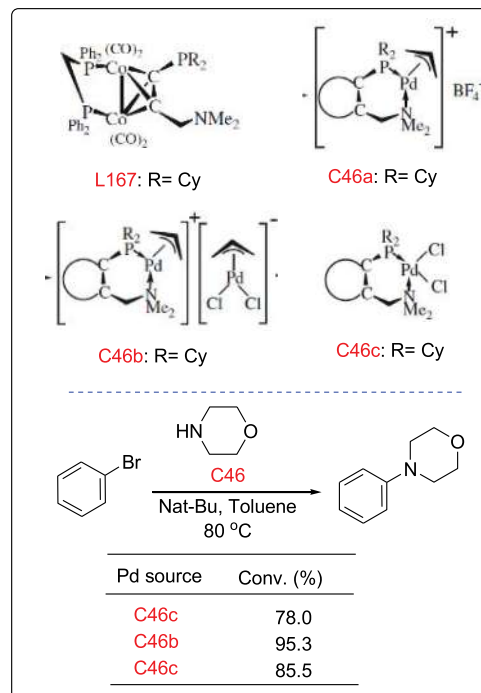
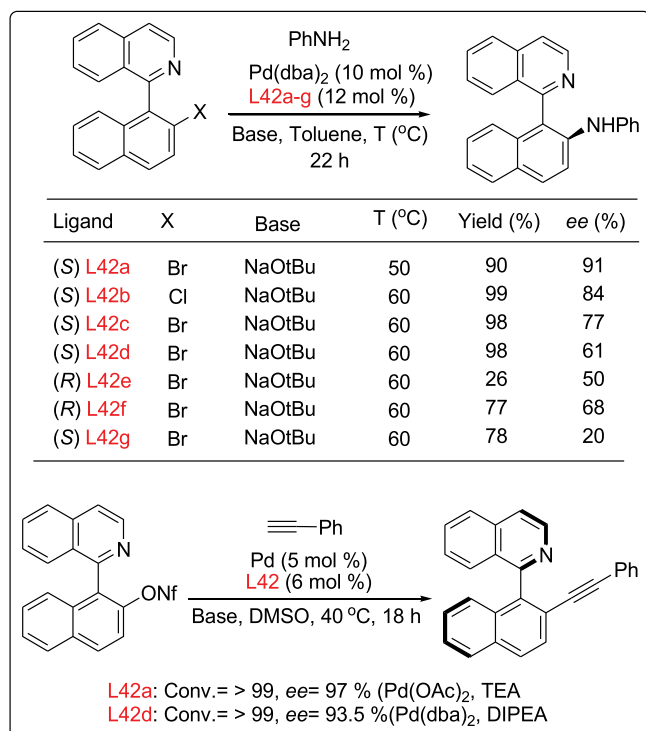


Scheme 113. Asymmetric Kumada coupling reactions using piperidine-based ligands.



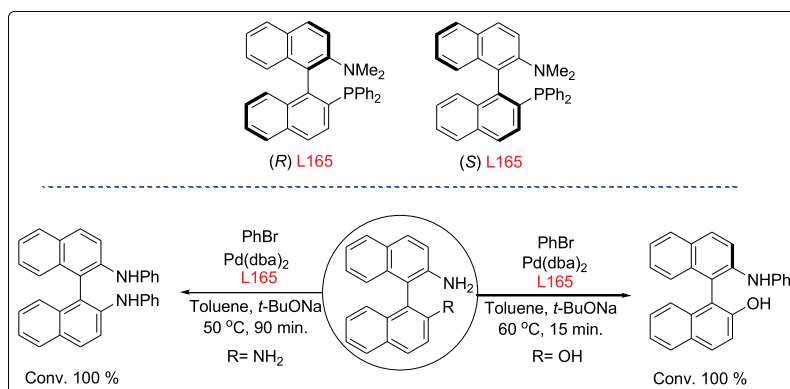
Scheme 114. P-N ligands for the Hiyama coupling reactions.

of the various strategies used in the activation and stabilization of Pd by the different P-N-based ligands. Most of the ligands are chiral scaffolds and have wide applications in Pd-catalyzed asymmetric alkylation reactions. Furthermore, distinctly different characteristics like a 'soft' P-ligand as a π -acceptor and a 'hard' N-ligand as a σ -donor provides extraordinary opportunities for cross-coupling strategies, especially in asymmetric catalysis. Most significantly, P-N ligands permit to carry out the reactions under mild reaction conditions. These ligands and catalytic systems showed broad substrate scope, supreme catalytic efficiency, and

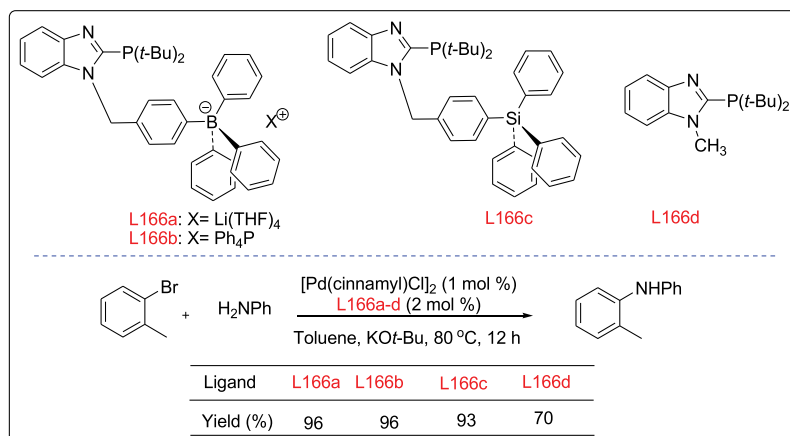


Scheme 115. Dynamic kinetic asymmetric Buchwald-Hartwig amination and alkynylation reactions using QUINAP ligands.

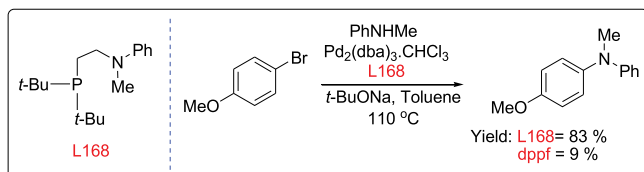
Scheme 118. Alkyne bridged dicobalt phosphine ligands for amination reactions.



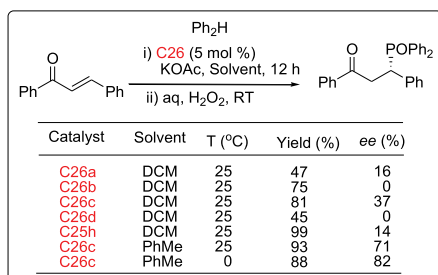
Scheme 116. MAP ligands for the Buchwald amination reactions.



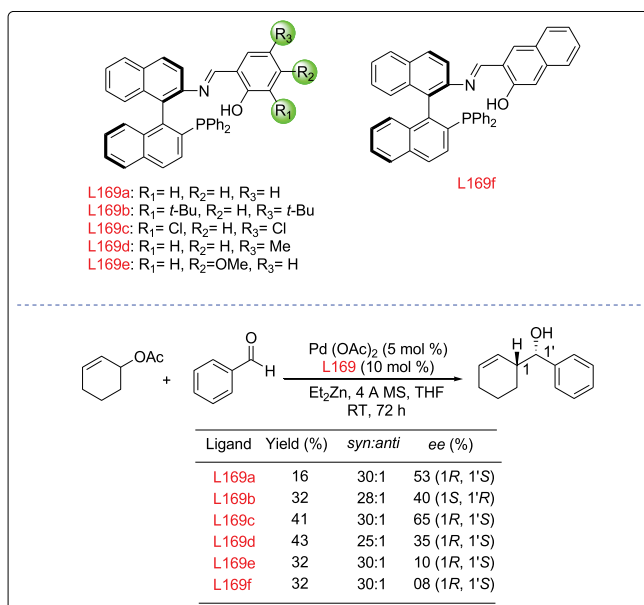
Scheme 117. P-N phosphinobenzimidazole ligands in C-N coupling reactions.



