

Name - Pooja Padhye Rangao

Roll No - 8465

Date - 11/12/2025

DATE		
PAGE NO.		

Date : 11/12/2025

We visited the OJT (on job training) centre for/at x-metric water testing lab. the lab is located at shiye MIDC first, we were shown the lab by ma'am then they told us about the rules, OJT timing, and what work is done in the lab. They explained how water testing is carried out, which instruments and machines are used, and which chemicals are required, they also told us about the precautions to be taken during testing. After that they showed us the water testing process and explained which tests are performed in water samples.

Date : 12/12/2025

on the second day, we performed some water tests. These are included : COD (chemical oxygen demand), TDS (Total Dissolved Solid)

COD $\rightarrow$ This is method used to measure the amount of oxygen required to oxidized wastewater or oxidized substance present in water. In this test, $K_2Cr_2O_7$ is used as an oxidizing agent, then H_2SO_4 , FAS (ferrous ammonium sulphate) and Ag_2SO_4 as a catalyst used. During the test, the sample is taken and heated at about $150^\circ C$ for 2 hrs. After that, titration is performed the colour changes from yellow-green to reddish brown at the end point. then calculⁿ are done to find the COD value.

$$\text{Formula} : \text{COD} = \frac{(A-B) \times M \times 8000}{\text{ml. of sample.}}$$

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TDS → This method measures the amount of organic and inorganic dissolved substance in water, it is a fast process that measure oxygen consumption for oxidation. a water sample is taken & placed in platinum - catalyzed combustion chamber at about 900°C to 1000°C . At high temperature, substance convert into CO_2 , H_2O , N and NO_x .

$$\text{Formula} - \text{TDS} = \frac{[B-A] \times 1000 \times 10 \text{ ml}}{\text{ml of sample}}$$

Date : 13/12/2025

Alkalinity test = 100 ml of sample in 250 ml & conical flask, pink, it is titrated with a standard acid solⁿ. 2-3 drops of phenolph indicator. if sample turns color until the pink colour just disappear, including indicating the end point for phenolphthalin alkalinity next, 2-3 drops of methyl orange indicator are added are added to the same flask, and titration continues until the color changes from yellow to faint pink or orange. the total volume of acid used from the beginning of the titration until the methyl orange end point is used to calculate the total Alkalinity of the sample.

pH meter = switch the pH meter, then calibrate using standard buffer solution. Rinse electrode with distilled water dip the electrode in test sample solⁿ & Record the pH

This is the 3rd day was a complete and test is pH meter and Alkalinity test.

Date : 14/12/2025

Then, on 4th day, the samples that come to the lab were tested at that time. tests such as chloride test, calcium test.

- calcium test → some of sample with 20 ml NaOH and 100-200 mg of murexide indicator, titrate against 0.01 M EDTA. then the end point is the colour pink to purple.
- chloride test → 50 ml of sample with 2 ml of $K_2Cr_2O_7$ it titrate against 0.02 $AgNO_3$ and the end point colour is Red ring (persistent).

$$\text{Formula : } \text{chloride mg/l} = \frac{\text{ml} \times 0.02 N \times 1000 \times 35.5}{\text{ml of sample}}$$

Date : 15/12/2025

Holiday

Date : 16/12/2025

Then on 5th day the sample that come in the lab were tested at that time tests such as sodium test.

- sodium : 50 ml in sample for a sodium test flame emission photometry where a water sample is aspirated into a flame and the resulting light emission at 589 nm is measured against a calibration curve. The procedure includes preparing standards NaCl add to 0.03 M EDTA solⁿ end point will red (wine red to blue).

- oil and grease $\rightarrow$ oil and grease in water are determined by acidifying a sample ($\text{pH} < 2$) add HCl or H_2SO_4 extracting the analysis using a solvent in a separatory funnel drying with anhydrous sodium sulfate and measuring the residue gravimetrically after solvent evaporation. This standard procedure and cooled the dish and record the weight.

Formula = $\frac{B}{A}$

$$\text{Final wt} = \frac{\text{Initial wt} \times 1000 \times 1000}{\text{ml. of sample}}$$

Date : 20/12/2025

on then 8th day ~~the~~ the samples that come to the lab were tested at that time. tests such as Turbidity ~~and~~ ~~sed~~

- Turbidity $\rightarrow$ Turbidity in water is measured by analyzing ~~ess~~ caused by suspended particles typically using a nephelometer to provide readings in NTU or JTU. The procedure involves calibrating the instrument stand solutions to ~~es~~ ensure accuracy water sample is well-mixed but free of air bubbles light causes false, high reading change switch from 2000 NTU to 200 NTU).

Date : 21/12/2025

on then test day we performed water sample These are including Batch lime Alum poly.

The batch water treatment procedure using lime, alum and polymer (poly) collect a raw water sample and pour 1000 ml into each jar test beaker. measure the raw water's pH turbidity and alkalinity. Add a calculated dosage of lime CaCO_3 to raise the pH (often to a target range of 9-12 for ~~PH~~ optimal precipitation in industrial cases). mix a short period flash mix the sample of high speed 1-2 minutes to ensure complete dispersion of the coagulant. Generally lime is added first to adjust pH followed by alum then the polymer to enhance settling turbidity, hardness or color removal often employed in lime, separate the colour water layer into a different beaker and use it for further testing (TS, TDS, COO, TSS etc).

Examiner
Head
04-03-2016

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