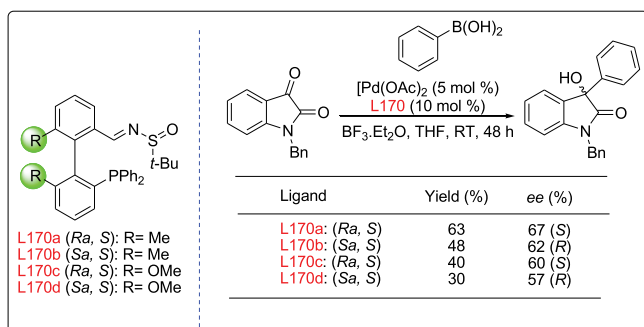
Scheme 119. Hemilabile amino group-containing bulky, electron-rich phosphorus ligand for amination reactions.



Scheme 120. Chiral Pd(II)-PCN pincer complexes for asymmetric addition of diaryl phosphines to enones.



Scheme 121. Asymmetric umpolung allylation of aldehydes using chiral phosphine-Schiff base ligands.



Scheme 122. Enantiopure tetra-ortho-substituted biphenyl phosphinoimines for asymmetric addition of aryl boronic acids to N-benzylisatin.

excellent enantioselectivities. In addition, commercially available natural amino acids have been utilized for the synthesis of most of these ligands. The P-N-based ligand systems have been continuously employed in industrial processes also. Indeed, most catalysts showed excellent catalytic activities and stereoselectivities with easy recovery and good recyclability. The design and development of new P-N ligands are likely to facilitate further progress in the field, and there is broad scope in these fields. A new design of these ligands for the range of substrates is to be studied. In forthcoming years, this area of catalysis will lead to the synthesis of many important biomolecules having wide applications in the field of theoretical and pharmaceutical chemistry.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

No data was used for the research described in the article.

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References

- [1] (a) I.P. Beletskaya, A.V. Chepurov, *Chem. Rev.* 100 (2000) 3009–3066; (b) N. Miyaura, A. Suzuki, *Chem. Rev.* 95 (1995) 2457–2483; (c) R. Chinchilla, C. Najera, *Chem. Rev.* 107 (2007) 874–922; (d) A. Molnar, *Chem. Rev.* 111 (2011) 2251–2320; (e) A. Balanta, C. Godard, C. Claver, *Chem. Soc. Rev.* 40 (2011) 4973–4985; (f) P. Ruiz-Castillo, S.L. Buchwald, *Chem. Rev.* 116 (2016) 12564–12649; (g) T. Zhou, M. Szostak, *Cat. Sci. Technol.* 10 (2020) 5702–5739.
- [2] (a) A.R. Muci, S.L. Buchwald, *Top. Curr. Chem.* 219 (2002) 131–209; (b) A.B. Dounay, L.E. Overman, *Chem. Rev.* 103 (2003) 2945–2964; (c) C.A. Busacca, D.R. Fendrick, J.J. Song, C.H. Senanayake, *Adv. Synth. Catal.* 353 (2011) 1825–1864.
- [3] (a) N.E. Leadbeater, M. Marco, *Chem. Rev.* 102 (2002) 3217–3274; (b) V. Polshettiwar, C. Len, A. Fihri, *Coord. Chem. Rev.* 253 (2009) 2599–2626; (c) A. Kumbhar, R. Salunkhe, *Curr. Org. Chem.* 19 (2015) 2075–2121; (d) A. Kumbhar, *Top. Curr. Chem.* (Z) 375 (2017) 2–28; (e) S.N. Jadhav, A.S. Kumbhar, C.V. Rode, R.S. Salunkhe, *Green Chem.* 18 (2016) 1898–1911.
- [4] (a) R. Zhong, A.C. Lindhorst, F.J. Groche, F.E. Kühn, *Chem. Rev.* 117 (2017) 1970–2058; (b) S. Jadhav, A. Kumbhar, S. Mali, C. Hong, R. Salunkhe, *New J. Chem.* 39 (2015) 2333–2341; (c) S. Shi, S.P. Nolan, M. Szostak, *Acc. Chem. Res.* 51 (2018) 2589–2599; (d) Q. Zhao, G. Meng, S.P. Nolan, M. Szostak, *Chem. Rev.* 120 (2020) 1981–2048; (e) C. Chen, F.S. Liu, M. Szostak, *Chem. Eur. J.* 27 (2021) 4478–4499.
- [5] A. Kumbhar, *J. Organomet. Chem.* 881 (2019) 79–129.
- [6] A. Kumbhar, *J. Organomet. Chem.* 848 (2017) 22–88.
- [7] A. Rajmane, S. Jadhav, A. Kumbhar, *J. Organomet. Chem.* 957 (2022), 122147.
- [8] A. Gillespie, E. Zuidema, P.W.N.M. van Leeuwen, P.C.J. Kamer, *Phosphorus(III) Ligands in Homogeneous Catalysis: Design and Synthesis*, John Wiley & Sons, Ltd., Chichester, 2012, pp. 1–22. P. C. J. Kamer and P. W. N. M. Van Leeuwen.
- [9] M.P. Mitoraj, A. Michalak, *Inorg. Chem.* 49 (2010) 578–582.
- [10] (a) C.A. Tolman, *Chem. Rev.* 77 (1977) 313–348; (b) J. Jover, J. Cirera, *Dalton Trans.* 48 (2019) 15036–15048.
- [11] P. Espinet, K. Soultanica, *Coord. Chem. Rev.* 195 (1999) 499–556.
- [12] (a) X. Liu, L. Lin, X. Feng, *Org. Chem. Front.* 1 (2014) 298–302; (b) E.N. Jacobsen, A. Pfaltz, H. Yamamoto (Eds.), *Comprehensive Asymmetric Catalysis*, Springer, New York, 1999; (c) Q.-L. Zhou (Ed.), *Privileged Chiral Ligands and Catalysts*, Wiley-VCH, Weinheim, 2011.

- [13] (a) P.W.N.M. van Leeuwen, P.C.J. Kamer, C. Claver, O. Pamies, M. Dieguez, *Chem. Rev.* 111 (2011) 2077–2118;
(b) P. Stepnicka, *Chem. Soc. Rev.* 41 (2012) 4273–4305.
- [14] (a) J. Margalef, M. Biosca, P. Sánchez, J. Faiges, O. Pàmies, M. Diéguez, *Coord. Chem. Rev.* 446 (2021), 214120;
(b) B.V. Rokade, P.J. Guiry, *ACS Catal.* 8 (2018) 624–643;
(c) J. Takaya, *Chem. Sci.* 12 (2021) 1964–1981.
- [15] (a) T. Zhou, S. Ma, F. Nahra, A.M.C. Obled, A. Poater, L. Cavallo, C.S.J. Cazin, S. P. Nolan, M. Szostak, *iScience* 23 (2020), 101377;
(b) G. Li, T. Zhou, A. Poater, L. Cavallo, S.P. Nolan, M. Szostak, *Catal. Sci. Technol.* 10 (2020) 710–716;
(c) M.C. D'Alterio, È. Casals-Cruañas, N.V. Tzouras, G. Talarico, S.P. Nolan, A. Poater, *Chem. Eur. J.* 27 (2021) 13481–13493;
(d) N. Fey, J.M. Lynam, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.* (2021) e1590, <https://doi.org/10.1002/wcms.1590>.
- [16] (a) B.M. Trost, D.L. Van Vranken, *Chem. Rev.* 96 (1996) 395–422;
(b) B.M. Trost, *Tetrahedron* 71 (2015) 5708–5733;
(c) S. Noreen, A.F. Zahoor, S. Ahmad, I. Shahzadi, A. Irfan, S. Faiz, *Curr. Org. Chem.* 23 (2019) 1168–1213;
(d) A. Esposito, C. Di Giovanni, M. Fenza, G. Talarico, M. Chino, G. Palumbo, A. Guaragna, D. D'Alonzo, *Chem. A Eur. J.* 26 (2020) 2597–2601.
- [17] J. Tsuji, H. Takahashi, M. Morikawa, *Tetrahedron Lett.* 6 (1965) 4387–4388.
- [18] B.M. Trost, T.J. Fullerton, *J. Am. Chem. Soc.* 95 (1973) 292–294.
- [19] (a) B.M. Trost, W. Tang, F.D. Toste, *J. Am. Chem. Soc.* 127 (2005) 14785–14803;
(b) D.J. Janssen, R.A. Shenvi, *J. Am. Chem. Soc.* 135 (2013) 1209–1212.
- [20] (a) G. Helmchen, A. Pfaltz, *Acc. Chem. Res.* 33 (2000) 336–345;
(b) A. Helen, M. Manus, P.J. Guiry, *Chem. Rev.* 104 (2004) 4151–4202;
(c) G.C. Hargaden, P.J. Guiry, *Chem. Rev.* 109 (2009) 2505–2550.
- [21] O. Reiser, *Angew. Chem. Int. Ed. Engl.* 32 (1993) 547–549.
- [22] (a) J. Spring, G. Helmchen, *Tet. Lett.* 34 (1993) 1769–1772;
(b) P. Matt, A. Pfltz, *Angew. Chem. Int. Ed.* 32 (1993) 566–568;
(c) G.J. Dawson, C.G. Frost, J.M.J. Williams, *Tetrahedron* 34 (1993) 3149–3150.
- [23] (a) I.C. Baldwin, J.M.J. Williams, *Tetrahedron: Asymmetry* 6 (1995) 679–682;
(b) R.N. Constantine, N. Kim, R.C. Bunt, *Org. Lett.* 5 (2003) 2279–2282.
- [24] D. Liu, Q. Dai, X. Zhang, *Tetrahedron* 61 (2005) 6460–6471.
- [25] O. Hoarau, H. Haddou, J. Daran, D. Cramailere, G.G.A. Balavoine, *Organometallics* 18 (1999) 4718–4723.
- [26] H. Haddou, O. Hoarau, D. Cramailere, F. Pezet, J. Daran, G.G.A. Balavoine, *Chem. Eur. J.* 10 (2004) 699–707.
- [27] K. Nordstrom, E. Macedo, C. Moberg, *J. Org. Chem.* 62 (1997) 1604–1609.
- [28] A. Frolander, S. Lutsenko, T. Privalov, C. Moberg, *J. Org. Chem.* 70 (2005) 9882–9891.
- [29] A.M. Porte, J. Reibenspies, K. Burgess, *J. Am. Chem. Soc.* 120 (1998) 9180–9187.
- [30] D.R. Hou, K. Burgess, *Org. Lett.* 11 (1999) 1745–1747.
- [31] D.R. Hou, J.H. Reibenspies, K. Burgess, *J. Org. Chem.* 66 (2001) 206–215.
- [32] G. Jones, C.J. Richards, *Tetrahedron Lett.* 42 (2001) 5553–5555.
- [33] S.R. Gilbertson, C.W.T. Chang, *J. Org. Chem.* 63 (1998) 8424–8431.
- [34] S. Woodward, M. Diéguez, O. Pàmies, *Coord. Chem. Rev.* 254 (2010) 2007–2030.
- [35] M. Diéguez, O. Pàmies, A. Ruiz, Y. Diaz, S. Castillón, C. Claver, *Coord. Chem. Rev.* 248 (2004) 2165–2192.
- [36] B. Gläser, H. Kunz, *Synlett* 1 (1998) 53–54.
- [37] H. Kunz, A. Harreus, *Liebigs Ann. Chem.* (1982) 41–48.
- [38] (a) K. Yonehara, T. Hashizume, K. Mori, K. Ohe, S. Uemura, *J. Org. Chem.* 64 (1999) 9374–9380;
(b) K. Yonehara, T. Hashizume, K. Mori, K. Ohe, S. Uemura, *Chem. Commun.* (1999) 415–416.
- [39] T. Hashizume, K. Yonehara, K. Ohe, S. Uemura, *J. Org. Chem.* 65 (2000) 5197–5201.
- [40] (a) Y. Mata, M. Dieguez, O. Pamies, C. Claver, *Adv. Synth. Catal.* 347 (2005) 1943–1947;
(b) Y. Mata, O. Pamies, M. Dieguez, *Adv. Synth. Catal.* 351 (2009) 3217–3234.
- [41] G.J.H. Buisman, P.C.J. Kamer, P.W.N.M. van Leeuwen, *Tetrahedron: Asymmetry* 4 (1993) 1625–1634.
- [42] (a) J. Kraft, T. Ziegler, *Carbohydr. Res.* 411 (2015) 56–63;
(b) M.R. Imrich, J. Kraft, C. Maichle-Mössmer, T. Ziegler, *Beilstein J. Org. Chem.* 14 (2018) 2082–2089;
(c) M.R. Imrich, C. Maichle-Mössmer, T. Ziegler, *Eur. J. Org. Chem.* 2019 (2019) 3955–3963.
- [43] (a) L.F. Tietze, J.K. Lohmann, *Synlett* 12 (2002) 2083–2085;
(b) N. End, C. Stoessel, U. Berens, P. di Pietro, P.G. Cozzi, *Tetrahedron: Asymmetry* 15 (2004) 2235–2239.
- [44] A. Sudo, K. Saigo, *J. Org. Chem.* 62 (1997) 5508–5513.
- [45] Y. Wang, A. Hämäläinen, J. Tois, R. Franzén, *Tetrahedron: Asymmetry* 21 (2010) 2376–2384.
- [46] Y. Wang, M.J.P. Vaismaa, A.M. Hämäläinen, J.E. Tois, R. Franzén, *Tetrahedron: Asymmetry* 22 (2011) 524–529.
- [47] (a) C. Blanc, J. Hannedouche, F. Agbossou-Niedercorn, *Tetrahedron Lett.* 44 (2003) 6469–6473;
(b) P. Giorgio Cozzi, N. Zimmermann, R. Hilgraf, S. Schaffner, A. Pfaltz, *Adv. Synth. Catal.* 343 (2001) 450–454.
- [48] (a) X.H. Yang, J.H. Xie, Q.L. Zhou, *Org. Chem. Front.* 1 (2014) 190–193;
(b) X. Liu, Z. Han, Z. Wang, K. Ding, *Angew. Chem., Int. Ed.* 53 (2014) 1978–1982;
(c) W.Y. Sun, H.R. Gu, X.F. Lin, *J. Org. Chem.* 83 (2018) 4034–4043.
- [49] (a) Z. Qiu, R. Suna, D. Teng, *Org. Biomol. Chem.* 16 (2018) 7717–7724;
(b) Y. Gao, Z. Qiu, R. Sun, N. Gao, G. Cao, D. Teng, *Tetrahedron Lett* 59 (2018) 3938–3941.
- [50] (a) Z. Qiu, R. Sun, K. Yang, D. Teng, *Molecules* 24 (2019) 1575–1586;
(b) S. Li, J. Zhang, H. Li, L. Feng, P. Jiao, *J. Org. Chem.* 84 (2019) 9460–9473.
- [51] S. Trudeau, J.P. Morken, *Tetrahedron* 62 (2006) 11470–11476.
- [52] J. Padevè, M.G. Schrems, R. Scheil, A. Pfaltz, *Beilstein J. Org. Chem.* 12 (2016) 1185–1195.
- [53] Y. Imai, W. Zhang, T. Kida, Y. Nakatsuji, I. Ikeda, *Tetrahedron Lett.* 39 (1998) 4343–4346.
- [54] M. Ogasawara, K. Yoshida, H. Kamei, K. Kato, Y. Uozumi, T. Hayashi, *Tetrahedron: Asymmetry* 9 (1998) 1779–1787.
- [55] W. Zhang, F. Xie, H. Yoshinaga, T. Kida, Y. Nakatsuji, I. Ikeda, *Synlett* 8 (2006) 1185–1188.
- [56] K.N. Gavrilov, S.V. Zheglov, I.M. Novikov, V.V. Lugovsky, V.S. Zimarev, I. S. Mikhail, *Tetrahedron: Asymmetry* 27 (2016) 1260–1268.
- [57] (a) S. Lee, C.W. Lim, C.E. Song, K.M. Kim, C.H. Jun, *J. Org. Chem.* 64 (1999) 4445–4451;
(b) T. Yamagishi, M. Ohnuki, T. Kiyooka, D. Masui, K. Sato, M. Yamaguchi, *Tetrahedron: Asymmetry* 14 (2003) 3275–3279.
- [58] (a) M.R. Castillo, S. Castillón, C. Claver, J.M. Fraile, A. Gual, M. Martín, J. A. Mayoral, E. Solac, *Tetrahedron* 67 (2011) 5402–5408;
(b) R. Bellini, M. Magre, M. Biosca, P. Norrby, O. Pamies, M. Dieguez, C. Moberg, *ACS Catal.* 6 (2016) 1701–1712.
- [59] Y. Okuyama, H. Nakano, H. Hongo, *Tetrahedron: Asymmetry* 11 (2000) 1193–1198.
- [60] (a) M.J. Jin, J.A. Jung, S.H. Kim, *Tetrahedron Lett.* 40 (1999) 5197–5198;
(b) M.J. Jin, M.S. Sarkar, J.Y. Jung, D.H. Lee, I.M. Lee, *Bull. Korean Chem. Soc.* 27 (2006) 773.
- [61] G. Helmchen, A. Pfaltz, *Acc. Chem. Res.* 33 (2000) 336–345.
- [62] (a) V. Fuente, R. Marcos, X.C. Cambeiro, S. Castillon, C. Claver, M.A. Perics, *Adv. Synth. Catal.* 353 (2011) 3255–3261;
(b) V. Fuente, N. Fleury-Brégeot, S. Castillón, C. Claver, *Green Chem.* 14 (2012) 2715–2718.
- [63] L. Mei, Z. Yuan, M. Shi, *Organometallics* 30 (2011) 6466–6475.
- [64] (a) B.V. Rokade, P.J. Guiry, *ACS Catal.* 8 (2018) 624–643;
(b) E. Fernández, P.J. Guiry, K.P.T. Connole, J.M. Brown, *J. Org. Chem.* 79 (2014) 5391–5400.
- [65] (a) S. Wu, S. Xiang, J.K. Cheng, B. Tan, *Tetrahedron Chem* 1 (2022), 100009;
(b) G.-J. Mei, W.L. Koay, C.-Y. Guan, Y. Lu, *Chem* (2022), <https://doi.org/10.1016/j.chempr.2022.04.011>;
(c) X. Min, X. Zhang, R. Shen, Q. Zhang, Y. He, *Org. Chem. Front.* 9 (2022) 2280–2292.
- [66] (a) N.W. Alcock, J.M. Brown, M. Pearson, S. Woodward, *Tetrahedron: Asymmetry* 3 (1992) 17–20;
(b) W. Alcock, J.M. Brown, D.I. Hulmes, *Tetrahedron: Asymmetry* 4 (1993) 743–756.
- [67] J.M. Brown, D.I. Hulmes, P.J. Guiry, *Tetrahedron* 50 (1994) 4493–4506.
- [68] J.M. Valk, T.D.W. Claridge, J.M. Brown, D. Hibbs, M.B. Hursthouse, *Tetrahedron: Asymmetry* 6 (1995) 2597–2610.
- [69] M. McCarthy, R. Goddard, P.J. Guiry, *Tetrahedron: Asymmetry* 10 (1999) 2797–2807.
- [70] D. Cai, J.F. Payback, D.R. Bender, D.L. Hughes, T.R. Verhoeven, P.J. Rider, *J. Org. Chem.* 59 (1994) 7180–7181.
- [71] S.P. Flanagan, R. Goddard, P.J. Guiry, *Tetrahedron* 61 (2005) 9808–9821.
- [72] T. Fekner, H. Müller-Bunz, P.J. Guiry, *Eur. J. Org. Chem.* (2008) 5055–5066.
- [73] W.J. Fleming, H. Müller-Bunz, P.J. Guiry, *Eur. J. Org. Chem.* (2010) 5996–6004.
- [74] T. Fekner, H. Müller-Bunz, P. Guiry, *J. Org. Lett.* 8 (2006) 5109–5112.
- [75] P.A. Evans, T.A. Brandt, *Tetrahedron Lett.* 31 (1996) 9143–9146.
- [76] S.Hata Mino, K. Ohtaka, M. Sakamoto, T. Fujita, *Tetrahedron Lett.* 42 (2001) 4837–4839.
- [77] R. Lazny, A. Nodzevska, *Chem. Rev.* 110 (2010) 1386–1434.
- [78] (a) T. Mino, W. Imiya, M. Yamashita, *Synlett* (1997) 583–584;
(b) T. Mino, M. Shiotsuki, N. Yamamoto, T. Suenaga, M. Sakamoto, T. Fujita, M. Yamashita, *J. Org. Chem.* 66 (2001) 1795–1797.
- [79] T. Mino, E. Komatsumoto, S. Nakadai, H. Toyoda, M. Sakamoto, T. Fujita, *J. Mol. Catal. A: Chem.* 196 (2003) 13–20.
- [80] M. Widhalm, M. Abraham, V.B. Arion, S. Sarasalu, U. Marg, *Tetrahedron: Asymmetry* 21 (2010) 1971–1982.
- [81] (a) T. Kohara, Y. Hashimoto, K. Saigo, *Synlett* 4 (2000) 517–519;
(b) J.D. Zhang, X.P. Hu, S.B. Yu, J. Deng, D.Y. Wang, Z.C. Duan, Z. Zheng, *J. Mol. Catal. A: Chem.* 270 (2007) 127–131.
- [82] X. Hu, C. Bai, H. Dai, H. Chen, Z. Zheng, *J. Mol. Catal. A: Chem.* 218 (2004) 107–112.
- [83] A. Saitoh, T. Morimoto, K. Achiwa, *Tetrahedron: Asymmetry* 8 (1997) 3567–3570.
- [84] A. Saitoh, K. Achiwa, T. Morimoto, *Tetrahedron: Asymmetry* 9 (1998) 741–744.
- [85] Y. Li, F. Liang, R. Wu, Q. Li, Q.R. Wang, Y.C. Xu, L. Jiang, *Synlett* 23 (2012) 1805–1808.
- [86] A. Michaelis, *Ann. Chim.* 326 (1903) 129–258.
- [87] J. Gopalakrishnan, *Appl. Organometal. Chem.* 23 (2009) 291–318.
- [88] H. Kubota, K. Koga, *Tetrahedron Lett* 35 (1994) 6689–6692.
- [89] (a) P. Wimmer, M. Widhalm, *Tetrahedron: Asymmetry* 6 (1995) 657–660;
(b) H. Kubota, K. Koga, *Heterocycles* 42 (1996) 543–547.
- [90] (a) P. Wimmer, M. Widhalm, *Monatsh. Chem.* 127 (1996) 669–681;
(b) M. Bourghida, M. Widhalm, *Tetrahedron: Asymmetry* 9 (1998) 1073–1083.

- [91] T. Mino, Y. Tanaka, K. Akita, K. Anada, M. Sakamoto, T. Fujita, *Tetrahedron: Asymmetry* 12 (2001) 1677–1682.
- [92] Y. Tanaka, T. Mino, K. Akita, M. Sakamoto, T. Fujita, *J. Org. Chem.* 69 (2004) 6679–6687.
- [93] T. Mino, Y. Sato, A. Saito, Y. Tanaka, H. Saotome, M. Sakamoto, T. Fujita, *J. Org. Chem.* 70 (2005) 7979–7984.
- [94] T. Mino, Y. Tanaka, Y. Hattori, T. Yabusaki, H. Saotome, M. Sakamoto, T. Fujita, *J. Org. Chem.* 71 (2006) 7346–7353.
- [95] T. Mino, K. Wakui, S. Oishi, Y. Hattori, M. Sakamoto, T. Fujita, *Tetrahedron: Asymmetry* 19 (2008) 2711–2716.
- [96] T. Mino, S. Komatsu, K. Wakui, H. Yamada, H. Saotome, M. Sakamoto, T. Fujita, *Tetrahedron: Asymmetry* 21 (2010) 711–718.
- [97] T. Mino, M. Ishikawa, K. Nishikawa, K. Wakui, M. Sakamoto, *Tetrahedron: Asymmetry* 24 (2013) 499–504.
- [98] T. Mino, H. Yamada, S. Komatsu, M. Kasai, M. Sakamoto, T. Fujita, *Eur. J. Org. Chem.* (2011) 4540–4542.
- [99] T. Mino, M. Asakawa, Y. Shima, H. Yamada, F. Yagishita, M. Sakamoto, *Tetrahedron* 71 (2015) 5985–5993.
- [100] T. Mino, J. Youda, T. Ebisawa, Y. Shima, K. Nishikawa, Y. Yoshida, M. Sakamoto, *J. Oleo Sci.* 67 (2018) 1189–1199.
- [101] X. Dai, S. Virgil, *Tetrahedron Lett.* 40 (1999) 1245–1248.
- [102] G. Chelucci, M.A. Cabras, C. Botteghi, M. Marchetti, *Tetrahedron Asymmetry* 7 (1996) 885–889.
- [103] (a) B.M. Trost, R.C. Bunt, *J. Am. Chem. Soc.* 116 (1994) 4089–4090; (b) B.M. Trost, A.C. Krueger, R.C. Bunt, J. Zambrano, *J. Am. Chem. Soc.* 118 (1996) 6520–6521.
- [104] G. Brenchley, E. Merfield, M. Will, M. Fedotoff, *Tetrahedron Lett.* 35 (1994) 2791–2794.
- [105] X.M. Sun, M. Koizumi, K. Manabe, S. Kobayashi, *Adv. Synth. Catal.* 347 (2005) 1893–1898.
- [106] I. Filipova, G. Stavrakov, V. Dimitro, *Tetrahedron: Asymmetry* 24 (2013) 1253–1256.
- [107] (a) Z. Császár, G. Farkas, A. Bényei, G. Lendvay, I. Tóth, J. Bakos, *Dalton Trans.* 44 (2015) 16352–16360; (b) G. Farkas, Z. Császár, K. Stágel, E. Nemes, S. Balogh, I. Tóth, A. Bényei, G. Lendvay, J. Bakos, *J. Organomet. Chem.* 846 (2017) 129–140.
- [108] C.F. Vasconcelos, N.P. Rath, C.D. Spilling, *Tetrahedron: Asymmetry* 9 (1998) 937–948.
- [109] M. Tollabi, E. Framery, C. Goux-Henry, D. Sinou, *Tetrahedron: Asymmetry* 14 (2003) 3329–3333.
- [110] A. Konovets, K.G. Alexandr, P.E. Framery, P. Jubault, C. Goux-Henry, K. M. Pietrusiewicz, J.C. Quirion, D. Sinou, *Tetrahedron: Asymmetry* 16 (2005) 3183–3187.
- [111] G.E. Framery, C. Goux-Henry, K.M. Pietrusiewicz, D. Sinou, *Tetrahedron* 63 (2007) 7133–7141.
- [112] C. Borriello, M.E. Cucciolito, A. Panunzi, F. Ruffo, *Inorg. Chim. Acta* 353 (2003) 238–244.
- [113] K. Glegoia, S.A. Johannesen, L. Thim, C. Goux-Henry, T. Skrydstrup, E. Framery, *Tetrahedron Lett.* 49 (2008) 6635–6638.
- [114] I. Szulc, R. Kotoldziuk, B. Kryczka, A. Zawiszain, *Tetrahedron Lett.* 56 (2015) 4740–4743.
- [115] C. Shena, H. Xia, H. Zheng, P. Zhang, X. Chen, *Tetrahedron: Asymmetry* 21 (2010) 1936–1941.
- [116] H. Brunner, M. Schonherr, M. Zabel, *Tetrahedron: Asymmetry* 14 (2003) 1115–1122.
- [117] S.A. Johannesen, K. Glegoia, D. Sinou, E. Framery, T. Skrydstrup, *Tetrahedron Lett.* 48 (2007) 3569–3573.
- [118] P.W.N.M. van Leeuwen, P.C.J. Kamer, C. Claver, O. Pamies, M. Dieguez, *Chem. Rev.* 111 (2011) 2077–2118.
- [119] M. Lega, J. Margalef, F. Ruffo, O. Pàmies, M. Diéguez, *Tetrahedron: Asymmetry* 24 (2013) 995–1000.
- [120] M. Magre, M. Biosca, P. Norrby, O. Pàmies, M. Dieguez, *ChemCatChem* 7 (2015) 4091–4107.
- [121] J. Mazuela, O. Pamies, M. Dieguez, *Chem. Eur. J.* 19 (2013) 2416–2432.
- [122] X. Meng, Y. Gao, X. Li, D. Xu, *Catal. Commun.* 10 (2009) 950–954.
- [123] M.D.K. Boele, P.C.J. Kamer, M. Lutz, A. L. Spek, J. G. de Vries, P.W.N.M. van Leeuwen, G.P.F. van Srijdonck, *Chem. Eur. J.* 10 (2004) 6232–6246.
- [124] J.M. Brunel, T. Constantieux, A. Labande, F. Lubatti, G. Buono, *Tetrahedron Lett.* 38 (1997) 5971–5974.
- [125] T. Constantieux, J.M. Brunel, A. Labande, G. Buono, *Synlett* (1998) 49–50.
- [126] M.J. Bravo, R.M. Ceder, A. Grabulosa, G. Muller, M. Rocamora, J.C. Bayon, D. Peral, *Organometallics* 34 (2015) 3799–3808.
- [127] M.J. Bravo, R.M. Ceder, A. Grabulosa, G. Muller, M. Rocamora, M. Font-Bardia, *J. Organomet. Chem.* 830 (2017) 42–55.
- [128] Y. Uozumi, K. Shibatomi, *J. Am. Chem. Soc.* 123 (2001) 2919–2920.
- [129] Y. Uozumi, H. Tanaka, K. Shibatom, *Org. Lett.* (2004) 281–283.
- [130] Y. Uozumi, M. Kimura, *Tetrahedron: Asymmetry* 17 (2006) 161–166.
- [131] J.F. Buerger, A. Togni, *Chem. Commun.* 47 (2011) 1896–1898.
- [132] Z. Hu, Y. Li, K. Liu, Q. Shen, *J. Org. Chem.* 77 (2012) 7957–7967.
- [133] (a) R. Pretot, G.C. Lloyd-Jones, A. Pfaltz, *Pure Appl. Chem.* 70 (1998) 1035–1040; (b) R. Pretot, A. Pfaltz, *Angew. Chem. Int. Ed.* 37 (1998) 323–325.
- [134] R. Hilgraf, A. Pfaltz, *Synlett* 11 (1999) 1814–1816.
- [135] R.J. van Haaren, C.J.M. Drujiven, G.P.F. van Srijdonck, H. Oevering, J.N. H. Reek, P.C.J. Kamer, P.W.N.M. van Leeuwen, *J. Chem. Soc., Dalton Trans.* (2000) 1549–1554.
- [136] A. Marson, J.E. Ernstring, M. Lutz, A.L. Spek, P.W.N.M. van Leeuwen, P.C. J. Kamer, *Dalton Trans.* (2009) 621–633.
- [137] J.T. Mohr, B.M. Stoltz, *Chem. Asian J.* 2 (2007) 1476–1491.
- [138] D.C. Behenna, B.M. Stoltz, *J. Am. Chem. Soc.* 126 (2004) 15044–15045.
- [139] J.A. Keith, D.C. Behenna, J.T. Mohr, S. Ma, S.C. Marinescu, J. Oxgaard, B. M. Stoltz, W.A. Goddard, *J. Am. Chem. Soc.* 129 (2007) 11876–11877.
- [140] J.T. Mohr, D.C. Behenna, A.M. Harned, B.M. Stoltz, *Angew. Chem. Int. Ed.* 44 (2005) 6924–6927.
- [141] A.Y. Hong, N.B. Bennett, M.R. Krout, T. Jensen, A.M. Harned, B.M. Stoltz, *Tetrahedron* 67 (2011) 10234–10248.
- [142] E. Belanger, M.F. Pouliot, M.A. Courtemanche, J.F. Paquin, *J. Org. Chem.* 77 (2012) 317–331.
- [143] B.M. Trost, J. Xu, *J. Am. Chem. Soc.* 127 (2005) 2846–2847.
- [144] B.M. Trost, J. Xu, *J. Am. Chem. Soc.* 127 (2005) 17180–17181.
- [145] (a) B.M. Trost, R.N. Bream, J. Xu, *Angew. Chem.* 118 (2006) 3181–3184; (b) B.M. Trost, R.N. Bream, J. Xu, *Angew. Chem. Int. Ed.* 45 (2006) 3109–3112.
- [146] B.M. Trost, G.M. Schroeder, *J. Am. Chem. Soc.* 121 (1999) 6759–6760.
- [147] (a) Kevin W. Wellington, Steven A. Benner, *Nucleosides, Nucleotides Nucleic Acids* 25 (2006) 1309–1333; (b) C. Torborg, M. Beller, *Adv. Synth. Catal.* 351 (2009) 3027–3043; (c) A.O. King, N. Yasuda, *Palladium-Catalyzed Cross-Coupling Reactions in the Synthesis of Pharmaceuticals. In: Organometallics in Process Chemistry. Topics in Organometallic Chemistry*, vol 6. Springer, Berlin, Heidelberg, 2004, pp. 205–245.
- [148] (a) F. Christoffel, T.R. Ward, *Catal. Lett.* 148 (2018) 489–511; (b) M. Alisha, R. Mary Philip, G. Anilkumar, E. J. *Org. Chem.* 18 (2022), e202101384; (c) P.J. Anju, M. Neetha, G. Anilkumar, *Chem. Select* 7 (2022), e202103564.
- [149] (a) R. Liang, Y. Jia, *Acc. Chem. Res.* 55 (2022) 734–745; (b) O. Loiseleur, M. Hayashi, M. Keenan, N. Schmees, A. Pfaltz, *J. Organomet. Chem.* 576 (1999) 16–22; (c) C. Zhu, H. Chu, G.L. Gen, S. Ma, J. Zhang, *J. Am. Chem. Soc.* 141 (2019) 19246–19251; (d) J. Xie, R. Liang, Y. Jia, *Chin. J. Chem.* 39 (2021) 710–728; (e) A.B. Dounay, L.E. Overman, *Chem. Rev.* 103 (2003) 2945–2964; (f) P. Guiry, D. Kiely, *Curr. Org. Chem.* 8 (2004) 781–794.
- [150] (a) K. Yonehara, K. Mori, T. Hashizume, K.G. Chung, K. Ohe, S. Uemura, *J. Organomet. Chem.* 603 (2000) 40–49; (b) Y. Mata, M. Dieguez, O. Pamies, C. Claver, *Org. Lett.* 7 (2005) 5597–5599; (c) Y. Mata, O. Pamies, M. Dieguez, *Chem. Eup. J. Chem.* 13 (2007) 3296–3304.
- [151] (a) T.G. Kilroy, P.G. Cozzi, P.J. Guiry, *Synlett* (2004) 106–110; (b) T.G. Kilroy, P.G. Cozzi, N. End, P.J. Guiry, *Synthesis* (2004) 1879–1888.
- [152] (a) D. Mc Cartney, C. Nottingham, H. Müller-Bunz, P.J. Guiry, *J. Org. Chem.* 80 (2015) 10177–10186; (b) W.-Q. Wu, Q. Peng, D.-X. Dong, X.-L. Hou, Y.-D. Wu, *J. Am. Chem. Soc.* 130 (2008) 9717–9725; (c) M. Rubina, W.M. Sherrill, M. Rubin, *Organometallics* 27 (2008) 6393–6395.
- [153] P.W.N.M. van Leeuwen, P.C.J. Kamer, C. Claver, O. Pamies, M. Dieguez, *Chem. Rev.* 111 (2011) 2077–2118.
- [154] J. Mazuela, P. Tolstoy, O. Pamies, P.G. Andersson, M. Dieguez, *Org. Biomol. Chem.* 9 (2011) 941–946.
- [155] (a) C.A. Busacca, D. Griesbach, R.C. So, E.M. O'Brien, E.M. Spinelli, *Org. Lett.* 5 (2003) 595–598; (b) C.A. Busacca, D. Grossbach, S.J. Campbell, Y. Dong, M.C. Eriksson, R. E. Harris, P.J. Jones, J.Y. Kim, J.C. Lorenz, K.B. McKillop, E.M. O'Brien, F. Qiu, R. D. Simpson, L. Smith, R.C. So, E.M. Spinelli, J. Vitous, C. Zavattaro, *J. Org. Chem.* 69 (2004) 5187–5195.
- [156] C. A. Busacca, **U.S. Patent 316, 620, (2001) European Patent 1218388.**
- [157] M.O. Fitzpatrick, A.G. Coyne, P.J. Guiry, *Synlett* 18 (2006) 3150–3154.
- [158] (a) C.J. Moulton, B.L. Shaw, *J. Chem. Soc. Dalton Trans.* (1976) 1020–1024; (b) H.D. Empsall, E.M. Hyde, R. Markham, W.S. McDonald, M.C. Norton, B. L. Shaw, B. Weeks, *J. Chem. Soc. Chem. Commun.* 17 (1977) 589–590.
- [159] (a) J.L. Niu, X.Q. Hao, J.F. Gong, M.P. Song, *Dalton Trans.* 40 (2011) 5135–5150; (b) D. Benito-Garagorri, K. Kirchner, *Acc. Chem. Res.* 41 (2008) 201–213.
- [160] C.S. Consorti, G. Ebeling, F.R. Flores, F. Rominger, J. Dupont, *Adv. Synth. Catal.* 346 (2004) 617–624.
- [161] G.R. Rosa, D.S. Rosa, *RSC Adv.* 2 (2012) 5080–5083.
- [162] (a) F.S. Mancilha, Biadiazole, A. DaSilveira Neto, A.S. Lopes, P.F. Moreira Jr., F. H. Quina, R.S. Gonçalves, J. Dupont, *Eur. J. Org. Chem.* 92 (2006) 4924–4933; (b) B.A. De oliveira Neto, A. Sant'Ana Lopes, G. Ebeling, R.S. Gonçalves, V.E. U. Costa, F.H. Quinab, J. Dupont, *Tetrahedron* 61 (2005) 10975–10982.
- [163] M. Huang, L. Liang, *Organometallics* 23 (2004) 2813–2816.
- [164] L. Fan, B.M. Foxman, O.V. Ozerov, *Organometallics* 23 (2004) 326–328.
- [165] E. Shirakawa, T. Hiyama, *J. Organomet. Chem.* 576 (1999) 169–178.
- [166] T. Schultz, N. Schemees, A. Pfaltz, *Appl. Organometal. Chem.* 18 (2004) 595–601.
- [167] G. Singh, S. Bali, A.K. Singh, *Polyhedron* 26 (2007) 897–903.
- [168] X. Liu, X. Zhao, M. Lu, *Appl. Organometal. Chem.* 29 (2015) 419–424.
- [169] M.K. Yilmaz, S. Ince, M. Keles, B. Güzel, *J. CO2 Utiliz.* 42 (2020) 101309.
- [170] M.V. Nandakumar, J.G. Verkade, *Tetrahedron* 61 (2005) 9775–9782.
- [171] N. Iranpoor, H. Firouzabadi, A. Tarassoli, M. Fereidoonzehad, *Tetrahedron* 66 (2010) 2415–2421.
- [172] (a) O. Baudoin, *Eur. J. Org. Chem.* 2005 (2005) 4223–4229; (b) D. Zhang, Q. Wang, *Coord. Chem. Rev.* 286 (2015) 1–16; (c) G. Hedouin, S. Hazra, F. Gallou, S. Handa, *ACS Catal.* 12 (2022) 4918–4937.
- [173] B. Zhang, W. Wang, D. Shao, X. Hao, J. Gong, M. Song, *Organometallics* 29 (2010) 2579–2587.

- [174] A. Ros, B. Estepa, A. Bermejo, E. Álvarez, R. Fernández, J. Lassaletta, *Org. Chem.* 77 (2012) 4740–4750.
- [175] A. Ros, B. Estepa, P.O. Ramírez-López, E. Álvarez, R. Fernández, J.M. Lassaletta, *J. Am. Chem. Soc.* 135 (2013) 15730–15733.
- [176] Y. Uozumi, Y. Matsuura, T. Suzuka, T. Arakawa, Y.M.A. Yamada, *Synthesis* 48 (2016). A-J.
- [177] (a) G.R. Rosa, G. Ebeling, J. Dupont, A.L. Monteiro, *Synthesis* 18 (2003) 2894–2897; (b) G.R. Rosa, C.H. Rosa, F. Rominger, J. Dupont, A.L. Monteiro, *Inorg. Chim. Acta* 359 (2006) 1947–1954.
- [178] L. Liang, P. Chien, L. Song, *J. Organomet. Chem.* 804 (2016) 30–34.
- [179] B. Ines, R. SanMartín, F. Churruga, E. Dominguez, M.K. Uriaga, M.I. Arriortua, *Organometallics* 27 (2008) 2833–2839.
- [180] B. Zhang, C. Wang, J. Gong, M. Song, *J. Organomet. Chem.* 694 (2009) 2555–2561.
- [181] J. Yorke, A. Sanford, A. Decken, A. Xia, *Inorg. Chim. Acta* 363 (2010) 961–966.
- [182] K. Lee, H. Jeon, S. Han, J. Ham, Y. Kim, S.W. Lee, *Dalton Trans.* (2009) 6578–6592.
- [183] G. Dhangar, J. L.Serrano, C. Schulzke, K.C. Gunturu, A.R. Kapdi, *ACS Omega* 2 (2017) 3144–3156.
- [184] (a) A. Scrivanti, V. Beghetto, U. Matteoli, S. Antonaroli, A. Marini, F. Mandoj, R. Paolesse, B. Crociani, *Tetrahedron Lett.* 45 (2004) 5861–5864; (b) A. Scrivanti, V. Beghetto, U. Matteoli, S. Antonaroli, B. Crociani, A. Marini, *Tetrahedron* 61 (2005) 9752–9758; (c) B. Crociani, S. Antonaroli, A. Marini, U. Matteoli, A. Scrivanti, *Dalton Trans.* (2006) 2698–2705.
- [185] S. Lemouzy, M. Jean, F. Deplante, M. Albalat, D. Hérault, G. Buono, *Chem. Sel.* 3 (2018) 12281–12286.
- [186] C.M. So, C. C.Yeung, C.P. Lau, F.Y. Kwong, *J. Org. Chem.* 73 (2008) 7803–7806.
- [187] C.M. So, C.P. Lau, A.S.C. Chan, F.Y. Kwong, *J. Org. Chem.* 73 (2008) 7731–7734.
- [188] S.M. Wong, C.M. So, K.H. Chung, C.P. Lau, F.Y. Kwong, *Eur. J. Org. Chem.* (2012) 4172–4177.
- [189] (a) A. Mukherjee, A. Sarkar, *Tetrahedron Lett.* 45 (2004) 9525–9528; (b) D. Saha, R. Ghosh, R. Dutta, A. Kumar Mandal, A. Sarkar, *J. Organometal. Chem.* 776 (2015) 89–97.
- [190] B. Crociani, S. Antonaroli, M. Burattini, F. Benetollo, A. Scrivanti, M. Bertoldini, *J. Organomet. Chem.* 693 (2008) 3932–3938.
- [191] J.A. Weeden, R. Huang, K.D. Galloway, P.W. Gingrich, B.J. Frost, *Molecules* 16 (2011) 6215–6231.
- [192] A. Buchard, B. Komly, A. Auffrant, X.F.L. Goff, P.L. Floch, *Organometallics* 27 (2008) 4380–4385.
- [193] (a) I.J.S. Fairlamb, A.R. Kapdi, J.M. Lynam, R.J.K. Taylor, A.C. Whitwood, *Tetrahedron* 60 (2004) 5711–5718; (b) N.M. Chaignon, I.J.S. Fairlamb, A.R. Kapdi, R.J.K. Taylor, A.C. Whitwood, *J. Mol. Catal. A: Chem.* 219 (2004) 191–199; (c) I.J.S. Fairlamb, P. Sehnal, R.J.K. Taylor, *Synthesis* 3 (2009) 508–510.
- [194] A. Kapdi, V. Gayakhe, Y.S. Sanghvi, J. Garcia, P. Lozano, I. da Silva, J. Perez, J. L. Serrano, *RSC Adv.* 4 (2014) 17567–17552.
- [195] (a) R. Chinchilla, C. Nájera, *Chem. Soc. Rev.* 40 (2011) 5084–5121; (b) R. Chinchilla, C. Nájera, *Chem. Rev.* 107 (2007) 874–922; (c) S. Amrutha, S. Radhika, G. Anilkumar, *Beilstein J. Org. Chem.* 18 (2022) 262–285.
- [196] A. Scharf, I. Goldberg, A. Vigalok, *J. Am. Chem. Soc.* 135 (2013) 967–970.
- [197] (a) A. Fleckhaus, A.H. Mousa, N.S. Lawal, N.K. Kazemifar, O.F. Wendt, *Organometallics* 34 (2015) 1627–1634; (b) A.H. Mousa, A. Fleckhaus, M. Kondrashov, O.F. Wendt, *J. Organomet. Chem.* 845 (2017) 157–164.
- [198] (a) M.M. Heravi, E. Hashemi, F. Azimian, *Tetrahedron* 70 (2014) 7–21; (b) M. Kosugi, K. Fugami, *J. Organomet. Chem.* 653 (2002) 50–53; (c) N. Srivastav, R. Singh, V. Kaur, *RSC Adv.* 5 (2015) 62202–62213.
- [199] M. Kopyrowski, R.M. Sebastian, V. Maraval, M. Zablocka, V. Cadierno, B. Donnadiou, A. Igau, A.M. Caminade, J.P. Majoral, *Organometallics* 21 (2002) 4680–4687.
- [200] C.M. Crawford, S.T. Burling, I.J.S. Fairlamb, A.R. Kapdi, R.J.K. Taylor, A. C. Whitwood, *Tetrahedron* 61 (2005) 9736–9751.
- [201] (a) A. Scrivanti, U. Matteoli, V. Beghetto, S. Antonaroli, B. Crociani, *Tetrahedron* 58 (2002) 6881–6886; (b) B. Crociani, S. Antonaroli, V. Beghetto, U. Matteoli, A. Scrivanti, *Dalton Trans.* (2003) 2194–2202; (c) B. Crociani, S. Antonaroli, L. Canovese, P. Uguagliati, F. Visentin, *Eur. J. Inorg. Chem.* (2004) 732–742.
- [202] P. Wang, L. Gong, *Acc. Chem. Res.* 53 (2020) 2841–2854.
- [203] B. Zhou, K. Li, C. Jiang, Y. Lu, T. Hayashi, *Adv. Synth. Catal.* 359 (2017) 1–8.
- [204] Z. Qiu, Li, Z. Zhang, D. Teng, *Transit. Met. Chem.* 44 (2019) 649–654.
- [205] (a) S.M. Khake, V. Soni, R.G. Gonnade, B. Punji, *Dalton Trans.*, 43 (2014) 16084–16096; (b) D.K. Pandey, S.M. Khake, R.G. Gonnade, B. Punji, *RSC Adv.* 5 (2015) 81502–81514; (c) S.M. Khake, R.A. Jagtap, Y.B. Dangat, R.G. Gonnade, K. Vanka, B. Punji, *Organometallics* 35 (2016) 875–886.
- [206] C. Wang, Y. Li, B. Lu, X. Hao, J. Gong, M. Song, *Polyhedron* 143 (2018) 184–192.
- [207] Y. Li, X. Yu, Y. Wang, H. Fu, X. Zheng, H. Chen, R. Li, *Organometallics* 37 (2018) 979–988.
- [208] M.M. Heravi, V. Zadsirjan, P. Hajjabbasi, H. Hamidi, *Monatsh Chem.* 150 (2019) 535–591.
- [209] H. Horibe, Y. Fukuda, K. Kondo, H. Okuno, Y. Murakami, T. Aoyama, *Tetrahedron* 60 (2004) 10701–10709.
- [210] (a) A. Monfared, R. Mohammadi, S. Ahmadi, M. Nikpassand, A. Hosseini, *RSC Adv.* 9 (2019) 3185–3202; (b) Y. Nakao, T. Hiyama, *Chem. Soc. Rev.* 40 (2011) 4893–4901.
- [211] C.P. Lau, F.Y. Kwong, *Org. Lett.* 11 (2009) 317–320.
- [212] H.W. Lee, F.L. Lam, C.M. So, C.P. Lau, A.S.C. Chan, F.Y. Kwong, *Angew. Chem. Int. Ed.* 48 (2009) 7436–7439.
- [213] (a) M.M. Heravi, Z. Kheilkordi, V. Zadsirjan, M. Heydari, M. Malmir, *J. Organomet. Chem.* 861 (2018) 17–104; (b) B.S. Takale, F. Kong, R.R. Thakore, *Organics* 3 (2022) 1–21; (c) R. Dorel, C.P. Grugel, A.M. Haydl, *Angew. Chem. Int. Ed.* 58 (2019) 17118–17129; (d) P. Ruiz-Castillo, S.L. Buchwald, *Chem. Rev.* 116 (2016) 12564–12649.
- [214] P. Ramirez-Lopez, A. Ros, A. Romero-Arenas, J. Iglesias-Sigüenza, R. Fernández, J.M. Lassaletta, *J. Am. Chem. Soc.* 138 (2016) 12053–12056.
- [215] V. Hornillos, A. Ros, P. Ramirez-Lopez, J. Iglesias-Sigüenza, R. Fernández, J. M. Lassaletta, *Chem. Commun.* 52 (2016) 14121–14124.
- [216] S. Vyskocil, M. Smrcina, P. Kocovsky, *Tetrahedron Lett.* 39 (1998) 9289–9292.
- [217] P. Kocovsky, S. Vyskocil, I. Cisarova, J. Sejbal, I. Tislerova, M. Smrcina, G. C. Lloyd-Jones, S.C. Stephen, C.P. Butts, M. Murray, V. Langer, *J. Am. Chem. Soc.* 121 (1999) 7714–7715.
- [218] J.P. Tassone, G.J. Spivak, *J. Organometal. Chem.* 841 (2017) 57–61.
- [219] W.Y. Chiang, F.E. Hong, *J. Organometal. Chem.* 694 (2009) 1473–1481.
- [220] S.L. Parisel, L.A. Adrio, A.A. Pereira, M.M. Perez, J.M. Vila, K.K. Hii, *Tetrahedron* 61 (2005) 9822–9826.
- [221] M. Yang, Y. Liu, J. Gong, M. Song, *Organometallics* 30 (2011) 3793–3803.
- [222] J.J. Jiang, D. Wang, W.F. Wang, Z.L. Yuan, M.X. Zhao, F.J. Wang, M. Shi, *Tetrahedron: Asymmetry* 21 (2010) 2050–2054.
- [223] H. Li, Z.H. Zhiyan, H. Qiong, W.Y. Qin, *J. Org. Chem.* 74 (2008) 283–288.